

Granular Matter II

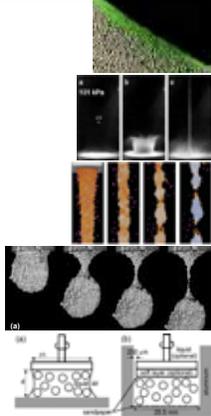
Heinrich Jaeger
University of Chicago

<http://jfi.uchicago.edu/~jaeger/group/index.html>



Outline

- Brief intro to granular matter
- Granular jets & freely falling streams
- Dense suspension droplet break-up
- Shear thickening in suspensions



Break-up = catastrophic deformation, topological transition

In pure simple liquids: distinct, smooth structures during break-up; signature of liquid type (viscous, inviscid, etc.)

Self-similarity and scaling near break-up at time $t = t_0$

Ingredients: surface tension, viscosity, density

Cut-off: molecular scale

- Model system to test scaling ideas about topological transitions (fluids, condensed matter physics, materials science, ..., black holes)
- Technological relevance (spraying, fuel injection, ...)



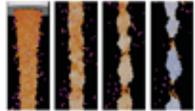
Break-up in dry granular systems (jets, streams):
distinct, much more ragged structures in neck region; signature of granularity

Scaling near break-up at time $t = t_0$; but: where does 2/3 power law come from?

Ingredients = ? (material density, eff. surface tension due to cohesion, coeff. of restitution, plus??)

Cut-off: grain scale

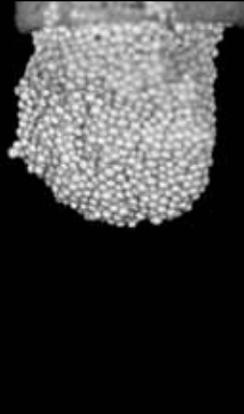
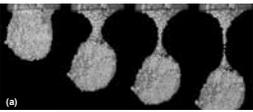
- Model system to test for tiny (nm, nN) cohesive forces between macroscopic grains



Suspension Break-Up

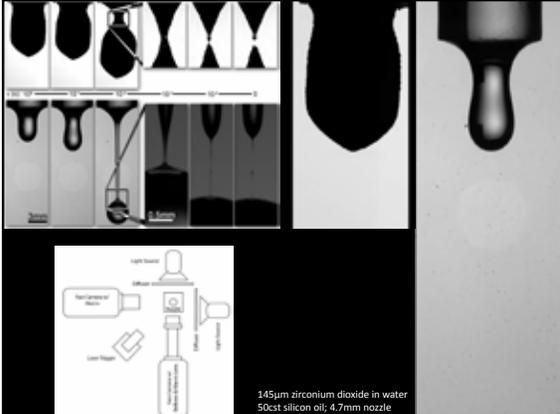
Large particles, $> 10\mu\text{m}$
Non-Brownian

