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A polarity-induced defect mechanism for conductivity and magnetism at polar-nonpolar oxide interfaces

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The discovery of conductivity and magnetism at the polar-nonpolar interfaces of insulating nonmagnetic oxides such as LaAlO_3 and SrTiO_3 has raised prospects for attaining interfacial functionalities absent in the component materials. Yet, the microscopic origin of such emergent phenomena remains unclear, posing obstacles to design of improved functionalities. Here we present first principles calculations of electronic and defect properties of $\text{LaAlO}_3/\text{SrTiO}_3$ interfaces and reveal a unifying mechanism for the origins of both conductivity and magnetism. We demonstrate that the polar discontinuity across the interface triggers thermodynamically the spontaneous formation of certain defects that in turn cancel the polar field induced by the polar discontinuity. The ionization of the spontaneously formed surface oxygen vacancy defects leads to interface conductivity, whereas the unionized Ti-on-Al antisite defects lead to interface magnetism. The proposed mechanism suggests practical design principles for inducing and controlling both conductivity and magnetism at general polar-nonpolar interfaces.

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Oxide interfaces exhibit many spectacular phenomena not found in the respective bulk components or in conventional semiconductor interfaces¹, providing new avenues for electronics². The LaAlO₃/SrTiO₃ interface is a paradigm example, exhibiting conducting two-dimensional (2D) electron gas (2DEG)^{3,4} and magnetism^{5–11} between two insulating nonmagnetic metal oxides. In the [001] direction, two different interfaces can be formed between polar LaAlO₃, which consists of alternating LaO⁺–(AlO₂)[–] layers, and nonpolar SrTiO₃, which consists of alternating (SrO)⁰–(TiO₂)⁰ layers. One is LaO/TiO₂ stacking configuration (so-called n-type) and the other is AlO₂/SrO configuration (so-called p-type). The remarkable feature is that the conductivity occurs only at n-type interfaces when the LaAlO₃ film thickness (n_{LAO}) is larger than three unit cells (uc)^{4,5}, whereas the magnetism has been observed both at n-type interfaces with $n_{\text{LAO}} > \sim 3$ uc and at insulating p-type interfaces⁸. Table 1 lists some experimental observations representing the main puzzles¹² that need to be resolved before the promised applications can be realized¹³.

For 2DEG at n-type interfaces, four main mechanisms have been suggested, yet no single one explains the full scope of these puzzles. The prevalent one is intrinsic electronic reconstruction (so-called polar catastrophe) involving ionization of the electrons from host valence band of LaAlO₃ within the abrupt and defect-free interfaces (Supplementary Fig. 1)^{3,4}. The other three mechanisms involve various defects, including the oxygen vacancies at the interface (denoted as $V_{\text{O}}(I)$, where ‘I’ means ‘Interface’)^{14–16}, oxygen vacancies at LaAlO₃ overlayer surface (denoted as $V_{\text{O}}(S)$, where S means ‘Surface’)^{17–22}, and the La-on-Sr (La_{Sr}) antisite donor defects induced by interfacial cation intermixing^{23–29}. As Table 1 shows, each of these proposed mechanisms represents one aspect of the interface physics, explains some experimental findings, but conflicts with a few others². None explains the insulating nature of p-type interfaces. Regarding interface magnetism, it was shown experimentally that the local magnetic moments are associated with Ti³⁺ ions^{5–11,30}. However, it is yet unclear whether such Ti³⁺ ions reside in the interface within SrTiO₃ side, or LaAlO₃ side, or both sides. Theoretically, it has been argued that the Ti³⁺ ions arise in SrTiO₃ side, owing to the occupation of the low-energy Ti- d_{xy} -like sub-bands caused by the interfacial splitting of orbital degeneracy³¹, or interfacial disorder^{32,33}, or interfacial oxygen vacancies³⁴. However, these scenarios are difficult to explain the fact that magnetism occurs at p-type interfaces and n-type interfaces with a critical thickness (L_c) similar to that for 2DEG.

The centrosymmetric III–III–O₃ perovskite has a non-zero formal polarization, as established by the modern theory of polarization^{35,36}. The discontinuity in the formal polarization of

LaAlO₃ and SrTiO₃ leads to a finite polar field that would cause the divergence of electrostatic potential as the n_{LAO} increases. A crucial issue associated with the emergent conductivity and magnetism at polar–nonpolar interfaces is what mitigates such potential divergence. Is it electronic reconstruction within the polar catastrophe scenario, or the atomic reconstruction scenario with or without chemical defects? Different mechanisms suggest different experimental designs that would control conductivity, mobility and magnetism. Particularly, for defects, it is unclear which defects can be induced and are responsible for the emergent interface phenomena. Using first principles electronic and defect calculations, we find that the certain defects would form spontaneously in response to the built-in polar field. The ensuing polarity-induced defect mechanism (Fig. 1) simultaneously explains the main features of both conductivity and magnetism at the interface, as summarized in Table 2.

The key defect-related physical quantities that feature in our explanation are (i) the formation energy ΔH of defects in various charge states (q) at the thermodynamic equilibrium Fermi energy E_{F} (Fig. 2). This ΔH controls the equilibrium defect concentration; (ii) the defect charge transition energy levels (deep or shallow; Fig. 3), $\varepsilon(q/q')$ defined as the E_{F} where the ΔH of a defect in two different charge states q and q' equal. A donor can produce electrons and compensate holes, whereas an acceptor can produce holes and compensate electrons. These two quantities (i) and (ii) calculated for charged defects located in different layers across the interfaces turn out to be crucial. The details of their first principles calculations are given in the Methods section.

The central point of the proposed mechanism is that the polar-discontinuity-induced built-in polar field triggers thermo-dynamically the spontaneous formation of certain defects at the surface and/or interface, which in turn compensate the built-in polar field and thus avoids the potential divergence. Thus, it is the polar-field-induced defects, rather than the electronic or atomic reconstruction, that are responsible for the conductivity and magnetism at the interface. Specifically, we find that the surface V_{O} has its donor levels located energetically above the SrTiO₃ conduction band at the interface but below the LaAlO₃ conduction band. This donor level position is a prerequisite for 2DEG formation. Although the occurring of the 2DEG is because of the surface donors, the density of 2DEG is controlled by the interfacial deep acceptor defects (mainly Al-on-Ti antisite). It has also turned out that the interface magnetic moment is caused by the unionized deep Ti-on-Al antisite defects located within the LaAlO₃ side near the interface.

We address below how this polar-field-induced defect mechanism resolves the long-standing puzzles on the origin of 2DEG, the critical thickness for 2DEG, the weak field in LaAlO₃

Table 1 | List of some important experimental observations at LaAlO₃/SrTiO₃ interfaces.

