

Fermions: even protons and odd number of neutrons $I = \frac{1}{2}$

two valence electrons \Rightarrow even protons \Rightarrow odd neutrons

Bosons: even protons, even neutrons $\Rightarrow I = 0$

Remember $2s+1$

Ground 1S_0 1) singlet 3P_0 1) Triplet
2) $L=0$ 2) $L=1$
3) $J=0$ 3) $J=0$

mHz vs MHz : 9 orders of magnitude narrower

Why: $J=0 \leftrightarrow J=0$ forbidden
Selection rules.

$S=0 \leftrightarrow S=1$ spin forbidden

Then how is that possible?

Nuclear spin: in fermionic isotopes

Hyperfine structure:

$$H = A \mathbf{I} \cdot \mathbf{J}$$

$${}^3P_0, {}^3P_1, {}^3P_2, {}^1P_1$$

$$\begin{array}{ccc} \downarrow & \downarrow & \swarrow \\ F=9/2 & F=9/2, 11/2, 7/2 & \end{array}$$

• Ramsey Spectroscopy

$$H = \hbar \Omega S_x + \hbar \delta S_z \quad \Omega \gg \delta$$

$$\langle \vec{S} \rangle = \frac{1}{2} \{ 0, \sin \theta, -\cos \theta \} \quad \theta = \Omega t$$

pulse area

(1) Rotating frame:

$$H = \hbar \omega_0 |e\rangle\langle e| + \frac{\hbar \Omega}{2} (\sigma^+ e^{-i\omega_L t} + \sigma^- e^{i\omega_L t})$$

$$\hat{U} = e^{-i\omega_L t \sigma_z / 2}$$

$$U^\dagger \sigma^\pm \hat{U} = \sigma^\pm e^{\pm i\omega_L t}$$

$$\frac{\partial \Psi}{\partial t} = U^\dagger H U + i\hbar \left(\frac{\partial U^\dagger}{\partial t} \right) U$$

$$H_{\text{eff}} = \hbar(\omega_0 - \omega_L) |e\rangle\langle e| + \frac{\hbar \Omega}{2} (\sigma^+ + \sigma^-)$$

$$\Rightarrow \hbar \delta / 2 \hat{\sigma}_z + \hbar \Omega S_x = \hbar \delta S_z + \hbar \Omega S_x$$

2) Dark time

$$\langle s^- \rangle = -\frac{c}{2} \sin \theta e^{-i\delta t} \rightarrow \text{phase}$$

$$|\langle s^- \rangle| = \text{contrast}$$

3) Detected by $\pi/2$ rotation, measure population

$$\langle s_y \rangle \Rightarrow \langle s_z \rangle = \frac{1}{2} \sin \theta \cos \delta t$$

\Rightarrow Optical Clocks: optical frequencies

Microwave clocks: GHz frequencies

1D lattice clock

Signal $\propto N$

Signal $\propto \sqrt{N}$
Noise

Noise $\propto \sqrt{N}$

SU(N) symmetry

Let's start with SU(2) e.g. Coulomb interactions between electrons

$V \propto \frac{1}{r}$ Independent of spin

However singlet and triplet states can have different energies due to quantum statistics

Imagine the case of contact interactions

$$|\psi\rangle = \frac{(|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle)}{\sqrt{2}} \underbrace{\psi(r_1, r_2)}_{\text{symmetric}}$$

$$|\psi\rangle = \frac{(|\uparrow\downarrow\rangle + |\downarrow\uparrow\rangle)}{\sqrt{2}} \underbrace{\psi(r_1, r_2)}_{\text{anti-symmetric}}$$

also $|\uparrow\uparrow\rangle$ and $|\downarrow\downarrow\rangle$

S-wave interactions are described by

$$H = -V(r_1, r_2) \times (\vec{S}_1 \cdot \vec{S}_2 - 1/4) \Rightarrow \text{only singlets interact via s-wave}$$

H is SU(2) symmetric

$$\text{commutes with } \hat{S}_x = \frac{1}{2} (\sigma_{x1} + \sigma_{x2})$$

$$\hat{S}_y = \frac{1}{2} (\sigma_{y1} + \sigma_{y2})$$

$$\hat{S}_z = \frac{1}{2} (\sigma_{z1} + \sigma_{z2})$$

3 generators of SU(2)

SU(N) symmetry ($N = 2J+1$)

$N=10$ for ^{87}Sr .

$$H_{\text{int}} = - \sum_{i,j} V(r_i, r_j) S_n^m(i) S_n^m(j)$$

$S_n^m(i) \equiv |n\rangle_i \langle m|$ ($N^2 - 1$) SU(N) generators.

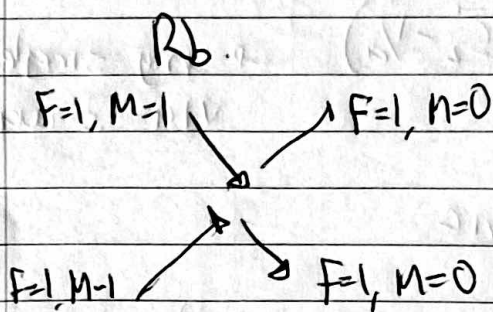
$$S_n^m = \sum_i S_n^m(i)$$

$$[H, S_n^m] = 0 \quad \forall n, m$$

→ No mode changing collisions

+ Equal scattering lengths

Not the case in alkali atoms



Rb. $a_{F=0}$ $a_{F=2}$
Two scattering lengths

This is forbidden
in alkaline-earth
atoms

All a are the same

$$\frac{\delta a_s}{a_s} \sim \Delta V \delta E$$

δE : time in short range potential $\sim 1 \text{ ps}$

ΔV : different potential experienced by nuclear spins

^{15}O

$$\Delta V \sim \frac{H_{\text{HF}}^2}{\Delta E} \approx \frac{(300 \text{ MHz})^2}{(400 \text{ THz})} \sim 200 \text{ Hz}$$

$$\frac{\delta a_s}{a_s} \sim 10^{-9}$$

$$^3\text{P}_0 - ^3\text{P}_0, \quad ^1\text{S}_0 - ^3\text{P}_0$$

$$\Delta V \sim H_{\text{HF}} \approx 300 \text{ MHz} \Rightarrow \frac{\delta a_s}{a_s} \sim 10^{-3}$$

first order.

Collisions $U(1) \times SU(N)$

\Downarrow

N_e conserved
 N_g conserved

up to ee losses