

Reaction-Diffusion Models in One Dimension

1. Overview of Problems & Techniques

Course overview:

* Problems & techniques

- ▽ Reaction/Diffusion - controlled limits
- ▽ Anomalous kinetics: why $d=1$?
- ▽ Fluctuations @ all length scales - trapping
 - $A+B \rightarrow \emptyset$
- ▽ Fluctuations in number space - Fisher front.
 - △ Effective reaction-order/rate
 - △ Scaling / Dimensional analysis
 - △ Reaction-diffusion equations

* One-species diffusion-limited coalescence, $A+A \rightarrow A$, and the Method of Empty Intervals

* $A+A \rightarrow \emptyset$ and other advanced applications of "empty intervals"

* Open challenges



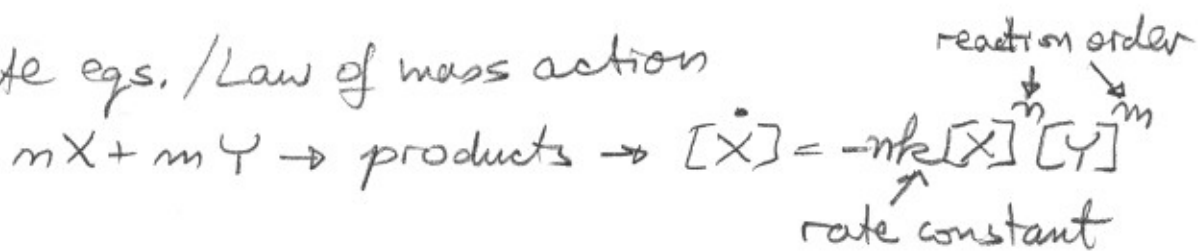
Diffusion time: t_D - typical time for "particles" to meet

Reaction time: t_R - typical time for particles to react, when held within reaction range from one another.

Reaction-Controlled: $t_R \gg t_D$

Diffusion-Controlled: $t_D \gg t_R$

→ Rate eqs. / Law of mass action



→ Fluctuation-dominated

[no coherent approach]

{ Anomalous kinetics (different from reaction-controlled)

Most anomalous for $d=1$
[upper critical dimension]

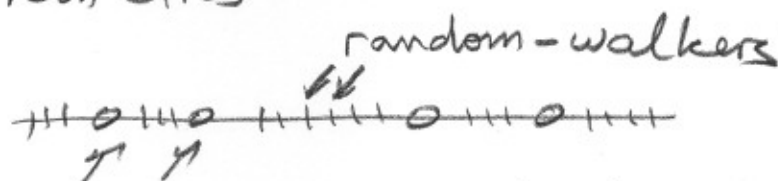
Role of fluctuations and examples of anomalous kinetics

(a) trapping

Donker-Varadhan

Abel prize, 2007

Large deviations



traps at concentration c

Reaction-limited: $S(t) \sim e^{-ct}$

Diffusion-limited:

$\left\{ \begin{array}{l} \text{Prob. of trap-free gap of size } L: ce^{-cL} \\ \text{survival in gap of size } L: e^{-Dt/L^2} \end{array} \right.$

exercise

$$S(t) = \int_0^\infty e^{-Dt/L^2} ce^{-cL} dL \sim e^{-t^{1/3}} \text{ (exercise)}$$

→ anomalous kinetics in all d :

$$S(t) \sim e^{-t^{d/(d+2)}} \text{ (most anomalous for } d=1)$$

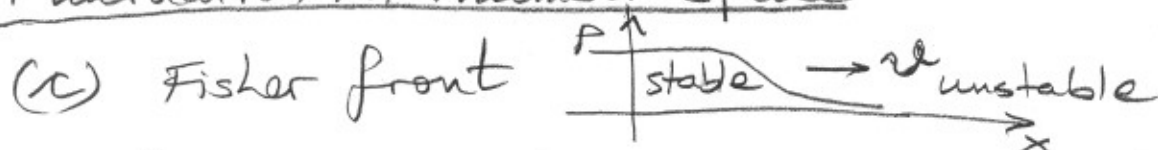
→ result of fluctuations in the size of trap-free regions



$$\rho \sim t^{-d/4}$$

$$d_x = 4 \quad (d=1 \text{ most anomalous})$$

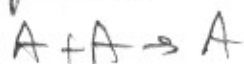
Fluctuation in number space



ρ inaccurate when one assumes ρ continuous (Also, with particles wave front has finite support).

* Exercise: radioactive decay $A \rightarrow \phi$
continuum vs. particle kinetics

Techniques



Reaction-limited
 $\dot{\rho} = -k\rho^2 \rightarrow \text{exercise}$

$$\boxed{\rho \sim \frac{1}{kt}}, \quad kt \gg \rho_0^{-1}$$

Diffusion-limited

Dimensional analysis:

$$[D] \approx \frac{L^2}{T}, \quad [t] = T \Rightarrow \rho \propto (Dt)^{-d/2}$$

$\uparrow$
 $[p] = \frac{1}{L^d}$

* But notice assumptions !!

Effective rate eqs.

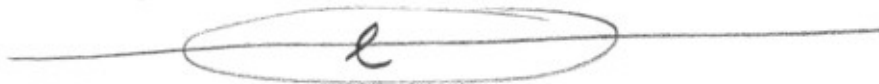
$$\dot{\rho} = -\rho^3 \xrightarrow{\text{effective order}} \rho \sim 1/\sqrt{t} \quad (\text{for } d=1)$$

or $\dot{\rho} = -\left(\frac{1}{\sqrt{t}}\right)\rho^2 \rightarrow \rho \sim 1/\sqrt{t}$

$\uparrow$
effective rate

ad-hoc

Scaling



$$t(l) \sim l^2/D$$

After $t(l)$ only one A left $\Rightarrow$

$$\rho = \frac{1}{l} \sim \frac{1}{\sqrt{Dt}}$$

[Also $A+B \rightarrow \emptyset$]

Reaction-Diffusion eqs.

$$\rho \rightarrow \rho(\vec{x}, t)$$

$$\frac{\partial}{\partial t} \rho = \underbrace{R\{\rho(\vec{x}, t)\}}_{\text{mean-field}} + \underbrace{D \Delta \rho}_{\text{diffusion}}$$

Often fails (can't manage both
a cell large enough for ρ continuous
yet small enough for MF reaction rate)