Interface structure	Experimental observations	Polar catastrophe	Cation mixing	V_{O} at interface	V_{O} at surface	Current mechanism
n-type	Critical thickness (L_c) = 4 uc	✓	X	X	?	✓
	2DEG density $< 0.5e S_{2D}^{-1}$	X	?	X	X	✓
	Weak E in LaAlO ₃ for $n_{\text{LAO}} < L_c$	X	?	X	X	✓
	Weak E in LaAlO ₃ for $n_{\text{LAO}} \geq L_c$	X	X	X	✓	✓
	LaAlO ₃ surface: insulating	X	?	?	✓	✓
	Interface: cation intermixed	X	✓	X	X	✓
	Interface magnetism	X	?	X	X	✓
p-type	Interface: insulating	X	?	X	?	✓
	LaAlO ₃ surface: insulating	X	X	?	?	✓
	Interface: cation mixed	X	✓	X	X	✓
	Interface magnetism	X	?	?	X	✓

The symbol of ‘✓’ and ‘X’ mean that the mechanism agrees or disagrees, respectively, with the experimental observation. The ‘?’ symbol denotes uncertainty.

film, the density of 2DEG, the insulating nature of p-type interfaces and the origin of the local magnetic moments. During this process, we also distil the general design principles that control the pertinent effects and could allow future section of better polar–nonpolar interface materials.

Results

The origin of the 2DEG. The 2DEG is unlikely to originate from the defect-free scenarios: these include the ionization of the intrinsic LaAlO₃ valence bands (suggested by the polar catastrophe model^{3,4}) or the ionization of the LaO interface layer (suggested by the interfacial charge-leaking model³⁷ (Supplementary Note 1). This conclusion stems from the fact that the creation of 2DEG in these defect-free scenarios requires the LaAlO₃ valence band maximum (VBM) to cross the SrTiO₃ conduction band minimum (CBM) or E_F. However, this is contrary to the experimentally observed weak field (negligible band-bending)^{38–41} in the LaAlO₃ film, clearly showing that the LaAlO₃ VBM is located energetically far below the E_F.

The 2DEG also is unlikely to originate from interfacial point donor defects (La_{Sr}, Ti_{Al} and V_O). Recall first that the defect formation energy (ΔH) depends on the E_F (or chemical potential) and the defect charge transition energy ε(q/q') needs to be close to band edges in order to produce free carriers. In thermodynamic equilibrium, the E_F of the system pins around the middle of SrTiO₃ band gap when n_{LaO} < L_c and around the SrTiO₃ conduction band edge near the interface when n_{LaO} ≥ L_c (Supplementary Note 2). In either case, Fig. 2ab shows that the ΔH of the interfacial antisite donor defects, La_{Sr}⁰ and Ti_{Al}⁰, is small positive or even negative (note: the superscript denotes the defect charge states, not the nominal oxidation state of the atom at the defect site). In other words, the formation of such antisite defects at the thermodynamic equilibrium E_F is energetically favourable and would inevitably lead to interfacial cation mixing. However, at such E_F, both La_{Sr}⁰ and Ti_{Al}⁰ defects are stable in their charge neutral states (as indicated by the superscript), contributing no free carriers. On the other hand, the interfacial V_O defects are energetically stable in the charged states, that is, V_O²⁺ (Fig. 2a,b). This means that, if formed, the V_O will donate electrons and thereby become positively charged. However, the ΔH of V_O²⁺ at such equilibrium E_F is rather high (>2.5 eV), implying that V_O²⁺ have very low concentration under thermodynamic equilibrium conditions. The high ΔH also means that even if the V_O defects are formed under non-equilibrium growth conditions, they can still be removed easily by the post O-rich annealing process⁴² (Supplementary Note 3). Thus, contrary to earlier postulations, these interfacial donor defects are not responsible for 2DEG, consistent with recent experiments⁴³.

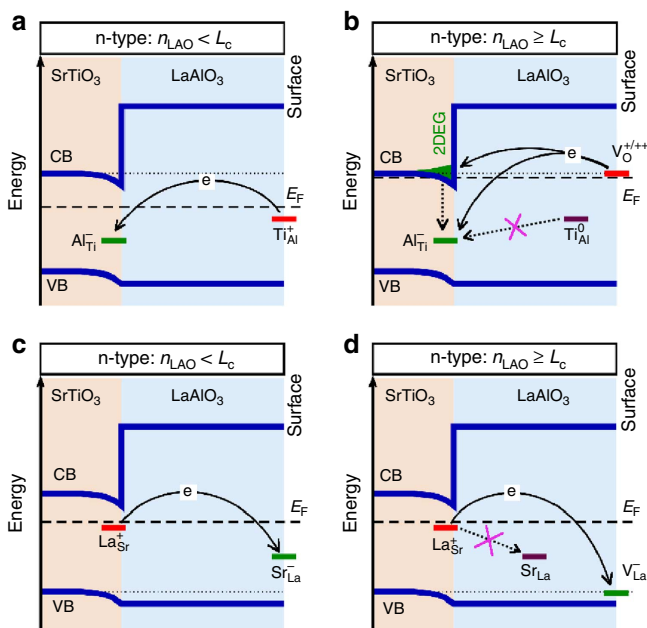


Figure 1 | Schematic band diagram and charge transfer among the defects at LaAlO₃/SrTiO₃ interfaces. (a) n-type interfaces with n_{LaO} < L_c: all electrons transferred from Ti_{Al}(S) are trapped by deep Al_{Ti}(I), causing no 2DEG. (b) n-type interfaces with n_{LaO} ≥ L_c: V_O(S) defects donate ~0.5 e S_{2D}⁻¹ to the interface. Part of ~0.5 e S_{2D}⁻¹ is trapped by the Al_{Ti}(I) and the rest leads to interfacial 2DEG. The formed Ti_{Al} defects are ionized, i.e., Ti³⁺-on-Al³⁺, having local magnetic moments. (c,d) p-type interfaces with n_{LaO} < L_c (~4 uc) and n_{LaO} ≥ L_c: all electrons transferred from La_{Sr}(L) are trapped by Sr_{La}(S) and V_{La}(S), respectively. All involved defects are deep and do not induce carriers. The un-ionized Ti_{Al}⁰ (not shown in c,d) also form and induce local moments. The superscripts (0, +, + +, -) in the Figure denote the defect charge states, not the oxidation states of the ions there.