Cut off bottom

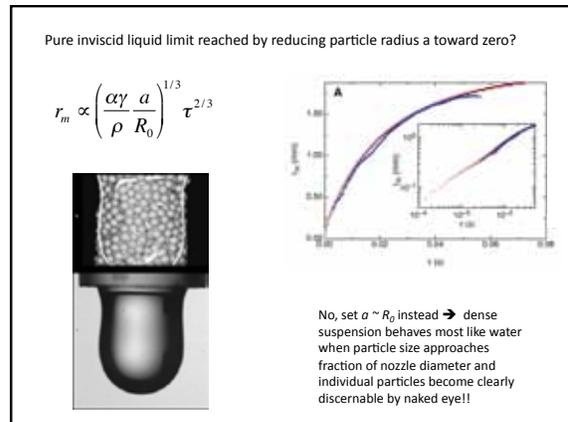
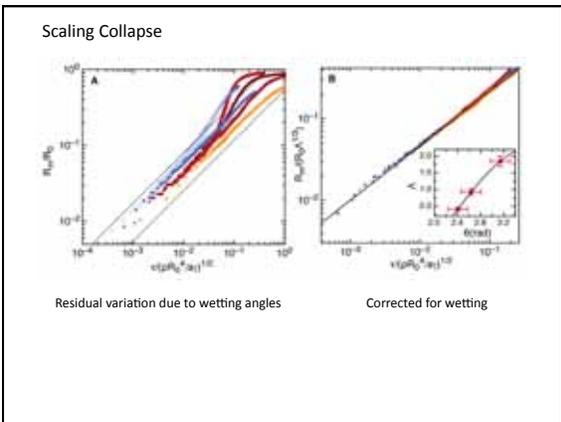
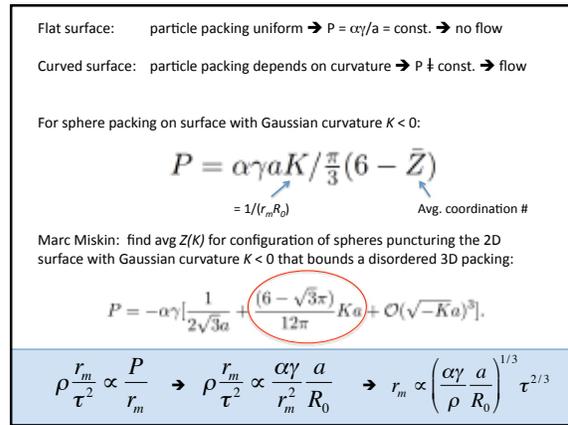
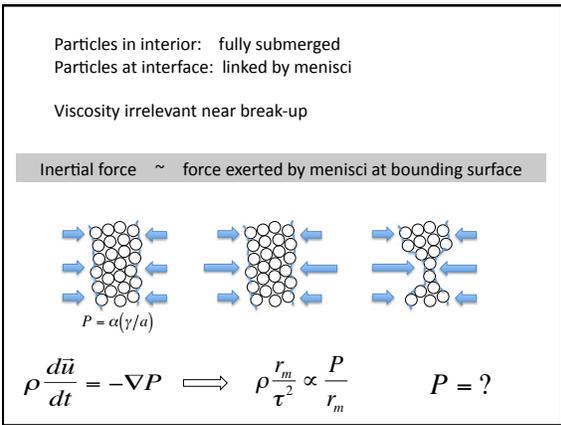
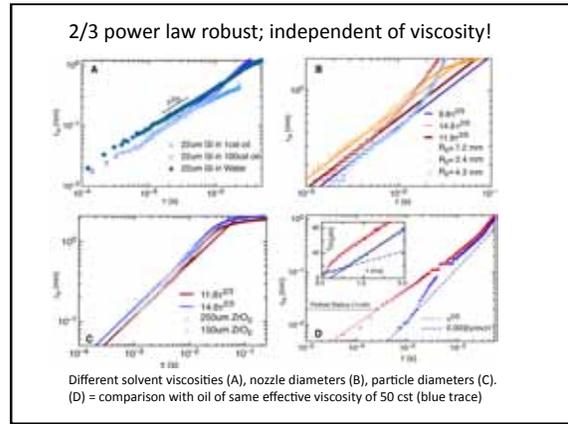
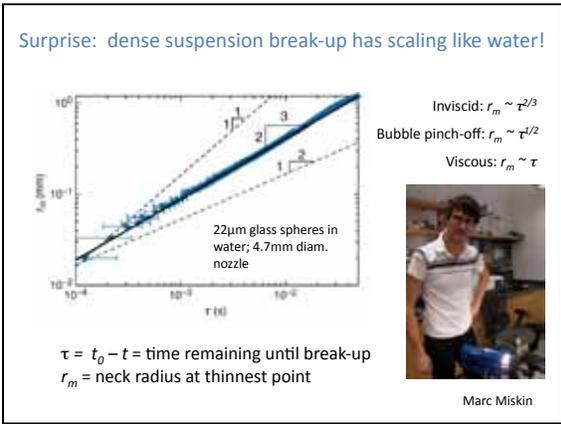




(a)

