

Luminescence Geochronology

EarthScope Short Course

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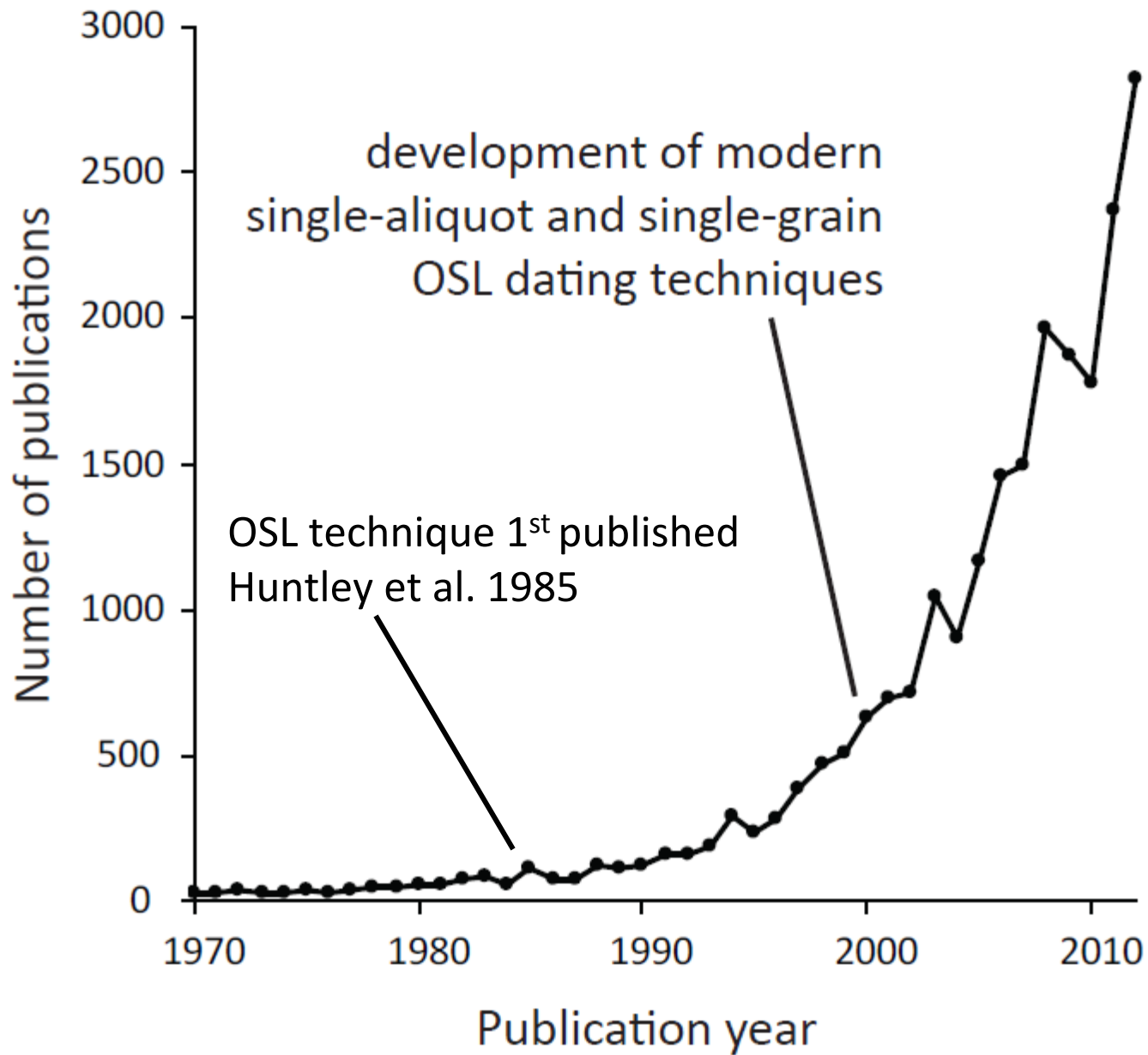
Luminescence techniques (OSL and TL)

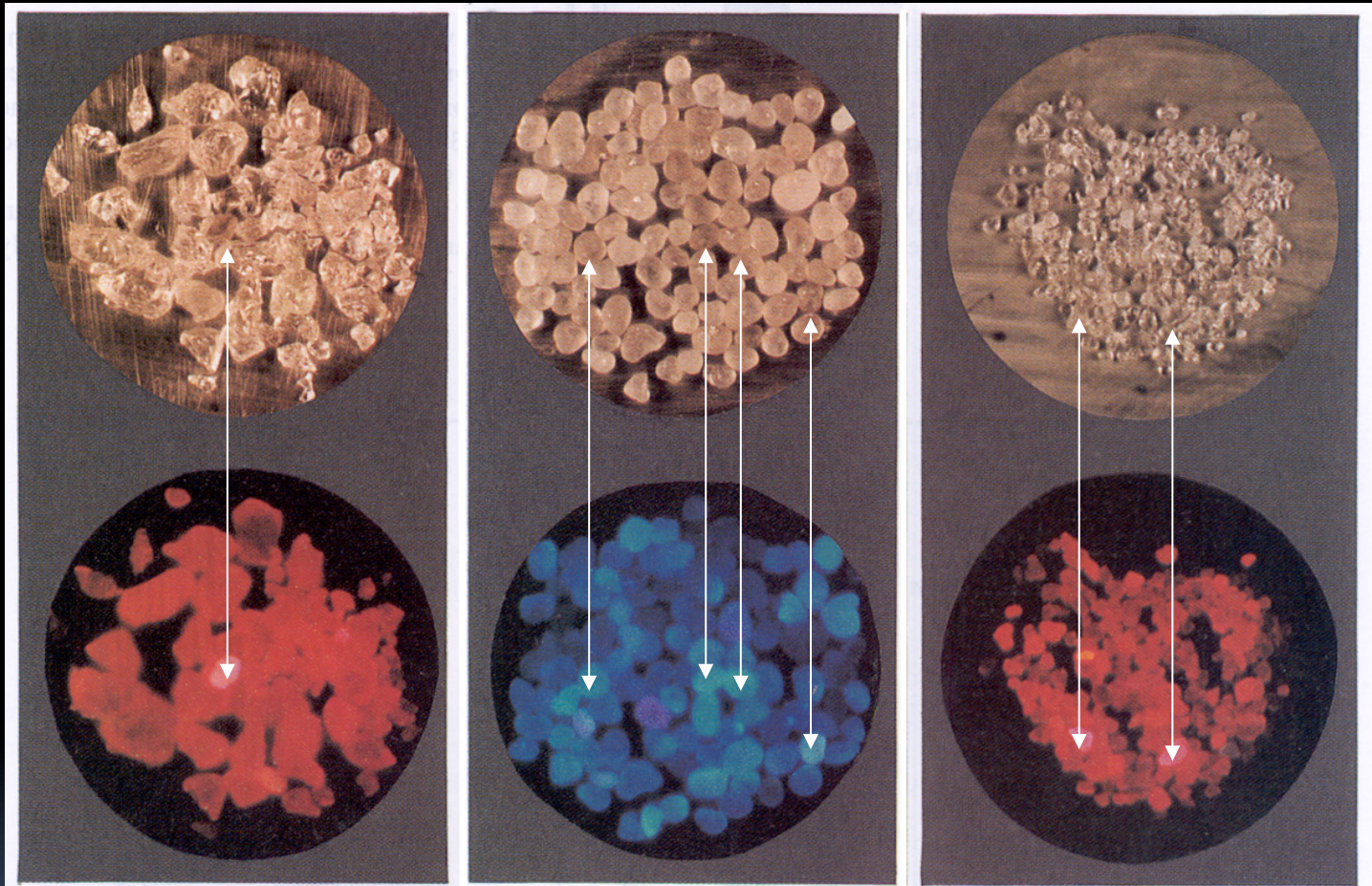
Optically stimulated luminescence, OSL

- Production of light through stimulation by photons of any wavelength

Thermoluminescence, TL

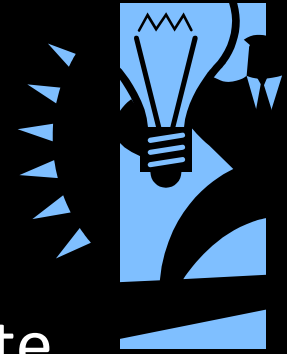
- Light production through stimulation by heat.





Luminescence (bottom) in quartz grains. Note the variation in luminescence between grains. In sample 1 most of the signal would be generated by a single grain (Ganzawa et al., 1997).

The General Idea:

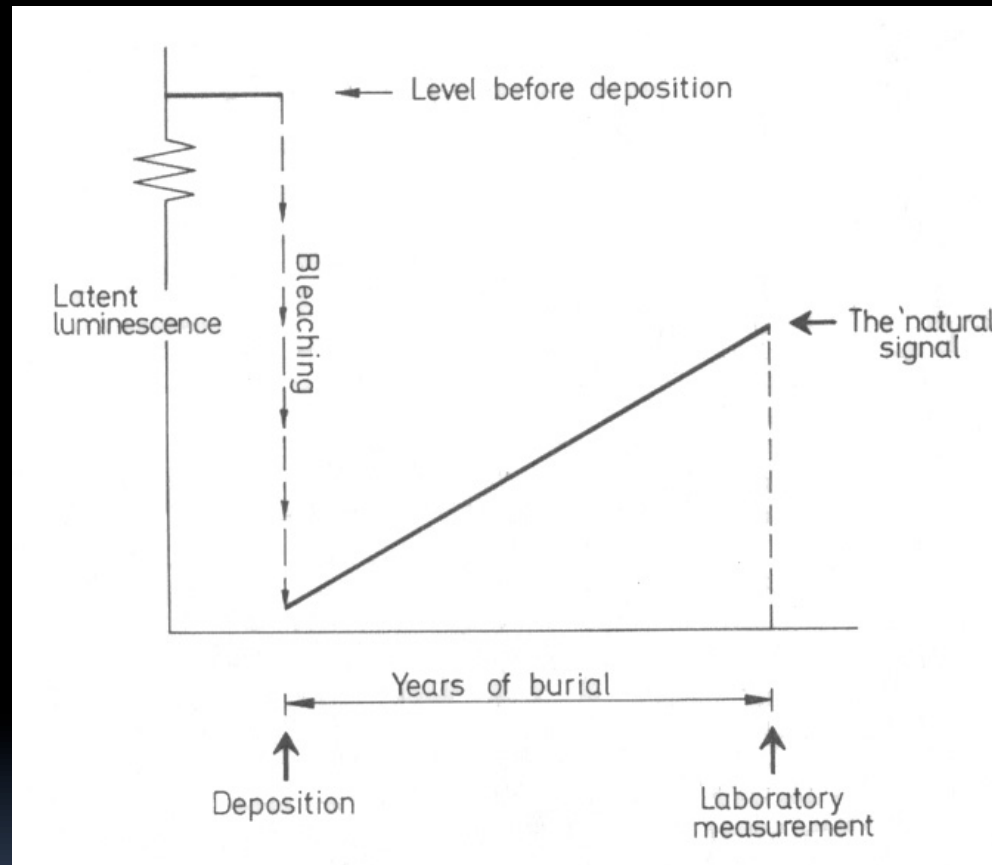


Luminescence techniques provide an estimate of the time since sediment was last exposed to light/heat (which zeros the luminescence signal).