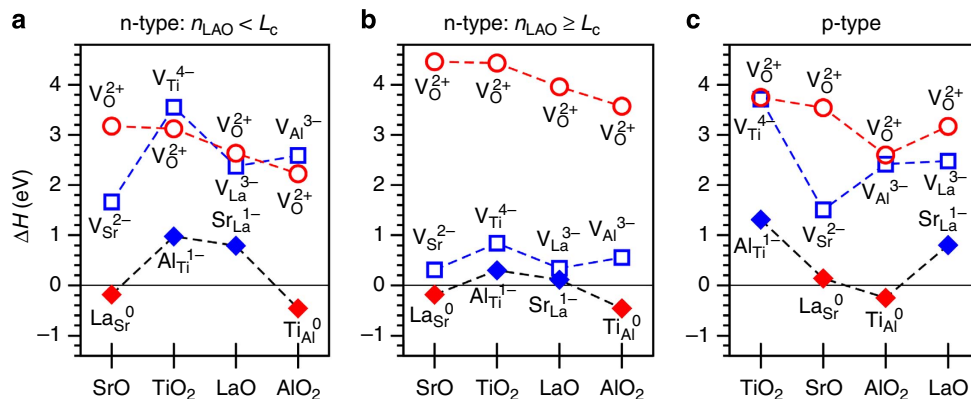


Figure 2 | Formation energy of the interfacial point defects at thermodynamical equilibrium Fermi energy. (a,b) n-type interfaces with n_{LaO} < L_c and n_{LaO} ≥ L_c, respectively. (c) p-type interfaces. At a given E_F, the defect in different charge states (for example, V_O²⁺, V_{Sr}¹⁻, V_{Sr}²⁻) usually has different ΔH and the only one with the lowest ΔH is shown in the Figure. The ΔH versus E_F for these defects are shown in Supplementary Fig. 2, which also includes other high-ΔH defects not shown here. The chemical potentials used for Sr, Ti, La, Al and O are -4.36, -6.20, -6.10, -5.46 and -2.0 eV, respectively, relative to their corresponding elemental solid or gas phases, which corresponds to T = 1050 K and P_{O₂} = 6.1 × 10⁻⁶ Torr (Supplementary Fig. 3).

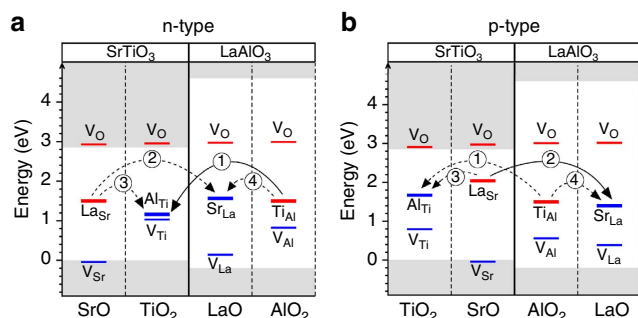


Figure 3 | Charge transition energy levels of the interfacial point defects.

(a) n-type interface. (b) p-type interface. The defect charge transition energy level is defined as the E_F where the ΔH of a given defect in two different charge states equal. Some defects may have multiple charge transition energy levels. For example, V_{Sr} has the two transition energy levels (one is for the transition between neutral charge state and -1 , and the other is between -1 and -2). In such case, if the defect is donor (red), only the lowest level is shown, and if the defect is acceptor (blue), the highest level is shown.

The oxygen vacancy, $V_O(S)$, at the free $LaAlO_3$ surface can explain the interfacial 2DEG. For this to happen, three conditions ('design principles') need to be satisfied. First, $V_O(S)$ in the polar film material needs to have a sufficiently low formation energy ΔH ; therefore, it could form in significant quantities. Figure 4a shows that the ΔH of $V_O(S)$ decreases linearly as the film thickness n_{LaO} increases, consistent with previous calculations^{20,44}. When $n_{LaO} \geq 3-4$ uc, the ΔH becomes zero or negative, and $V_O(S)$ will form spontaneously. The large negative ΔH means that even exposing the surface to air or post annealing under O-rich environment cannot remove these vacancies. Second, the system needs to have a non-zero built-in polar field that would enable the electron to transfer from the surface of the polar material to the interface. Such transfer sets up an opposite dipole (proportional to n_{LaO}), which in turn cancels the field and lowers the ΔH . The larger the n_{LaO} , the lower the ΔH . Note that in the absence of such field, the surface-to-interface charge transfer would not occur since such a transfer would create a dipole that would increase the electrostatic energy (proportional to that dipole) and thus raise the total energy of the system. Third, the donor transition level of $V_O(S)$ in the polar film should be higher in energy than in the substrate ($SrTiO_3$) conduction band edge at the interface (Fig. 3). These three conditions are satisfied in this $LaAlO_3/SrTiO_3$ system.

It is noteworthy that it is the built-in polar field that triggers the spontaneous formation of the $V_O(S)$ when $n_{LaO} \geq L_c$. Such built-in polar field always exists in the $LaAlO_3$ to be grown during the layer-by-layer growth. This is because that the surface defects (here V_O) can cancel the built-in polar field only in the $LaAlO_3$ film between the interface and the surface, not the built-in polar field in the $LaAlO_3$ film to be grown on top of the surface.

In the absence of interfacial defects, the emerging picture for creating 2DEG is that the electrons ionized from $V_O(S)$ of the polar film material transfer to the nonpolar substrate material $SrTiO_3$ conduction bands at the interface via the built-in polar field, thus forming the 2DEG at that interface. This charge transfer in turn cancels the built-in polar field in $LaAlO_3$, which caused the low ΔH of the surface vacancies in the first place. After the built-in field has been cancelled, the ΔH of $V_O(S)$ return to a high value (> 3 eV) characteristic of the bulk, and $V_O(S)$ become again hard to form in thermodynamic equilibrium²⁰. Thus, the theoretical maximum concentration of $V_O(S)$ is $0.25 S_{2D}^{-1}$ (where S_{2D} is 2D unit cell area), that is, one of eight oxygen missing at

surface. These would donate maximally $0.5e S_{2D}^{-1}$ that would completely cancel the polar field in $LaAlO_3$. The compensation of polar field by $V_O(S)$ also means that the band bending in $LaAlO_3$ because of polar field is removed. Thus, the $LaAlO_3$ valence bands fall well below the E_F , contrary to what the polar catastrophe model would suggest. Consequently, no free holes can arise from depopulation of the $LaAlO_3$ valence bands at the surface, consistent with experiments^{3,22}.