850 μm zirconium dioxide in water



145 μm zirconium dioxide in water
50cst silicon oil; 4.7mm nozzle



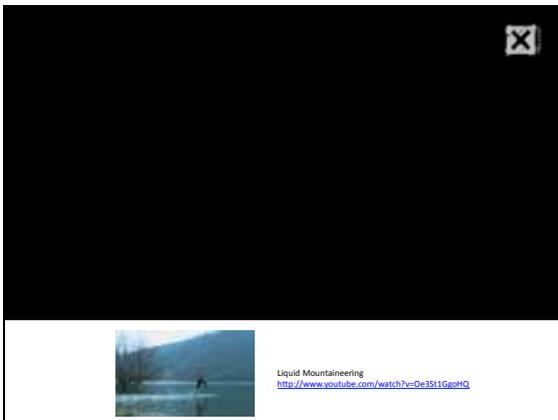
Dense Suspension Break-Up

- New class of break-up phenomena, intermediate between simple liquids and dry grains
- Interplay between break-up dynamics and particle arrangement on curved bounding surface
- Scaling but no self-similar structure; instead memory of initial conditions (nozzle size)
- Viscosity unimportant → dense suspension not simply a very viscous liquid as often assumed
- Beyond single particle scale: break-up proceeds like pure liquid

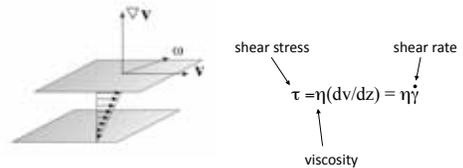
When and how to walk on water: A new perspective on shear thickening in suspensions

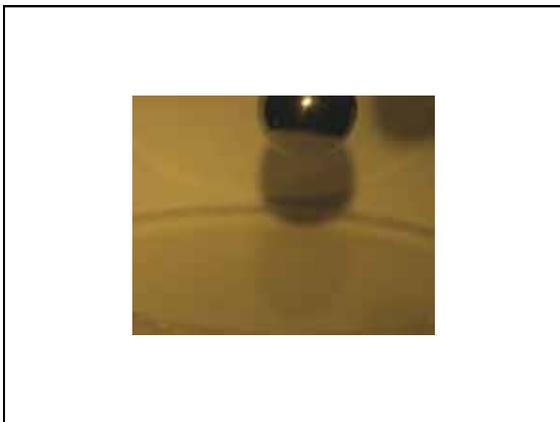
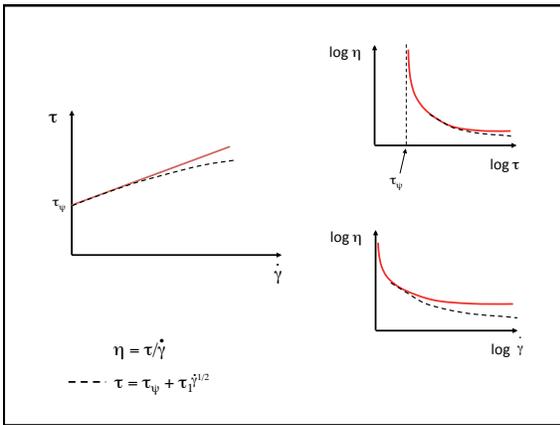
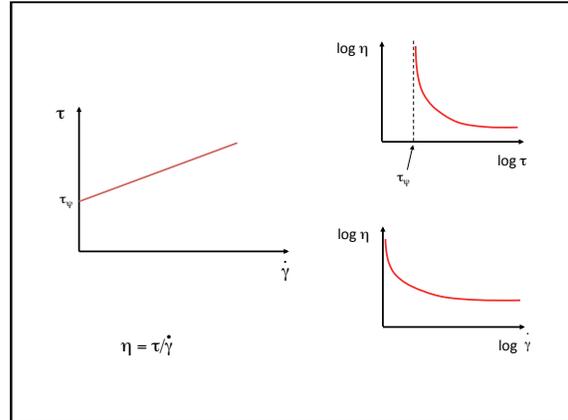
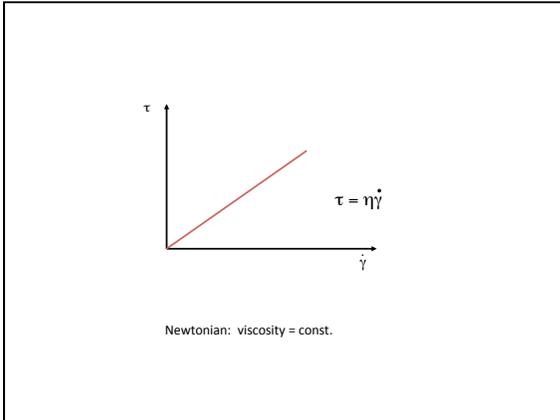