After burial, the luminescence signal grows with exposure to radiation in the surrounding sediments (K, U, Th, Rb) plus cosmic radiation

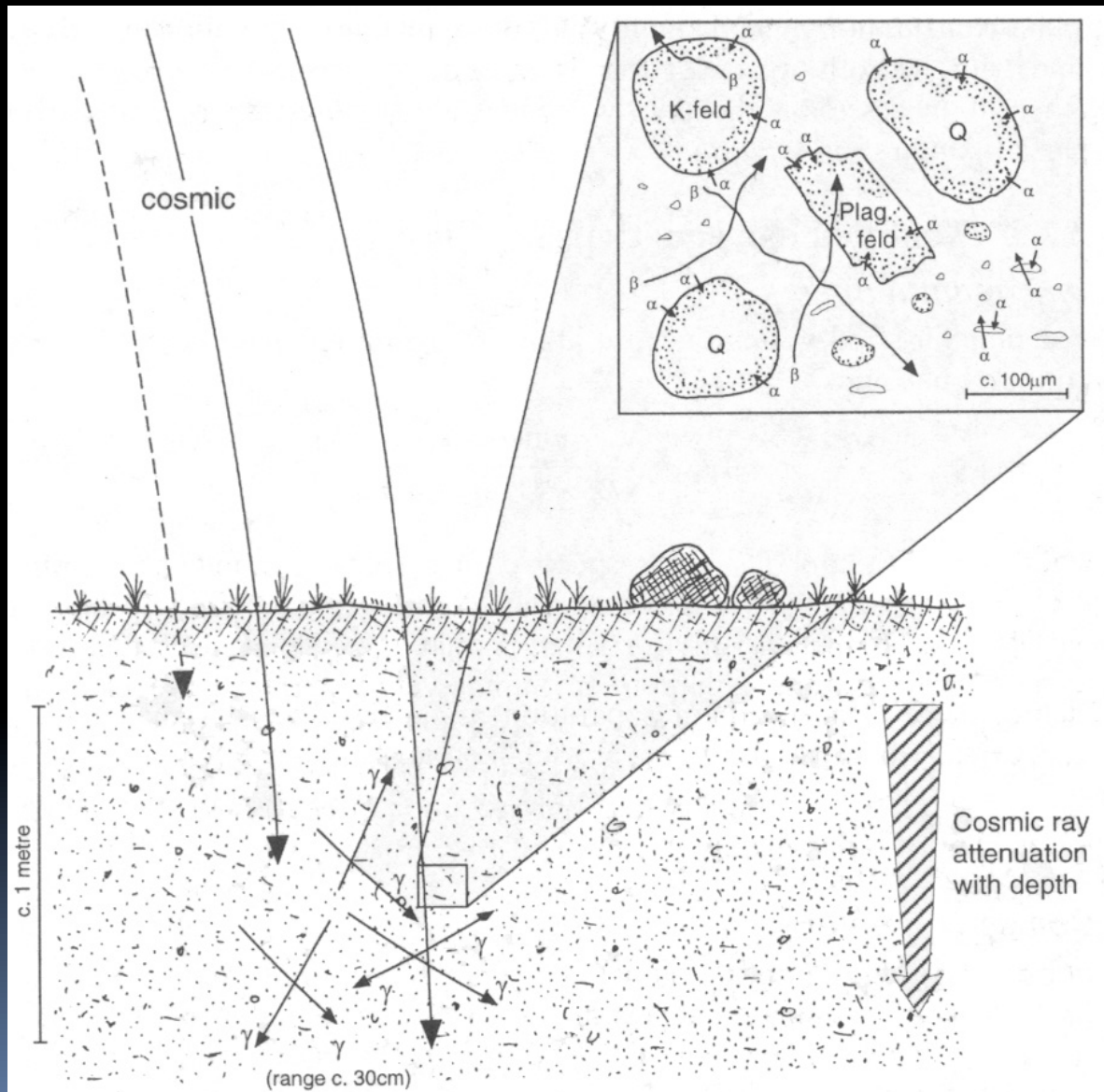
Measured luminescence signal is related to the length of burial and radioactive environment

Bleaching and Signal Generation



Bleaching (exposure to light) sets the signal to zero; the 'natural' signal regenerates over time by exposure to natural radiation. The strength of the signal is related to the number of years of burial and the flux of radiation

Generation of the burial dose:



The 'natural' signal is produced by the flux of α , β , γ and cosmic radiation.

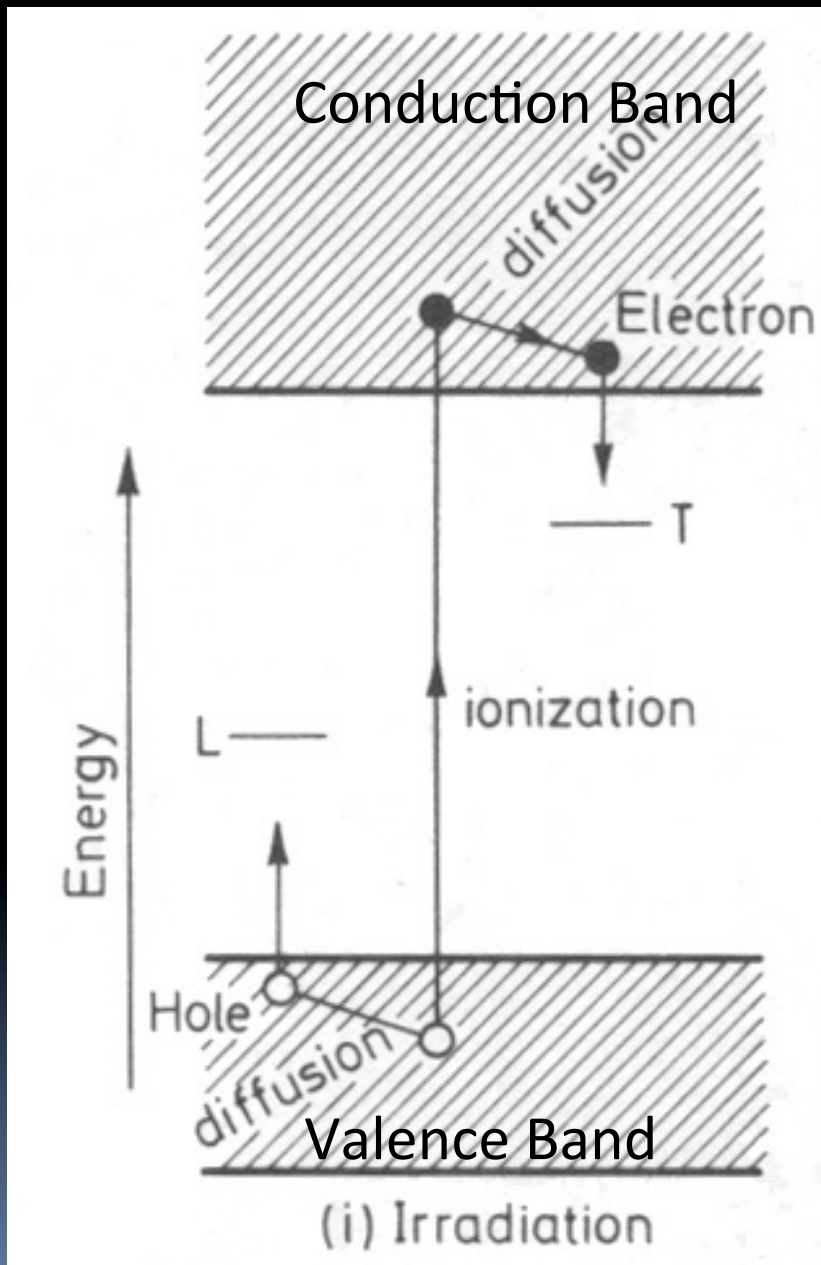
Generation of the Luminescence signal-

Ionizing radiation ejects electrons from atoms in the crystal lattice

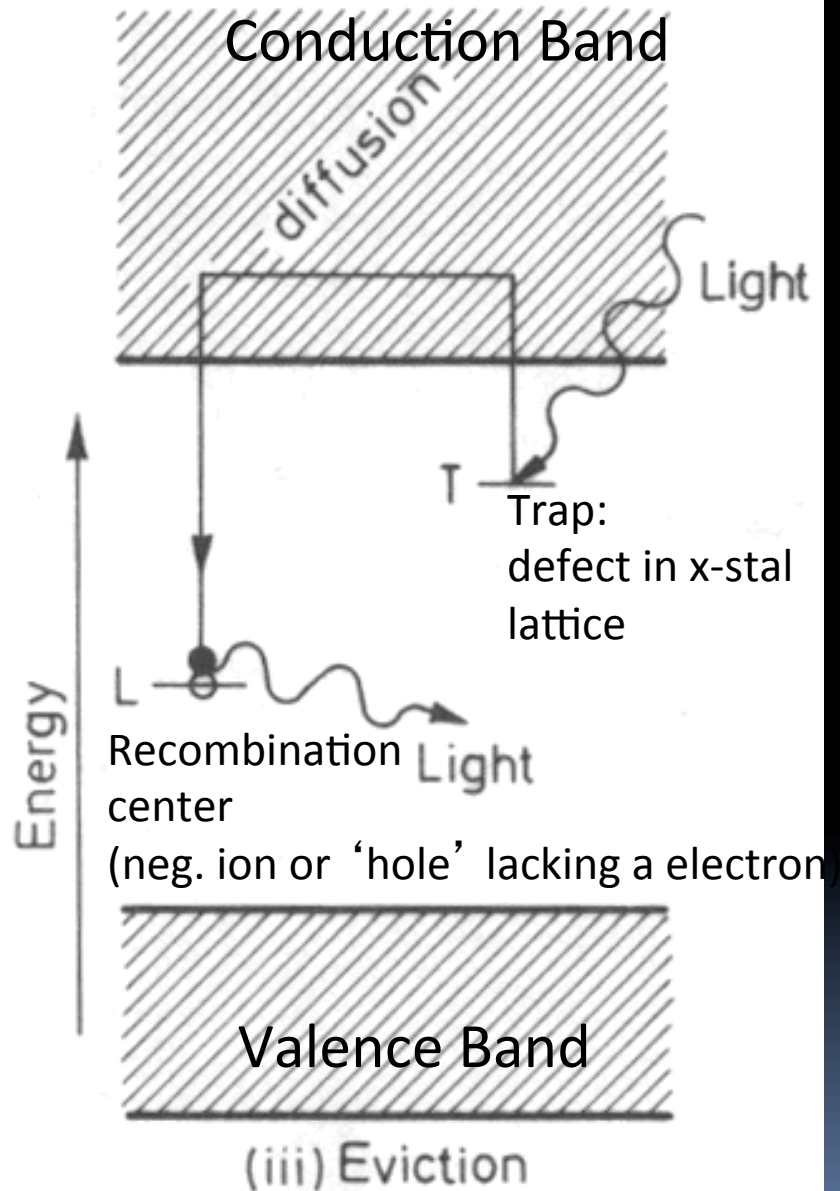
These electrons are trapped in ‘positive’ defects in the crystal lattice (holes or traps) and stored here over geological time until stimulated by external energy (light, heat)