The emerging design principles for selecting materials that will form interfacial 2DEG are: (i) the nonpolar material needs to have a CBM positioned in the band gap of the polar material; (ii) the polar material needs to have at least one donor defect with its donor level higher in energy than the conduction band of the nonpolar material at the interface. This picture suggests that the 2DEG at n-type $LaAlO_3/SrTiO_3$ interfaces may also be induced and/or tuned by using certain surface adsorbates (for example, H_2O and H)⁴⁵⁻⁴⁷ or metallic contacts⁴⁸ provided that the ionization energy of the surface adsorbate or the metallic contact is not lower than the donor level of the $V_O(S)$.

The origin of the critical thickness. The linear decrease in ΔH of $V_O(S)$ with increasing polar film thickness n_{LaO} naturally explains the critical thickness L_c for the metal-insulator transition. The calculated rate of decrease (that is, the slope $d\Delta H/dn_{LaO}$) equals $0.19 \text{ eV } \text{\AA}^{-1}$, which is the same as the calculated built-in polar field in the defect-free $LaAlO_3$ film (Supplementary Note 4). The $V_O(S)$ defects start to form spontaneously when the ΔH becomes zero at the L_c of ~ 4 uc under a typical O-rich growth condition (Fig. 4a). For the $LaAlO_3$ film that is 1 uc thinner than this L_c , the calculated ΔH of $V_O(S)$ is 0.75 eV, which is too high to produce significant free carrier concentration. Thus, the appearance of $V_O(S)$ (and the ensuing metal-insulator transition) at L_c is predicted to be a sharp transition (Supplementary Note 5), distinct from the gradual appearance of 2DEG behaviour as predicted from polar catastrophe model, but consistent with experiments⁴⁹.

Figure 3a suggests that the L_c resulting from $V_O(S)$ can be written as $L_c = \Delta H_o/eE_p$, where ΔH_o is the formation energy of the V_O at interface (or the ΔH extrapolated at $n_{LaO} = 0$) and E_p is the built-in polar field. Using $E_p = 4\pi P_o/\epsilon$ (where ϵ and P_o are the dielectric constant and formal polarization of $LaAlO_3$ film), this relation can be written as

$$L_c = \Delta H_o \epsilon / 4\pi e P_o \quad (1)$$

which predicts an L_c of ~ 4 uc, depending slightly on the O-poor/rich growth conditions (Supplementary Note 6). The above formula provides an alternative explanation for the observed variation of the L_c with the fraction x in $(LaAlO_3)_{1-x}(SrTiO_3)_x$ overlayer (where P_o is proportional to x)⁵⁰. This observation was originally explained by $L_c = \Delta\Phi\epsilon/4\pi e P_o$ (where $\Delta\Phi$ is the energy difference between $LaAlO_3$ VBM and $SrTiO_3$ CBM) within polar catastrophe model⁵⁰. Since $\Delta\Phi$ and ΔH_o have accidentally similar value ($\sim 3-4$ eV) in this system, it is not surprising that the L_c predicted from these two models is also similar. However, the $V_O(S)$ model clearly explains many other observations in which the polar catastrophe model fails (Table 1).

Implication on the design of carrier mobility: (i) the relatively high 2DEG mobility could be enabled by a modulated doping effect⁵¹, whereby the source of carriers (here at the $LaAlO_3$ surface) is spatially separated from the location where the carriers reside (here at the $LaAlO_3/SrTiO_3$ interface), thus minimizing carrier scattering by the ionized defects. This minimal spatial separation is measured by the critical thickness L_c . The equation (1) suggests that a large L_c (hence maintaining good mobility) could be achieved by selecting a polar materials with small polarization, large dielectric constant and donor defects

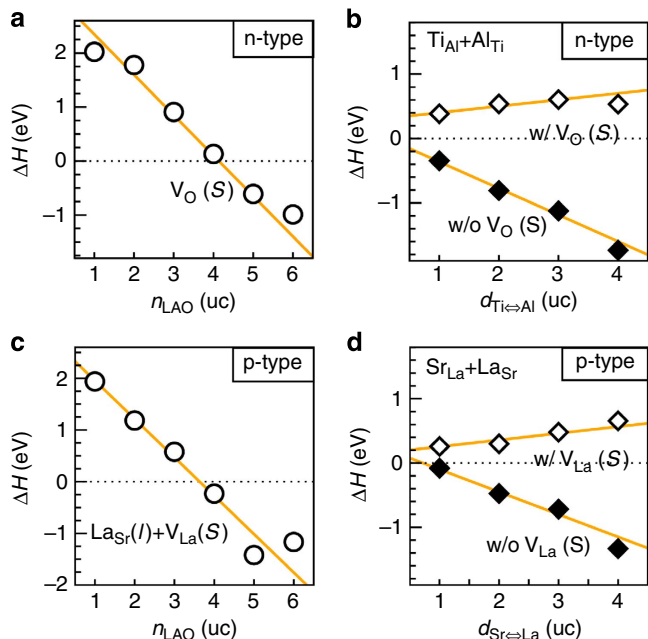


Figure 4 | Properties of surface defects and defect complexes. (a) The GGA-calculated ΔH of $V_{\text{O}}(\text{S})$ defect, under the O-rich growth condition (that is, $\Delta\mu_{\text{O}} = -1.5$ eV, Supplementary Fig. 3a). (b) the ΔH of $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ defect pair created from a $\text{Ti} \leftrightarrow \text{Al}$ exchange out of the ideal interface with and without a $V_{\text{O}}(\text{S})$ in a 2×2 $(\text{SrTiO}_3)_6/(\text{LaAlO}_3)_4/\text{vacuum}$ supercell. (c) The GGA-calculated ΔH of $[\text{La}_{\text{Sr}}(\text{I}) + \text{V}_{\text{La}}(\text{S})]$ defect complex as a function of n_{LaO} , under $\Delta\mu_{\text{Sr}} = -4.36$ eV (Supplementary Fig. 3b). (d) the ΔH of $[\text{La}_{\text{Sr}} + \text{Sr}_{\text{La}}]$ defect pair created from a $\text{La} \leftrightarrow \text{Sr}$ exchange out of the ideal interface with and without a $V_{\text{La}}(\text{S})$ in a 2×2 $(\text{SrTiO}_3)_6/(\text{LaAlO}_3)_4/\text{vacuum}$ supercell, respectively. The $d_{\text{Ti} \leftrightarrow \text{Al}}$ and $d_{\text{La} \leftrightarrow \text{Sr}}$ in **b,d** are the distance between the components of corresponding defect pair. The orange lines are the guides to the eye.

having high ΔH at the interface or in the bulk. On the other hand, (ii) the concentration of interfacial defects must be minimized in order to take advantage of (i). In addition, (iii) since the 2DEG is located at the conduction bands of the nonpolar material, it is advantageous to select the nonpolar material with low electron effective mass in order to achieve higher mobility.