National Geographic
<http://www.youtube.com/watch?v=d5yabrnzxxk&feature=fvw>
 BBC 1
<http://www.youtube.com/watch?v=AC6C0wvW08&feature=related>



Liquid Mountain climbing
<http://www.youtube.com/watch?v=De3t1GqoH0>



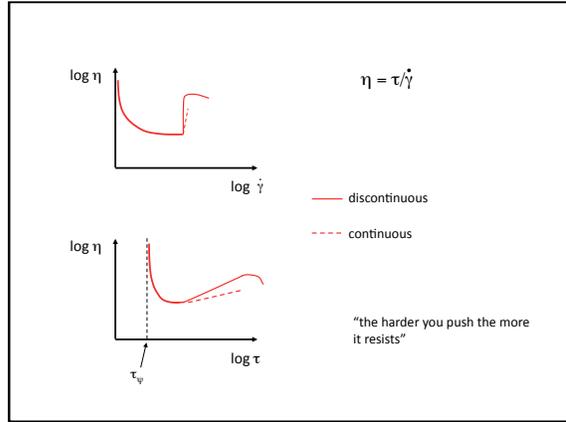
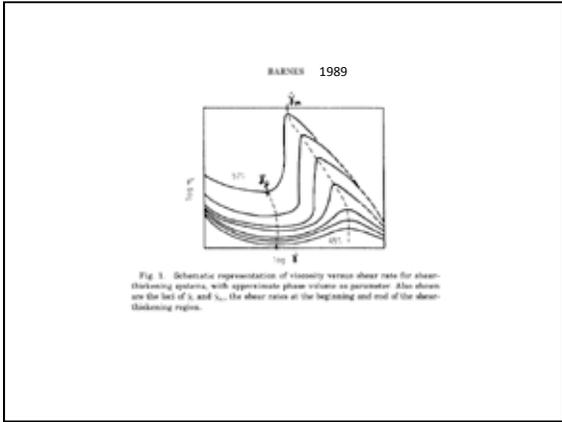


Norman J. Wagner & John F. Brady,
Physics Today 2009

Shear thickening in colloidal dispersions

“Shampoos, paints, cements, and soft body armor that stiffens under impact are just a few of the materials whose rheology is due to the change in viscosity that occurs when colloidal fluids experience shear stress.”

Viscosity of colloidal latex dispersions, as a function of applied shear stress.

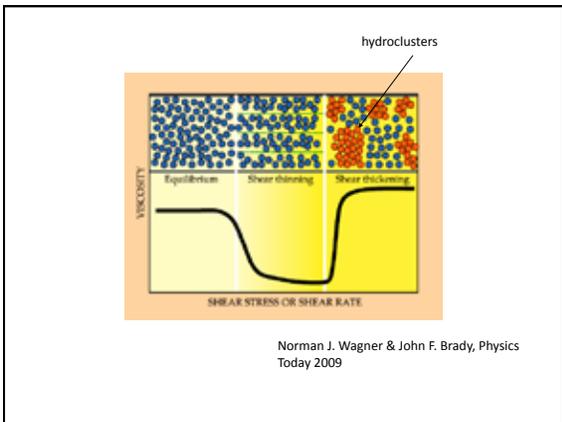


Shear-Thickening (“Dilatancy”) in Suspensions of Nonaggregating Solid Particles Dispersed in Newtonian Liquids

H. A. BARNES, *Unilever Research, Port Sunlight Laboratory, Bebington, Wirral, Merseyside, England L63 3JW*

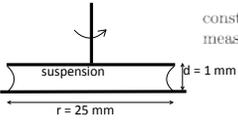
We shall find that so many kinds of suspensions show shear thickening that one is soon forced to the conclusion that given the right circumstances, *all* suspensions of solid particles will show the phenomenon.