Upon absorption of energy, the electron is released from the trap and when it enters a recombination center (another site attractive to electrons) it releases a photon of energy (wavelength depend on trap type)

Production of Electrons and Holes



Ionization due to exposure of the crystal to nuclear radiation, with trapping of electrons and holes at defects, T and L, respectively (Aitken, 1998).



Generation of Luminescence Signal

Light of an appropriate wavelength (or heat) evicts electrons from the traps. Some reach luminescence centers, producing light (i.e. OSL)

Age Calculation:

Dependent upon:

Flux of radiation as measured in the field and/or determined by chemistry (i.e. concentrations of U, Th, K, Rb) = dose-rate

Intensity of OSL signal as related to a known flux of radiation = equivalent dose (De)
(equivalent to the natural luminescence signal)

Equivalent Dose (D_E)

def:

Laboratory dose of radiation needed to induce a luminescence signal equal to the natural luminescence signal of the sample.

Unit: Gray (Gy) $1 \text{ Gy} = 1 \text{ J/kg}$

Approximately equivalent to dose sample received in nature, the “Paleodose”, P .

Dose-Rate (D_R)

def:

Dose per unit of time received by the sample while buried.

Typical unit: Grays per thousand years (Gy ka^{-1}).

Calculated from measured concentrations of U, Th, K, Rb and cosmic radiation, with correction for absorption of radiation by water.

Can also be measured directly or indirectly in the field.

Age:

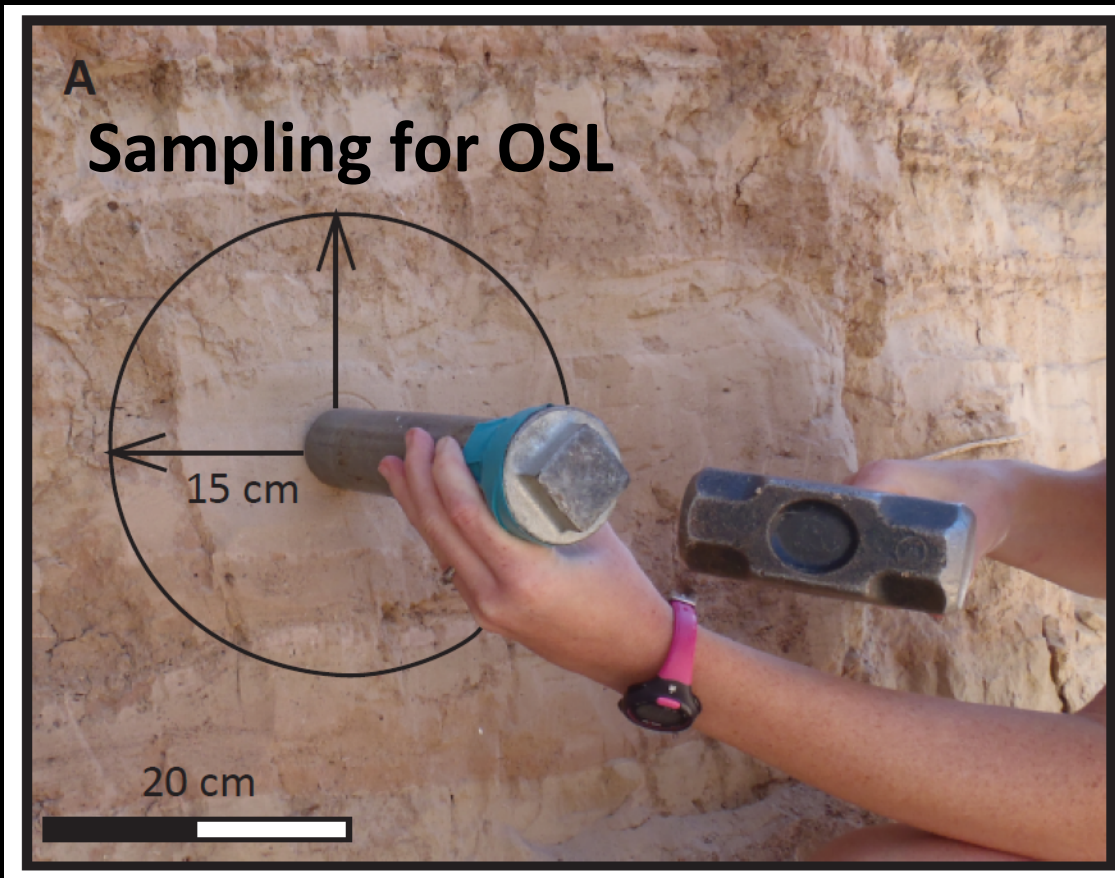
$$\text{Age} = \frac{\text{Equivalent Dose (Gy)}}{\text{Dose-Rate (Gy ka}^{-1}\text{)}} = \text{ka}$$

Sample Collection (typical)

- Sample collected by driving metal tubes into sediment.
- Sample tube removed and ends sealed.
- Water and chemistry samples also collected for dose-rate determination.
- Sample depth, latitude, longitude recorded for cosmic contribution

OSL Tool kit:





In the Lab

Detector/photomultiplier tube

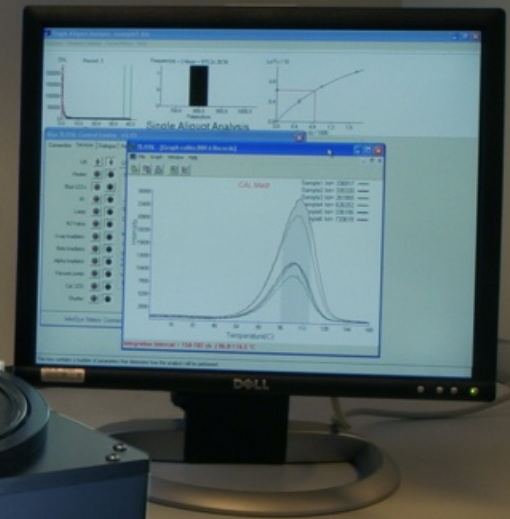
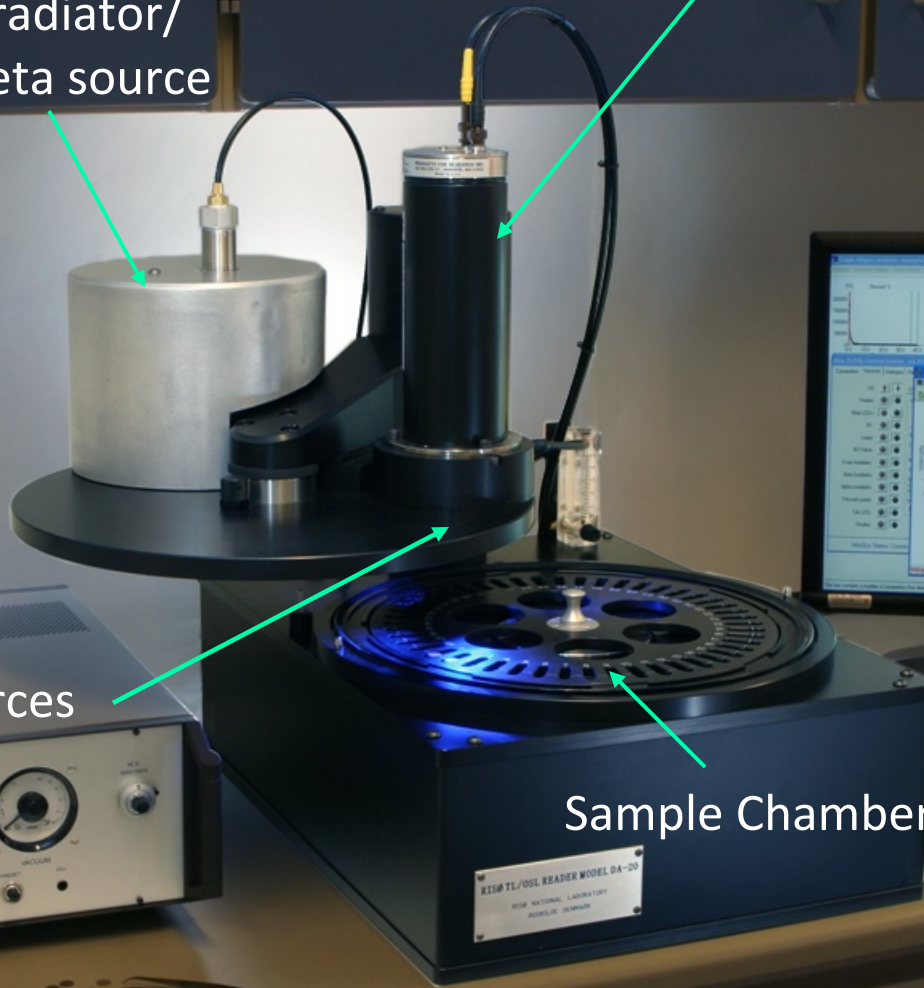
Risø TL/OSL reader
Model DA-20