Polar field compensation. Experimentally, only very weak residual field has been observed in the LaAlO_3 film no matter whether its thickness is below or above the L_c (refs 38–41,52). This observation cannot be explained within the defect-free interface scenario, even including the ionic relaxations⁵³. In turn, whereas the $V_{\text{O}}(\text{S})$ model explains the weak electric field in LaAlO_3 film above the L_c , it does not explain it below the L_c . This leads us to inspect the effects of all possible cation antisite defects across the interface.

Each individual interfacial antisite alone cannot cancel the polar field. Figure 2ab shows that the La_{Sr} , Sr_{La} , Ti_{Al} and Al_{Ti} antisite defects have lower ΔH than other point defects (for example, cation vacancies) in the layer where they are located. Therefore, the former are the dominant defects in their corresponding layers. The interfacial La_{Sr} donor in the SrTiO_3 side cannot set up an opposite dipole across the LaAlO_3 film that can cancel the polar field inside the LaAlO_3 film. Regarding the Ti_{Al} donor in the LaAlO_3 side, the donor level is lower than the SrTiO_3 conduction band at the interface. Therefore, the ionized electrons cannot be transferred to the latter so as to cancel the

polar field. Regarding the interfacial Al_{Ti} and Sr_{La} acceptors, the polar field compensation is similar to that in the polar catastrophe model: before the LaAlO_3 VBM reaches the acceptor levels of Al_{Ti} or Sr_{La} , the polar field cannot be cancelled.

The $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ defect pair is the most potent source of polar field cancellation among those donor–acceptor antisite defect pairs at n-type interfaces. The four leading antisite defects can form four types of donor–acceptor pairs: $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$, $[\text{La}_{\text{Sr}} + \text{Sr}_{\text{La}}]$, $[\text{La}_{\text{Sr}} + \text{Al}_{\text{Ti}}]$ and $[\text{Ti}_{\text{Al}} + \text{Sr}_{\text{La}}]$, denoted as ①, ②, ③ and ④, respectively, in Fig. 3. Clearly, the electron transfer from donor to acceptor in both pairs ② and ③ is unlikely since it will create a dipole in the same direction as the intrinsic dipole in LaAlO_3 , and thus increase the dipole moment (also the electrostatic energy) and destabilize the interface. In pairs ① and ④, the charge transfer can cancel the polar field. However, the electron transfer in pair ① is energetically much more favourable because, first, Al_{Ti} has a lower acceptor level than Sr_{La} and, second, the donor–acceptor separation distance (also the associated opposite dipole moment that lowers the total energy of the system) is larger in pair ① (Fig. 3a). We thus next focus on $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ (that is, pair ①).

For $n_{\text{LaO}} < L_c$, the $[\text{Al}_{\text{Ti}} + \text{Ti}_{\text{Al}}]$ antisite pair can form spontaneously via $\text{Ti} \leftrightarrow \text{Al}$ exchange across the interface and cancel the polar field. Figure 4b (filled symbols) shows that the energy required to form such defect pair is negative (that is, exothermic), and the largest energy gain is obtained when a Ti atom of TiO_2 -interface monolayer is exchanged with an Al of AlO_2 -surface monolayer, that is, $\text{Al}_{\text{Ti}}(\text{I}) + \text{Ti}_{\text{Al}}(\text{S})$, which is consistent with previous first principles calculations⁵⁴. This means that Ti atom at the interface would hop to the AlO_2 -surface layer and exchange with Al atom there. Similar to the case of $V_{\text{O}}(\text{S})$, the linear decrease in ΔH with increasing donor–acceptor separating distance (Fig. 4b) is a sign of polar field compensation. Indeed, the electron transfer from the Ti_{Al} donor to the Al_{Ti} acceptor is expected, since the donor level is higher in energy than the acceptor level (Fig. 3a). Figure 4a also shows that the $V_{\text{O}}(\text{S})$ has too high ΔH to form for $n_{\text{LaO}} < L_c$ (Fig. 4a). Therefore, the polar field is cancelled by those spontaneously formed $[\text{Al}_{\text{Ti}}(\text{I}) + \text{Ti}_{\text{Al}}(\text{S})]$ pairs. On the other hand, since these defects are deep, they cannot cause free carriers in the both interface and surface regions (whence insulating).

For $n_{\text{LaO}} \geq L_c$, the polar field is cancelled by spontaneously formed $V_{\text{O}}(\text{S})$, not by $[\text{Al}_{\text{Ti}}(\text{I}) + \text{Ti}_{\text{Al}}(\text{S})]$. Recall that the polar field always exists in the LaAlO_3 layers during the layer-by-layer growth. Such polar field can trigger the formation of $V_{\text{O}}(\text{S})$ and/or $\text{Ti}_{\text{Al}}(\text{S})$ defects as n_{LaO} increases. For $n_{\text{LaO}} \geq L_c$, both $V_{\text{O}}(\text{S})$ (Fig. 4a) and $[\text{Al}_{\text{Ti}}(\text{I}) + \text{Ti}_{\text{Al}}(\text{S})]$ pair (Fig. 4b) have zero or negative ΔH , meaning that both could form in ideal interfaces. However, if both $V_{\text{O}}(\text{S})$ and $\text{Ti}_{\text{Al}}(\text{S})$ are present, since V_{O} has an energetically higher donor level than Ti_{Al} (Fig. 3a), the $V_{\text{O}}(\text{S})$ would transfer electrons to the Ti_{Al} defects. The polar field that was initially cancelled by the electrons transferred from the Ti_{Al} defects are then released and get cancelled by the electrons transferred from $V_{\text{O}}(\text{S})$ defects. Consequently, the polar field in the whole LaAlO_3 film would be cancelled by the $V_{\text{O}}(\text{S})$ defects (if present). The larger the n_{LaO} , the lower the ΔH of the $V_{\text{O}}(\text{S})$. After the polar field has been cancelled by $V_{\text{O}}(\text{S})$, Fig. 4b (open symbols) shows that the ΔH of $[\text{Al}_{\text{Ti}}(\text{I}) + \text{Ti}_{\text{Al}}(\text{S})]$ pair becomes positive (0.4–0.7 eV), meaning that $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ pairs cannot be formed via $\text{Ti} \leftrightarrow \text{Al}$ exchange over a distance beyond L_c . In brief, the presence of the $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ defect pairs in the sample does not change the linear-decreasing behaviour in the ΔH of the $V_{\text{O}}(\text{S})$ (Fig. 4a), suggesting that the metal–insulator transition still occurs at the L_c of ~ 4 uc. However, the presence of $V_{\text{O}}(\text{S})$ would prevent $[\text{Ti}_{\text{Al}} + \text{Al}_{\text{Ti}}]$ pairs forming further above the L_c , and reduce the concentration of these pairs formed below the L_c .