Irradiator/
Beta source

Light sources

Sample Chamber

Mini-computer

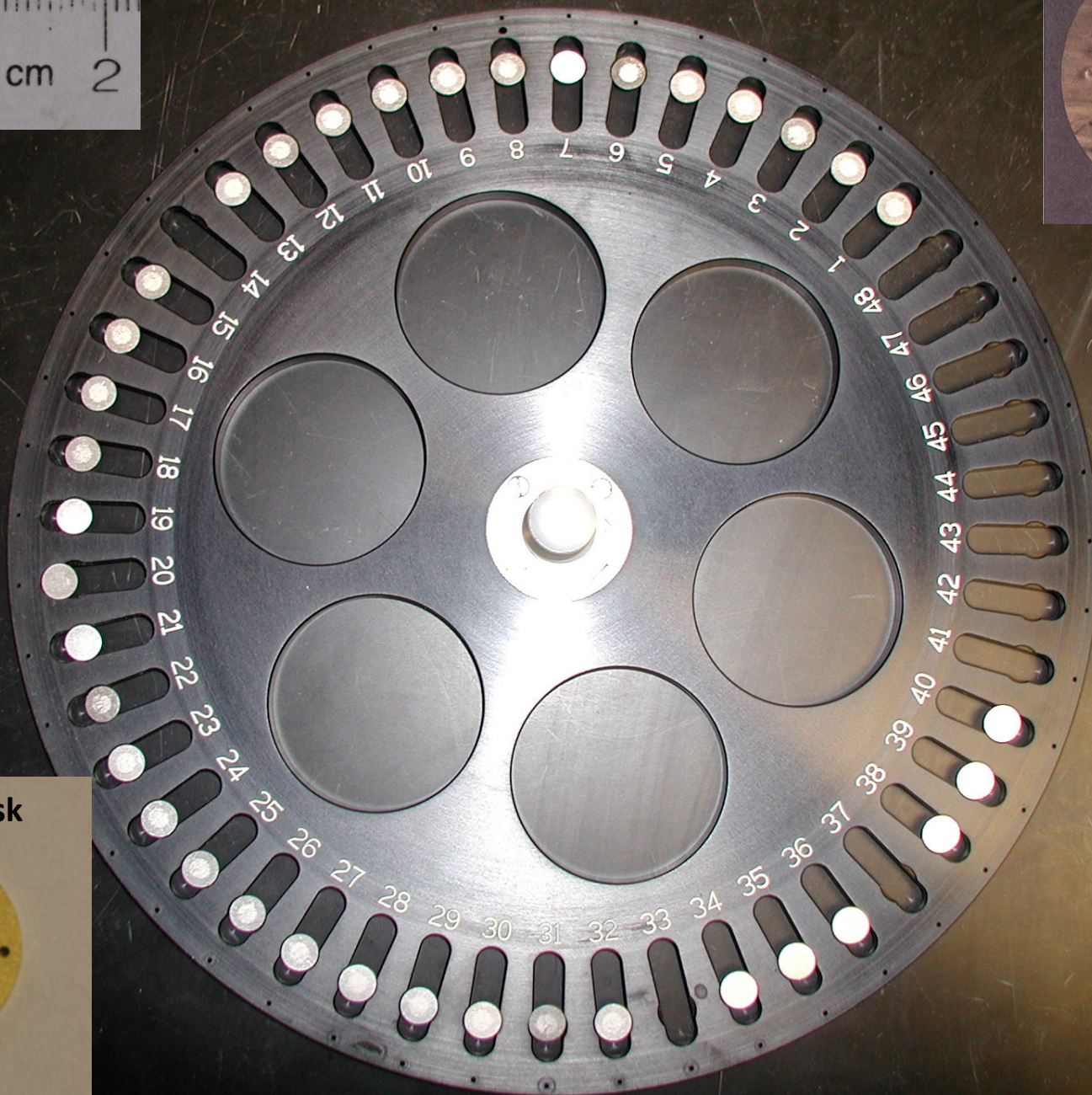
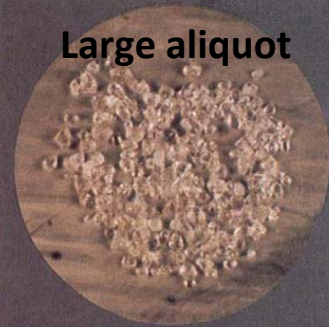


Small aliquot

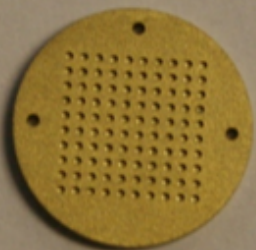


Sample holder for Risø

Large aliquot



Single-grain disk



OSL Sample Preparation

Isolate desired grain size

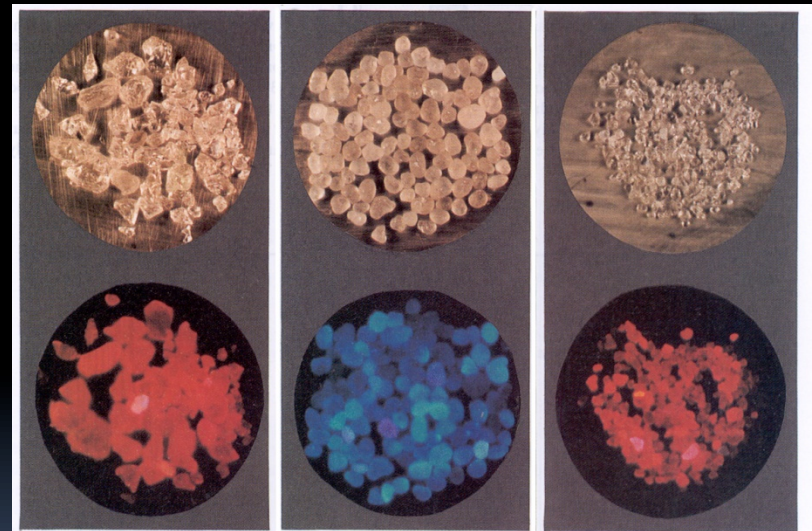
- Wet sieving (60-90, 90-125, 90-150, 150-200 μ ...)

Isolate desired mineral (quartz)

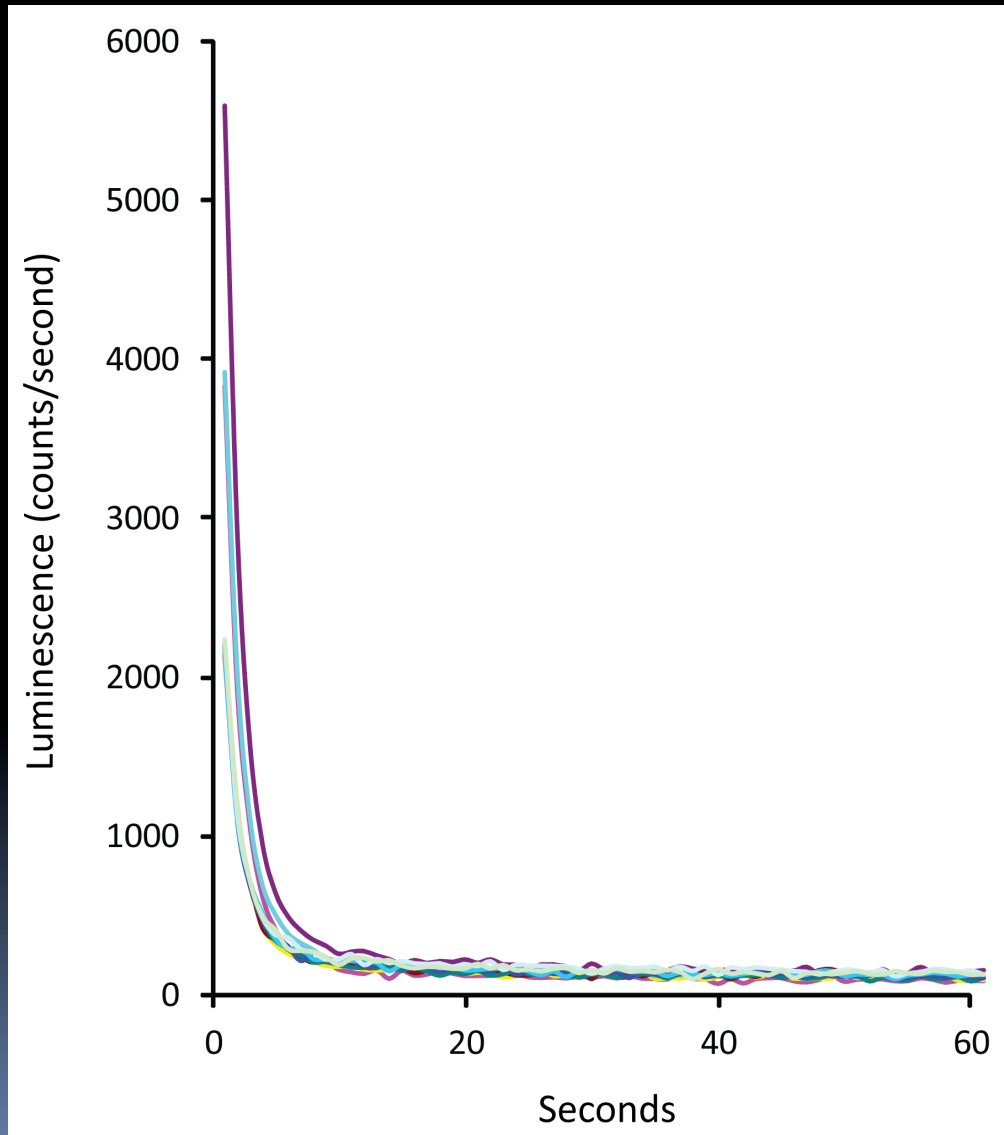
- Carbonates removed with HCl.
- Organics removed with hydrogen peroxide or bleach.
- Quartz & feldspar floated with 2.7 g cm⁻³ Na-polytungstate.
- Feldspar removed and quartz grains etched with HF.
- HCl to remove any fluorides produced.
- Re-sieve & discard fine fraction with residual feldspars.

D_E and OSL Age Determination

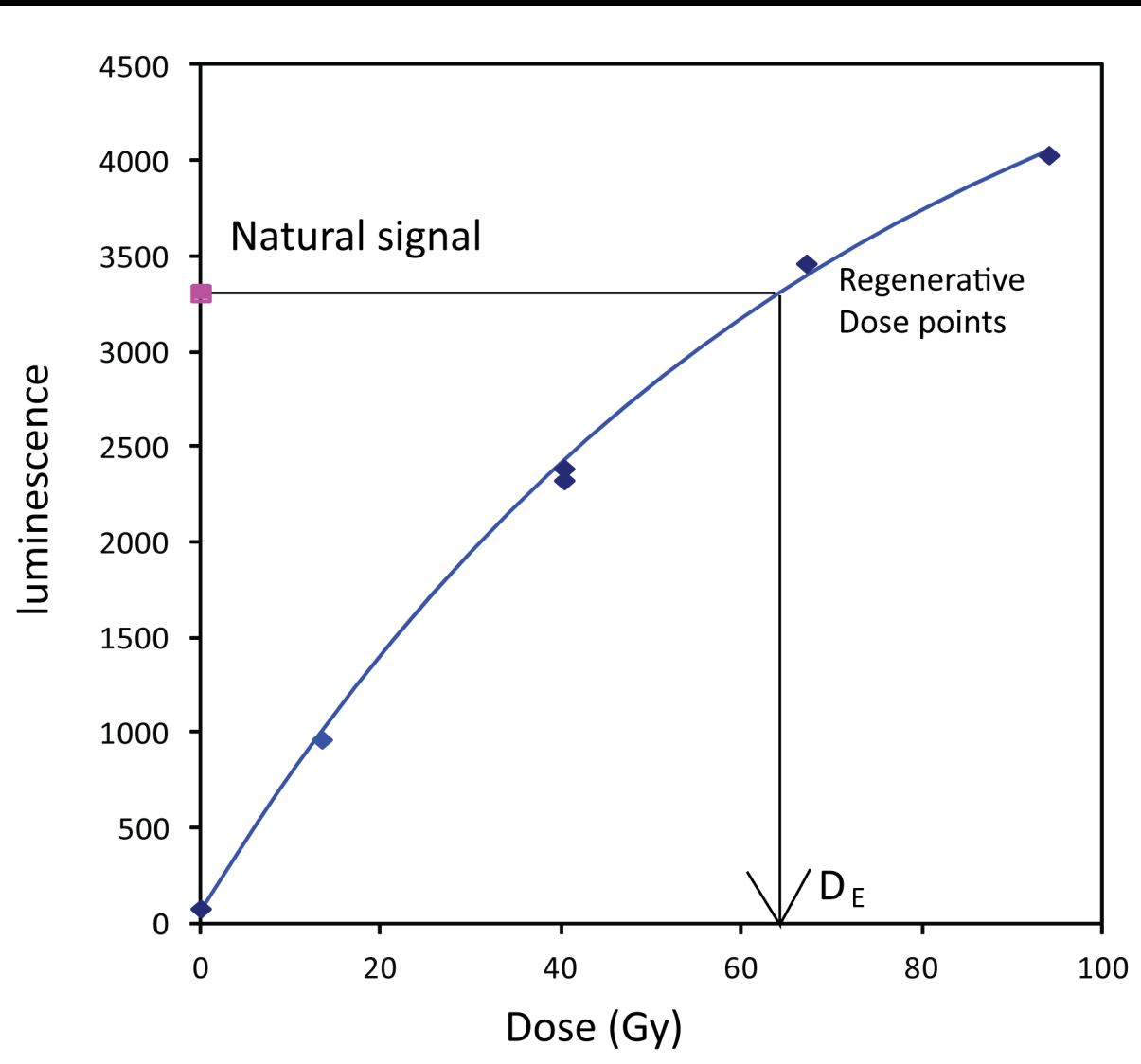
- Monolayer of quartz attached to ss disk using silicon spray or single-grain disc.
- Several cycles of preheating, irradiation, and measurement of OSL signal.

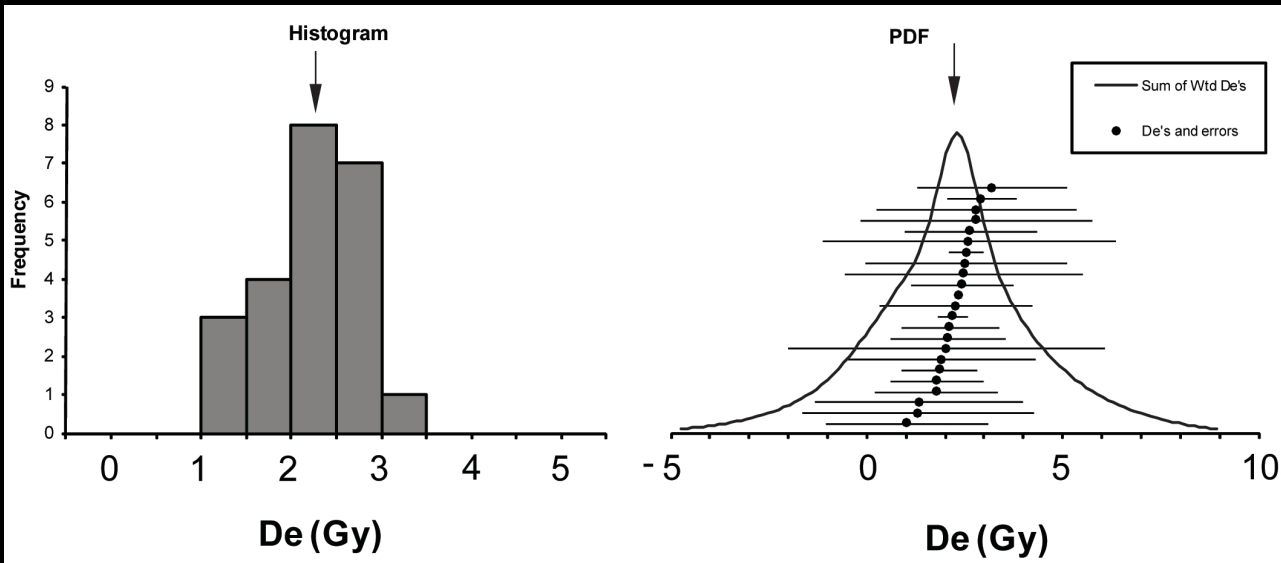


Quartz Decay Curve (OSL)

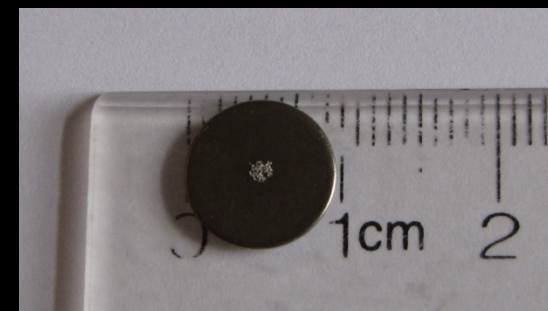


Dose-response curve

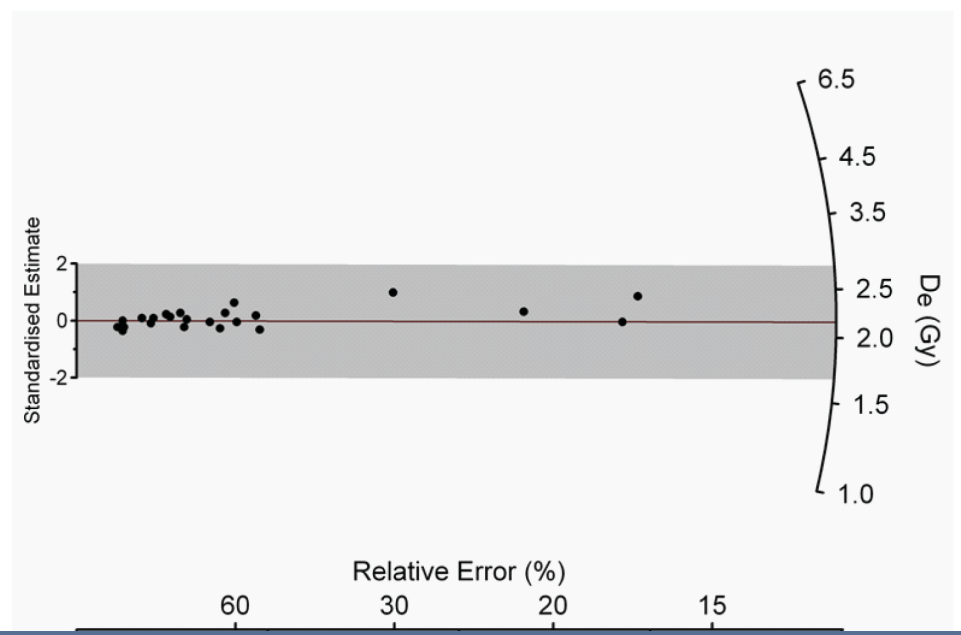




Small aliquot



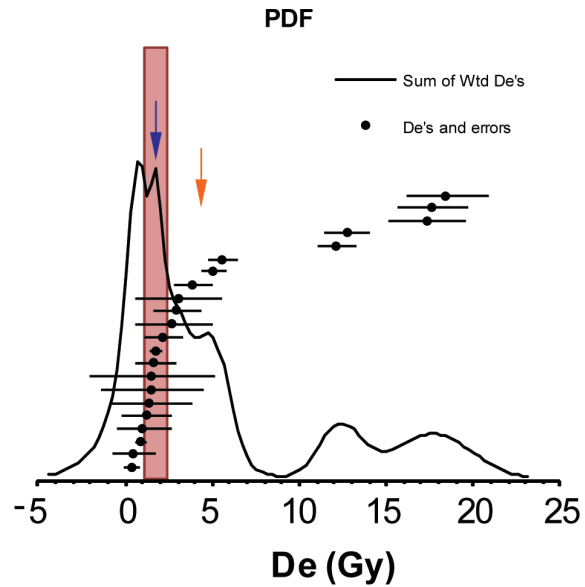
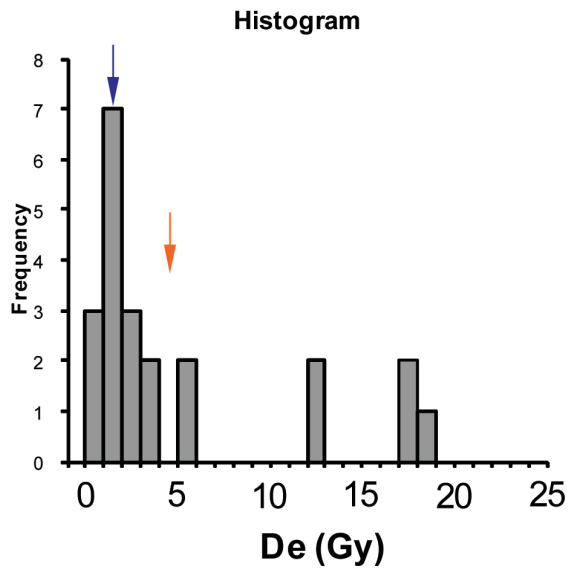
Mean De =
2.17 Gy