The density of the 2DEG. Reinterpretation of the puzzle: According to Gauss' law, the experimentally observed weak electric field in LaAlO₃ film means that the total external charge density (mobile and/or immobile) at the interface must be $\sim 0.5e S_{2D}^{-1}$ (Supplementary Note 7), as recently observed⁵⁵. For $n_{LAO} < L_c$ there is no interfacial conductivity and thus none of these interfacial charge contribute to the conductivity. For $n_{LAO} \geq L_c$, only a fraction of $0.5e S_{2D}^{-1}$ interfacial charge is seen in transport, and so the majority of $0.5e S_{2D}^{-1}$ charges do not contribute to the conductivity. The puzzle thus is why the $\sim 0.5e S_{2D}^{-1}$ charge exists at the interface with any n_{LAO} , but only a small part of it contributes to conducting 2DEG when $n_{LAO} \geq L_c$.

This puzzle cannot be explained by defect-free polar catastrophe model^{3,4} or interfacial charge-leaking model³⁷, since both predict zero interfacial charge for $n_{LAO} < L_c$ and an interfacial charge density much higher than the measured 2DEG density for $n_{LAO} \geq L_c$ (Supplementary Fig. 1). The possibility of 'multiple carrier types' at defect-free interfaces (that is, those electrons occupying interfacial d_{xy} sub-band and those occupying d_{xz}/d_{yz} sub-bands contribute differently in transport) has also been suggested to explain the measured 2DEG density above the L_c (refs 56–59). However, this scenario could not explain the total $0.5e S_{2D}^{-1}$ interface charge that is independent of n_{LAO} . Moreover, it is also difficult to explain why a full carrier density of $0.5e S_{2D}^{-1}$ has been observed at GdTiO₃/SrTiO₃ interfaces (where the same multiple carrier types exist)⁶⁰.

The 2DEG density is controlled by the concentration of immobile acceptor defects that can trap itinerant electrons. Within the emerging defect picture, the total interfacial charge is always $\sim 0.5e S_{2D}^{-1}$, which corresponds to the (almost) complete polar field cancellation. In the SrTiO₃ side (where the 2DEG is located), there are mainly three types of acceptor defects, namely, Al_{Ti}, V_{Sr} and V_{Ti}. At equilibrium E_F , Fig. 2a,b shows that these acceptor defects all prefer to stay in negative charge states, that is, Al_{Ti}¹⁻, V_{Sr}²⁻ and V_{Ti}⁴⁻. (In other defect charge states, these defects have much higher ΔH and are not shown in Fig. 2a,b). This means that once these defects form they will trap free electrons from the system and get negatively charged. Among these acceptor defects, the Al_{Ti}¹⁻ acceptors have the lowest ΔH and thus they the most potent electron-trapping agents. For $n_{LAO} < L_c$, the Al_{Ti} defects resulting from Ti \leftrightarrow Al exchange trap all free electrons transferred from Ti_{Al}(S) defects, and hence no free carrier can occur. For $n_{LAO} \geq L_c$, the ΔH of [Ti_{Al} + Al_{Ti}] pair changes from negative to positive because of V_O(S) (Fig. 4b), meaning that the concentration of Al_{Ti} defect resulting from Ti \leftrightarrow Al exchange is reduced, compared with that formed below the L_c . Therefore, the Al_{Ti} defect concentration is not sufficient to trap all $0.5e S_{2D}^{-1}$ electrons transferred from V_O(S). Therefore, only a small fraction of $0.5e S_{2D}^{-1}$ can contribute to interface 2DEG.

The recently observed LaAlO₃ cation-stoichiometry effect on 2DEG formation⁴³ may also be understood within the above picture. For Al-rich LaAlO₃ film, where both A-site and B-site sublattices are fully occupied (hence having no cation vacancies), the Al_{Ti} antisites are the only electron-trapping defects and the incomplete trapping of $0.5e S_{2D}^{-1}$ interface charge by Al_{Ti} defects leads to interface conductivity. However, for La-rich LaAlO₃ film, where B-site sublattice is not fully occupied, the cation vacancies (V_{Ti} and V_{Al}) also become the main electron-trapping agents, in addition to Al_{Ti}(I). Although the concentration of Al_{Ti} is reduced, each cation vacancy induced in the La-rich film traps more electrons than an Al_{Ti}. The insulating character can thus be then attributed to the complete interfacial electron trapping by both interfacial cation vacancies and Al_{Ti}(I).

The picture of Al_{Ti}(I) as the main electron-trapping agent may be extended to SrTiO₃/GdTiO₃ interfaces. The observed full

carrier density of $0.5e S_{2D}^{-1}$ there⁶⁰ can be ascribed to the fact that both SrTiO₃ and GdTiO₃ have the same Ti atom at B-site sublattice and thus have no Al_{Ti}-like antisite defects at the interface.

Implication on how to increase the density of 2DEG: The above picture suggests that the main controlling factor for the interface carrier density is the concentration of the acceptor defects (mainly Al_{Ti} in stoichiometric or Al-rich film), which should be minimized for enhancing carrier density. Such Al_{Ti}-like electron-trapping defects may be completely removed by designing other oxide interfaces such as GdTiO₃/SrTiO₃ interfaces, whose bulk components have a common cation atom with multiple valence states.

The origin of the insulting nature of p-type interfaces. An intriguing fact is that the so-called p-type interfaces are not p-type (hole) conducting but are actually insulating. The defect-free polar-catastrophe model for p-type interface predicts a hole-conducting interface and an electron-conducting surface when $n_{LAO} > \sim 7.3$ uc (Supplementary Fig. 1) in contradiction with the insulating behaviour observed robustly in experiment. To explain this, defects must be involved. The emerging defect picture below differs from the literature model based solely on interfacial hole-polaron⁶¹ or interfacial hole-compensating V_O defects^{23,44}, which assumes that the interface has holes arising from the depopulation of the intrinsic SrTiO₃ valence bands.

Individual point defects alone at p-type interfaces can neither cause conductivity nor cancel the polar field. As was the case for n-type interfaces, Fig. 2b shows that the interfacial La_{Sr} and Ti_{Al} are stable at their charge neutral states and have negligible or negative ΔH at equilibrium E_F . This means that they cause inevitable interfacial cation intermixing but induce no free carriers. The V_O and other defects at the interface require too high ΔH to form, and thus they do not produce free carriers either. For similar reason, each of the leading antisite defects (La_{Sr}, Sr_{La}, Ti_{Al} and Al_{Ti}) alone at p-type interfaces cannot cancel the polar field.