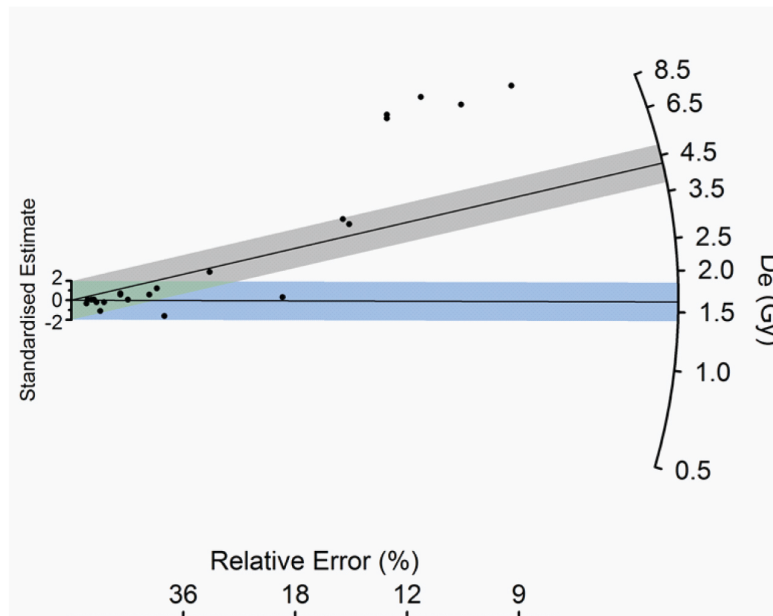
6.5
4.5
3.5
2.5
2.0
1.5
1.0

De (Gy)

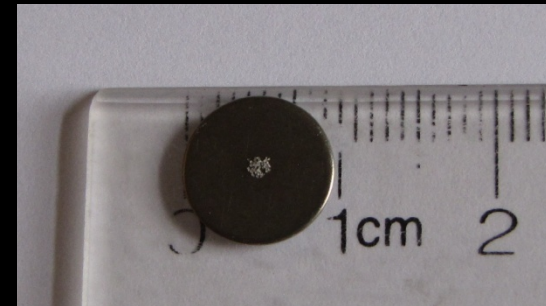
(a) USU-603



MAM De = 1.53 Gy
CAM De = 4.23 Gy

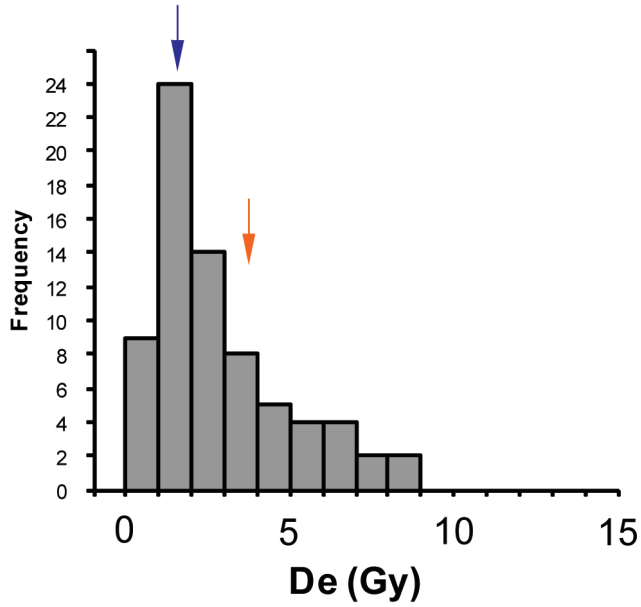


Small aliquot

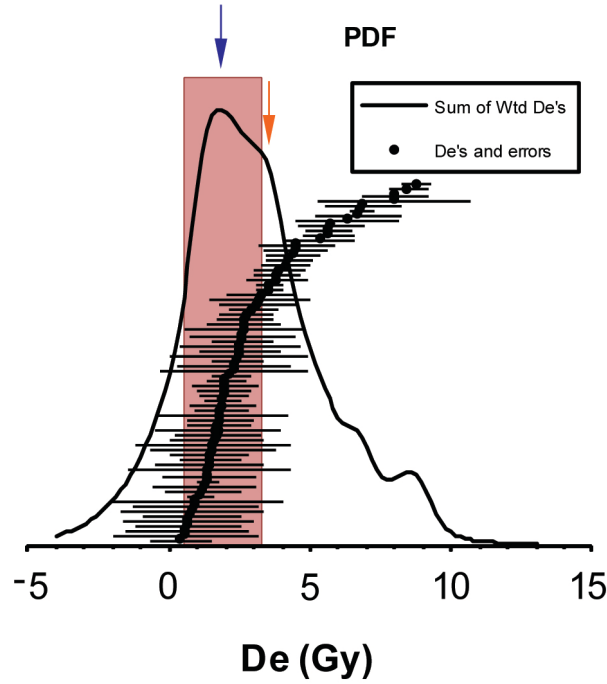


USU-815

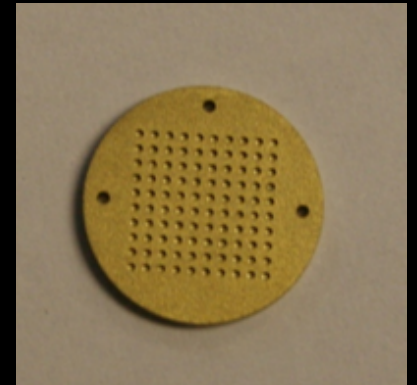
Histogram



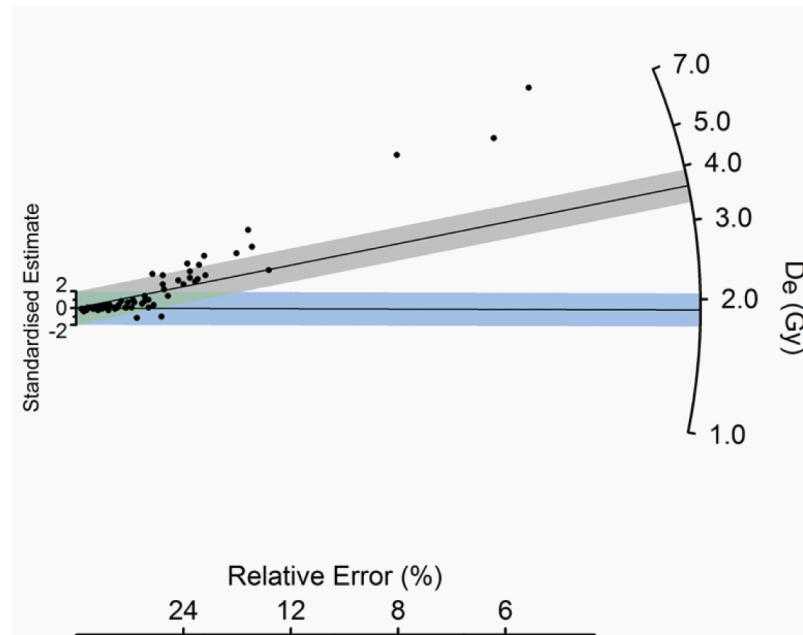
PDF



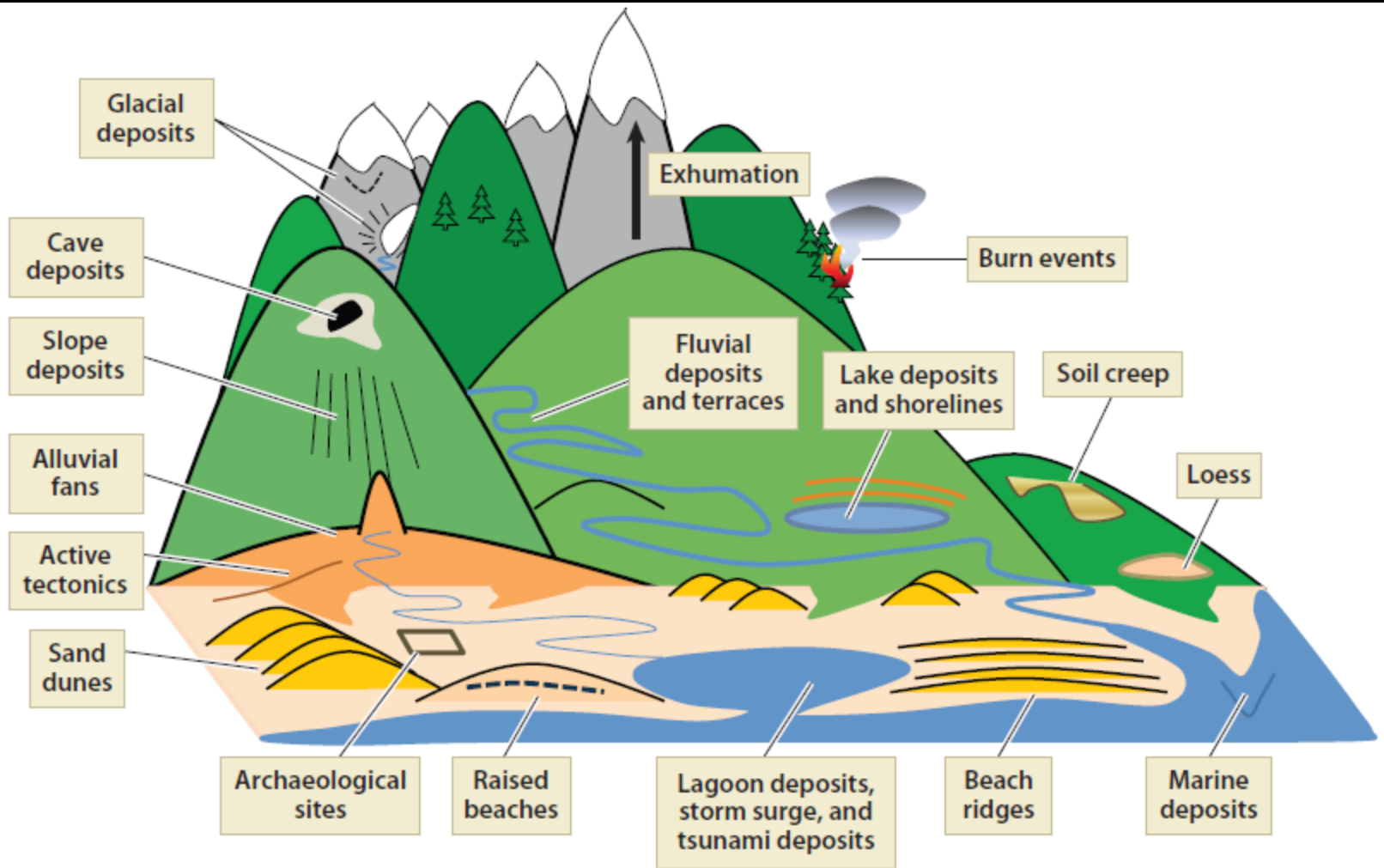
Single-grain



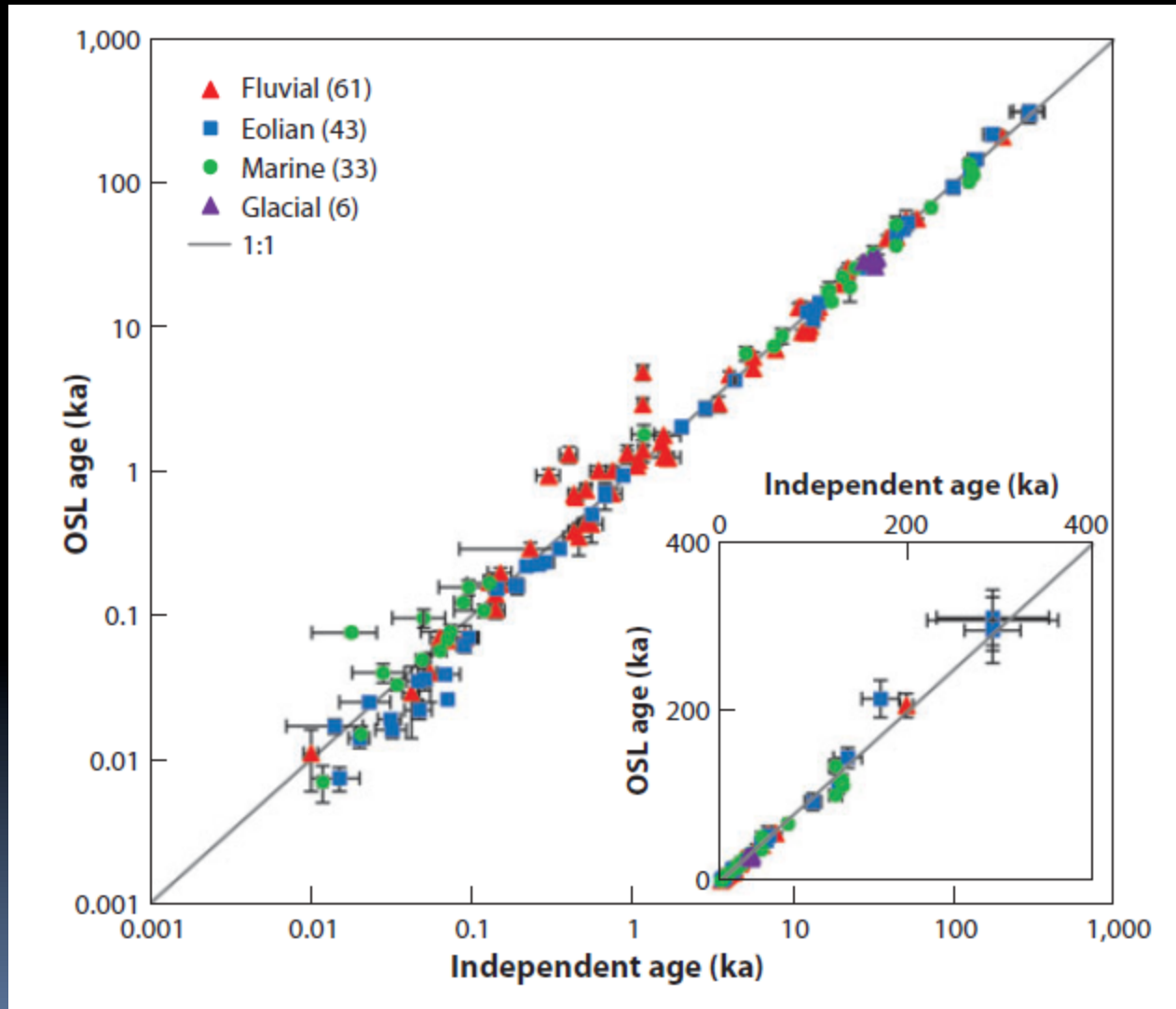
MAM De = 1.90 Gy
CAM De = 3.56 Gy



Datable deposits: nearly any...



Age range – few decades up to 300ka with Quartz



Applications:

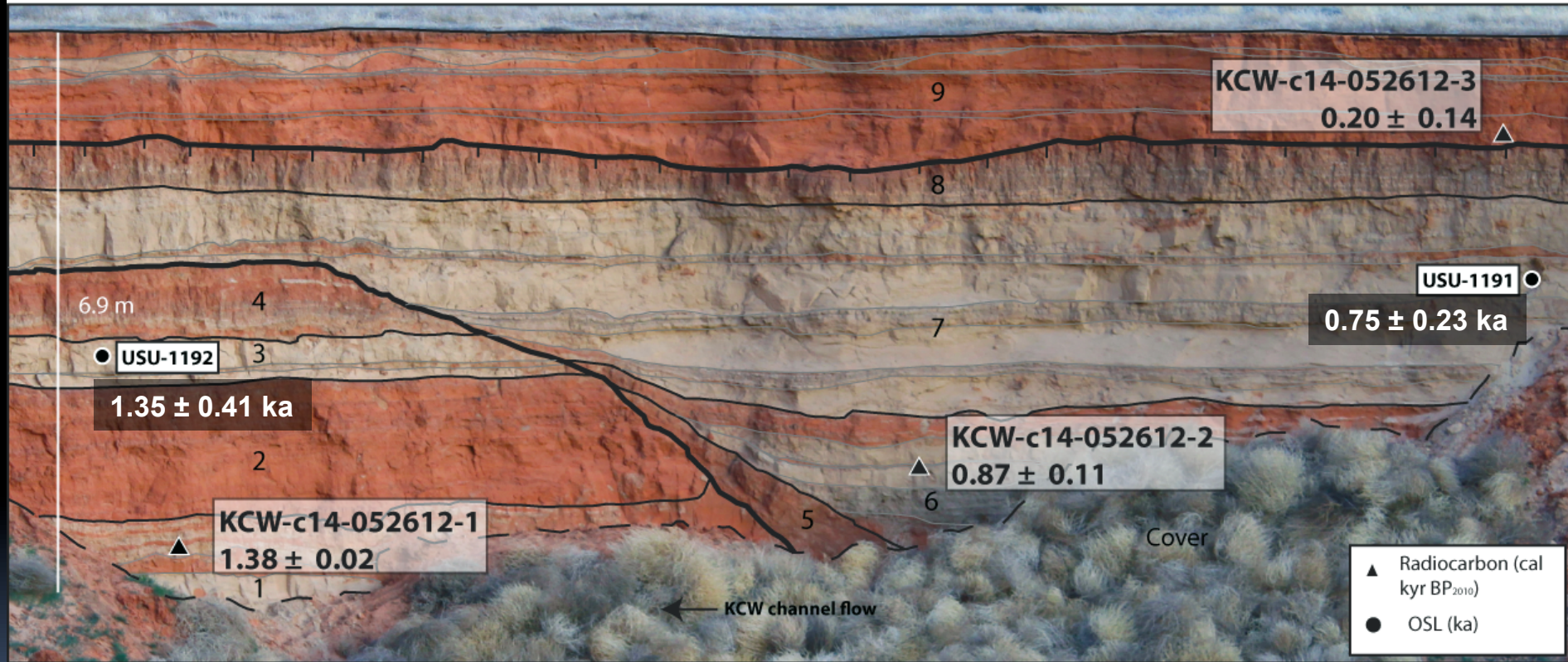
- Dating deposits (terraces, sand dunes, glacial deposits, beach ridges, alluvial fans etc, etc)
- Dating rock surfaces
- Thermochronology