The spontaneously formed donor-acceptor defect pairs always cancel the polar field but do not induce free carriers. Among the four donor-acceptor defect pairs as indicated in Fig. 3b, the [La_{Sr} + Sr_{La}] (that is, pair ②) is energetically most favourable to cause polar field cancellation. For $n_{LAO} < \sim 4$ uc, the [La_{Sr}(I) + Sr_{La}(S)] pairs have negative ΔH (Fig. 4d) and can form spontaneously via La \leftrightarrow Sr exchanges, whereas the [La_{Sr}(I) + V_{La}(S)] have too high ΔH to form (Fig. 4c). Therefore, the polar field is cancelled by the charge transfer from La_{Sr}(I) to Sr_{La}(S), which can be expected from their relative defect levels (Fig. 3b) and their linear decreasing behaviour in ΔH as a function of n_{LAO} (Fig. 4d). For $n_{LAO} \geq \sim 4$ uc, the ΔH of [La_{Sr}(I) + V_{La}(S)] become negative (Fig. 4c) and can also form spontaneously. Since V_{La} has a lower acceptor level than Sr_{La} (Fig. 3b), the polar field is cancelled by the charge transfer from La_{Sr}(I) to V_{La}(S), rather than to Sr_{La}(S). In absence of an electric field, Fig. 4d (open symbols) indicates that the La \leftrightarrow Sr exchanges cannot occur anymore over a distance of ~ 4 uc. Unlike the case in n-type interfaces, the V_O(S) defects in p-type interface always have too high ΔH to form. The defects involved in polar field cancellation are all deep. The calculated equilibrium E_F according to those point defects turns out to be always around the middle of SrTiO₃ band gap. This means that both VBM and CBM are far away from the E_F , and there are no free carrier arising from the depopulation of VBM and CBM in both interface and surface regions (whence insulating).

Implication on the design of 2D hole conductivity: Clearly, the formation of interfacial free holes is prevented by these

spontaneously formed deep La_{Sr} defects that have donor level higher than the VBM at the interface. Therefore, to induce interfacial hole conductivity, one should search for the polar–nonpolar interfaces where all such donors have high enough formation energy to form or (ii) their donor levels below the VBM at the interface. Practically, the (ii) may be achieved more easily by searching for the polar material whose VBM is higher than the charge transition energy levels of those spontaneously formed interfacial donor defects.

The origin of interface magnetism. Distinct from previous models^{31–34} that explain magnetism based on the intrinsic interfacial Ti^{3+} ion in the SrTiO_3 (that is, not a defect), we find below that the local magnetic moment originates from the unionized deep Ti_{Al} antisite defect (that is, Ti^{3+} -on- Al^{3+} within LaAlO_3 side near the interface. The interface magnetism depends on the concentration and spatial distribution of such Ti_{Al} defects. This picture explains not only why the magnetism appears at n-type interfaces with a similar critical thickness to that for 2DEG but also why the magnetism also appears at insulating p-type interfaces⁸.

What causes local moment? As discussed earlier, for n-type interfaces, when $n_{\text{LaO}} < L_c$, the polar field in LaAlO_3 is cancelled by the charge transfer from $\text{Ti}_{\text{Al}}(\text{S})$ defects to the interface. These formed Ti_{Al} defects are thus ionized, that is, $\text{Ti}_{\text{Al}}^{4+}$ (where superscript denotes the defect charge states). The Ti ion at this defect site has the oxidation states of $4+$, denoted as Ti^{4+} , which has no local magnetic moment. Moreover, noted before, when $n_{\text{LaO}} \geq L_c$, the polar field in LaAlO_3 is cancelled by the charge transfer to the interface from $\text{V}_{\text{O}}(\text{S})$ instead of Ti_{Al} . In absence of internal field, all $\text{Ti}_{\text{Al}}(\text{I})$ defects in the LaAlO_3 film are most stable in their charge neutral (or unionized) states, that is, Ti_{Al}^0 , where Ti appears as Ti^{3+} oxidation state, having a finite local magnetic moment. Therefore, the interface magnetism at n-type interfaces because of those unionized Ti_{Al}^0 defects should also have a critical thickness of ~ 4 uc. For p-type interfaces, it is the charge transfer among the defects other than Ti_{Al} defects that cancels the polar field in LaAlO_3 . Thus, all Ti_{Al} defects formed there are not ionized, having local magnetic moments, and cause interface magnetism.

The magnitude of local magnetic moment: The local moment of a single Ti_{Al} defect at the interface can be estimated from that in bulk LaAlO_3 , which is $0.84\mu_{\text{B}}$ from our hybrid functional calculation. For ferromagnetic order as observed in the experiment, the total interface magnetic moment depends on the concentration of unionized Ti_{Al} defects in LaAlO_3 and can be

very small per Ti atom in average. The experimentally observed inhomogeneous landscape of magnetism that also varies from sample to sample^{8,9} may be attributed to the various spatial distributions of Ti_{Al} defects, which may be sensitive to sample preparation conditions (such as temperature and P_{O_2}) and local strain.

The $\text{Ti}_{\text{Al}}(\text{I})$ defects within LaAlO_3 side being the origin of the local moment are more reasonable than $\text{V}_{\text{O}}(\text{I})$ in two aspects. First, the deep Ti_{Al} defect is spatially localized and has an unambiguous local moment. In contrast, V_{O} is a shallow donor that transfer electrons to the lower-energy interfacial Ti d_{xy} subbands that have light effective mass inside the interface plane⁵⁹; therefore, the resulting Ti^{3+} may then be itinerant. Second, the Ti_{Al} defects would form readily because of the small or negative ΔH of Ti_{Al} , whereas the interfacial V_{O} requires significant energy to form and if formed it may be removed completely after annealing.

Discussion

We establish a physical link between polar discontinuity and defect formation: the polar discontinuity triggers spontaneous formation of certain defects that in turn cancel the polar field induced by polar discontinuity. It is the subtle interplay of those spontaneously formed surface vacancy defects and interfacial cation antisite defects that control the physics of the system by their formation energies and relative defect levels. Table 2 summarizes how those defects shown in Fig. 1 explain the leading experimental observations and puzzles in Table 1. The explanation leads to a set of design principles for both conductivity and magnetism at $\text{LaAlO}_3/\text{SrTiO}_3$ and other polar–nonpolar interfaces and enables the design of better polar–nonpolar interfaces.