Kitchen Corral Wash

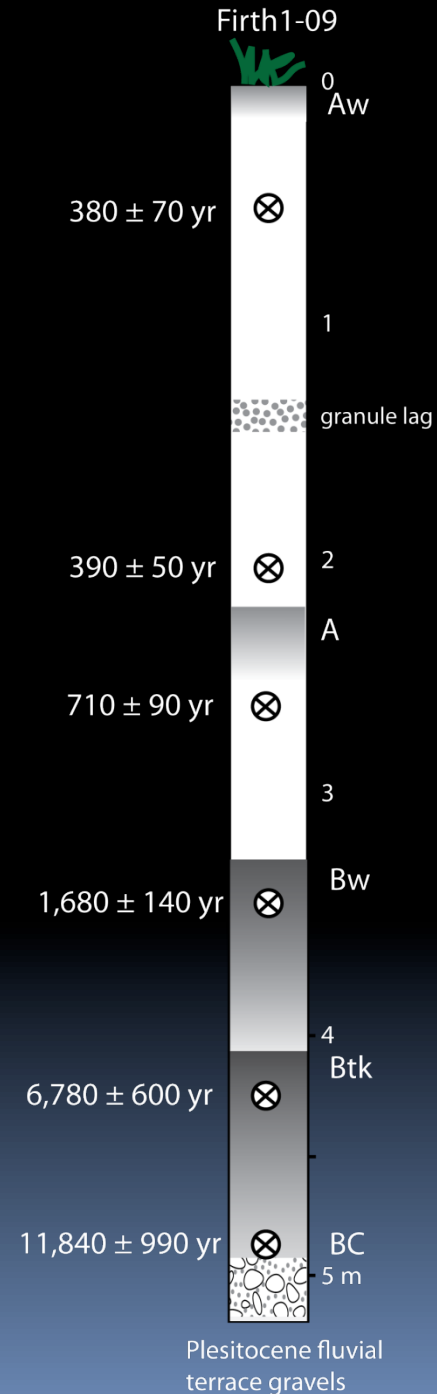
East

KCW-E

1:1 Scale West



Sand Dune Records - Idaho



Timing of Fault Rupture or displacement



Archaeological Research



Slab-lined hearth

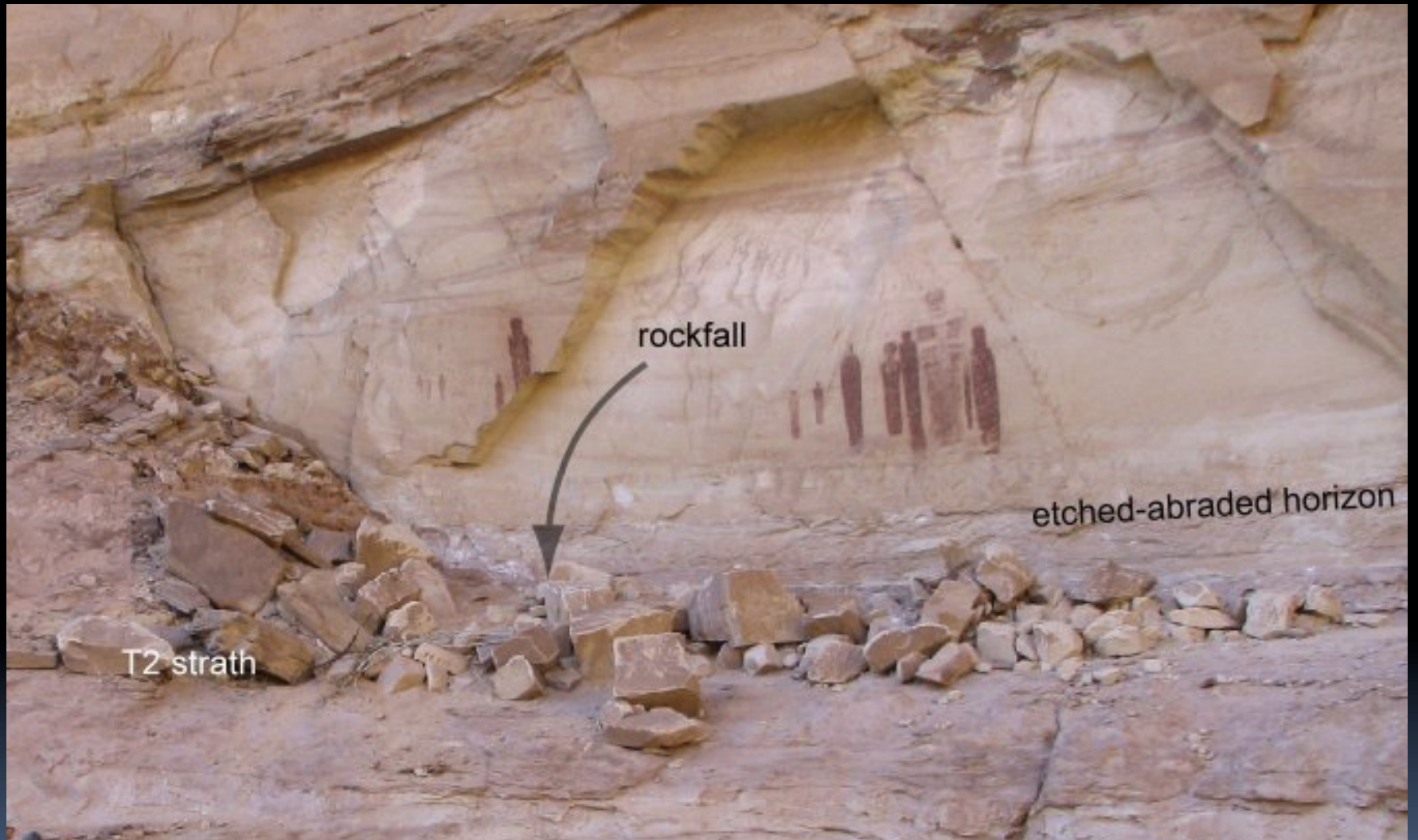
^{14}C sample = 1520-1330 cal yr BP

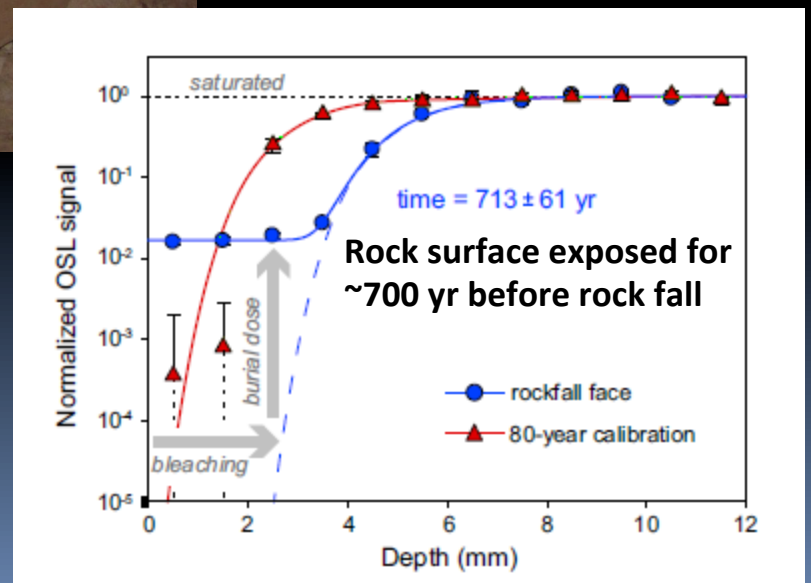
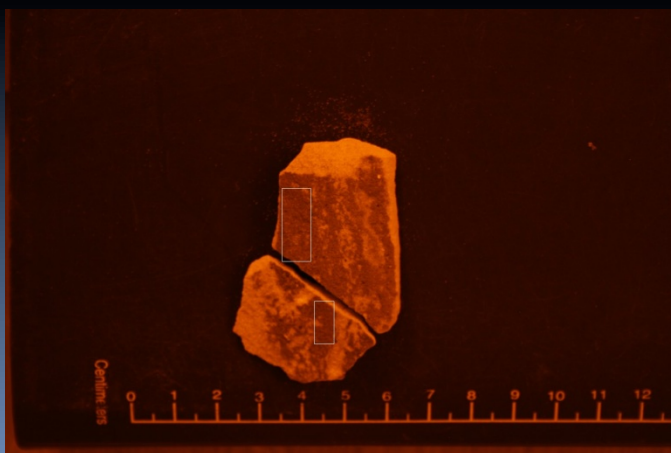
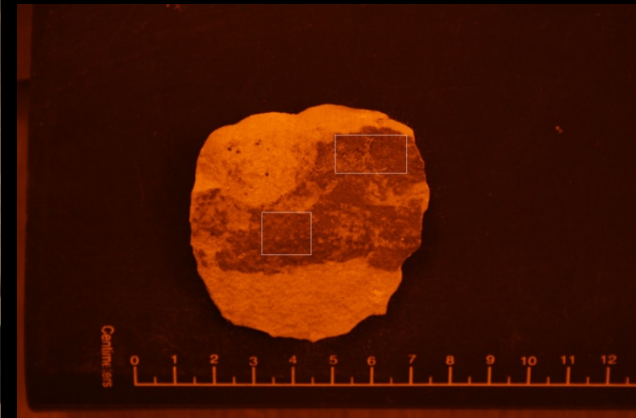
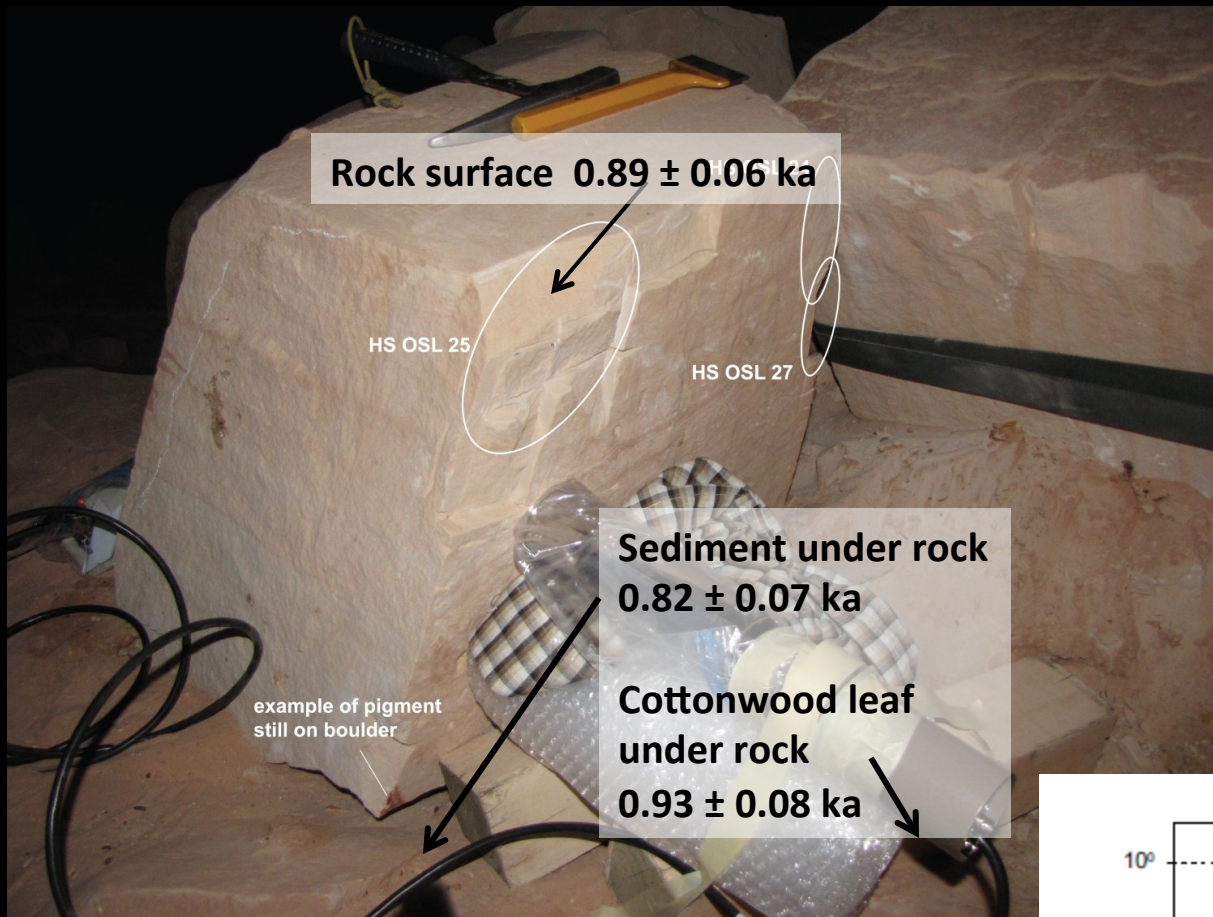


OSL sample = 1480 ± 130 ka

Dating Rock surfaces (exposure dating)







OSL Thermochronology