Having ruled out the electronic reconstruction, interfacial V_{O} and interfacial cation intermixing mechanism as the possible origin of 2DEG in our calculations, we conclude that the 2DEG at n-type interfaces with $n_{\text{LaO}} \geq L_c$ originates from the spontaneously formed $\text{V}_{\text{O}}(\text{S})$ defects. This conclusion stems from the finding that the donor level of deep V_{O} in the LaAlO_3 side is higher than the SrTiO_3 conduction band edge at the interface. This finding explains why the formation energy of $\text{V}_{\text{O}}(\text{S})$ decreases linearly as n_{LaO} increases. This linear decrease relation leads to some new controlling parameters for the critical thickness of sharp metal–insulator transition in absence of the electric field in the polar LaAlO_3 film. Instead of causing the 2DEG, the anti site defect pair turns out to play a key role in canceling the polar field, controlling the density of the 2DEG, and inducing the local magnetic moments at the interface (Table 2).

The emerging mechanism provides three distinctive predictions to be tested in experiment as further validation. (i) For

Table 2 | The specific defects and their charge transfer processes that explain the leading experimental observations at stoichiometric $\text{LaAlO}_3/\text{SrTiO}_3$ interfaces.

	n-type interface structure		p-type interface structure	
	$n_{\text{LaO}} < L_c$	$n_{\text{LaO}} \geq L_c$	$n_{\text{LaO}} < L_c$	$n_{\text{LaO}} \geq L_c$
Polar field compensation	$\text{Ti}_{\text{Al}}(\text{S}) \rightarrow \text{Al}_{\text{Ti}}(\text{I})$	$\text{V}_{\text{O}}(\text{S}) \rightarrow \text{Al}_{\text{Ti}}(\text{I})$ and $\text{V}_{\text{O}}(\text{S}) \rightarrow \text{CBM}(\text{I})$	$\text{La}_{\text{Sr}}(\text{I}) \rightarrow \text{Sr}_{\text{La}}(\text{S})$	$\text{La}_{\text{Sr}}(\text{I}) \rightarrow \text{V}_{\text{La}}(\text{S})$
Origin of 2DEG/2DHG	No 2DEG: $\text{Al}_{\text{Ti}}(\text{I})$ traps all electrons from $\text{Ti}_{\text{Al}}(\text{S})$	$\text{V}_{\text{O}}(\text{S}) \rightarrow \text{CBM}(\text{I})$: $\text{Al}_{\text{Ti}}(\text{I})$ traps part of electrons from $\text{V}_{\text{O}}(\text{S})$	No 2DHG: $\text{La}_{\text{Sr}}(\text{I})$ traps all holes from $\text{Sr}_{\text{La}}(\text{S})$	No 2DHG: $\text{La}_{\text{Sr}}(\text{I})$ traps all holes from $\text{V}_{\text{La}}(\text{S})$
Density of 2DEG/2DHG	Zero	$< 0.5 e S_{2\text{D}}^{-1}$	Zero	Zero
Origin of critical thickness	Polar-field induced $\text{V}_{\text{O}}(\text{S})$ formation		Polar-field induced $\text{V}_{\text{La}}(\text{S})$ formation	
Origin of interface magnetism	Ti^{4+} -on- Al^{3+} forms but has no local moment	Ti^{3+} -on- Al^{3+} forms and induces local moment	Ti^{3+} -on- Al^{3+} forms and induces local moment	Ti^{3+} -on- Al^{3+} forms and induces local moment

The “S” and “I” denote the LaAlO_3 free surface and $\text{LaAlO}_3/\text{SrTiO}_3$ interface, respectively. $S_{2\text{D}}$ is the two-dimensional unit cell area.

n-type interfaces, the AlO_2 -surface layer is dominated by Ti_{Al} defects when $n_{\text{LaO}} < L_c$ and by V_{O} defect when $n_{\text{LaO}} \geq L_c$. (ii) For p-type interfaces, the LaO -surface layer is dominated by Sr_{La} and V_{La} defects, respectively, below and above an L_c of ~ 4 uc. (iii) Ti^{4+} and Ti^{3+} signals exist in both sides of the interface. The appearance of the Ti^{3+} signals should not be taken as a sign of conductivity. Whether the Ti^{3+} signals detected by photoemission below the L_c (refs 21,40,62,63) can be truly assigned to those Ti^{3+} ions in the SrTiO_3 side should be revisited carefully. How these Ti_{Al} local moments are ordered (ferromagnetic, or antiferromagnetic, or else) and whether and how they interact with the itinerant 2DEG are still open questions that should be investigated further.

Methods

Computational techniques. All calculations were performed using density functional theory and plane-wave projector-augmented wave⁶⁴ method as implemented in the VASP code⁶⁵. An energy cutoff of 400 eV was used. The Brillouin zone was sampled by $8 \times 8 \times 1$ and $4 \times 4 \times 1$ k-point mesh for 1×1 and 2×2 in-plane supercell, respectively. The atomic forces were relaxed to be less than $0.03 \text{ eV } \text{\AA}^{-1}$. The in-plane lattice constant was fixed to 3.943 \AA (the relaxed lattice constant of SrTiO_3 by GGA⁶⁶). In slab calculations, the 4-uc ($\sim 16 \text{ \AA}$) vacuum layer was used and the dipole correction was always applied to remove artificial dipole interactions⁶⁷. The results in Figs 2 and 3 were obtained by using HSE hybrid functional⁶⁸ on top of the GGA-relaxed structures.

First principles defect theory. The formation energy of a defect (D) calculated from $\Delta H_{\text{D}}^q(E_{\text{F}}, \mu) = E_{\text{D}}^q - E_{\text{H}} + \sum_{\alpha} n_{\alpha} (\mu_{\alpha}^{\text{gas}} + \Delta \mu_{\alpha}) + q(E_{\text{V}} + E_{\text{F}})$, where E_{D}^q and E_{H} are the total energies of a supercell with and without defect, respectively, and D is in charge state q . n_{α} is the number of atoms of species α needed to create a defect. E_{F} is the Fermi energy relative to VBM (E_{v}). $\Delta \mu_{\alpha}$ is the relative chemical potential of species α with respect to its elemental solid (gas; μ^{0}). The equilibrium Fermi energy was calculated self-consistently according to charge neutrality condition⁶⁹. The chemical potentials relative to their elemental solid (or gas) phase are taken as variables and are bounded by the values that maintain a stable host compound and avoid the formation of other competing phases in thermodynamic equilibrium (Supplementary Fig. 3). The details of theory and calculations can be found in ref. 70.

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Author contributions

L.Y. carried out the calculations, analysed the results and wrote the paper. A.Z. initiated the study of this topic and contributed to the analysis of the results and to the writing of the paper.

Additional information

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