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Journal of Rheology, 11(2), 329–366 (1969) CCC 0148-6075/69/020329-38\$04.00



- 
- Eric Brown
- What determines the range of observable shear thickening?
OR: Why don't all suspensions shear thicken?
 - What is the mechanism for discontinuous shear thickening?
- See <http://arxiv.org/abs/1010.4921>

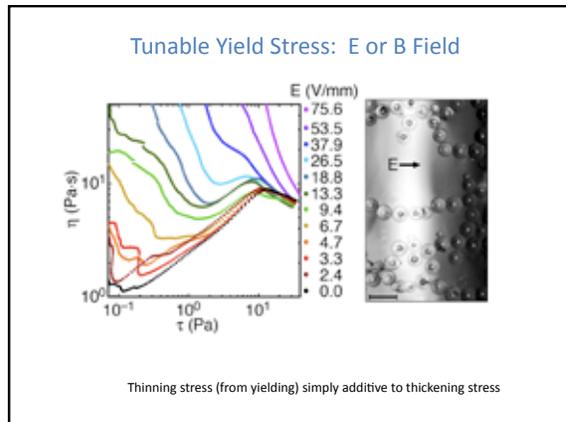
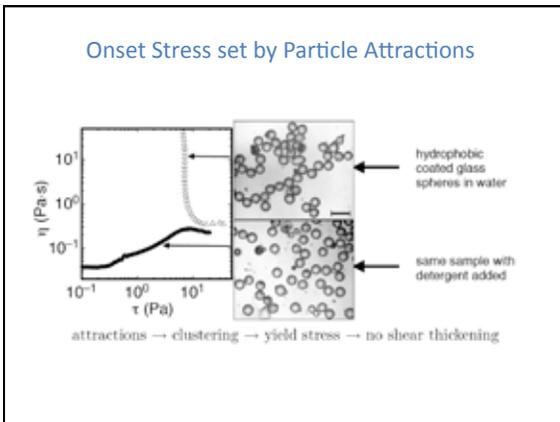
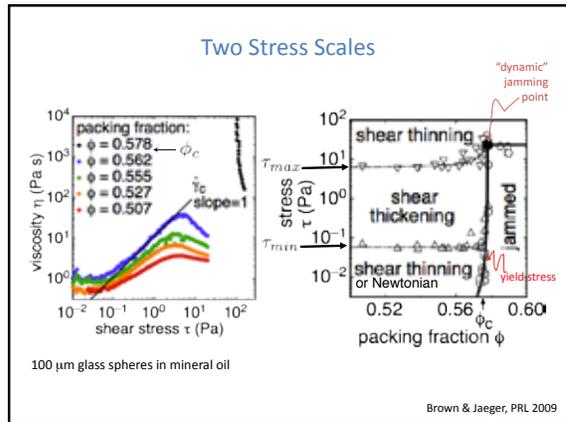
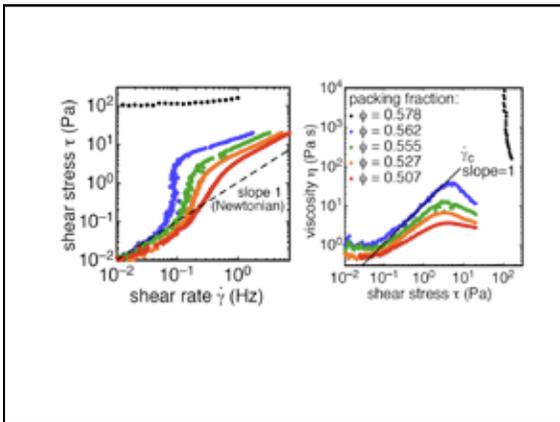
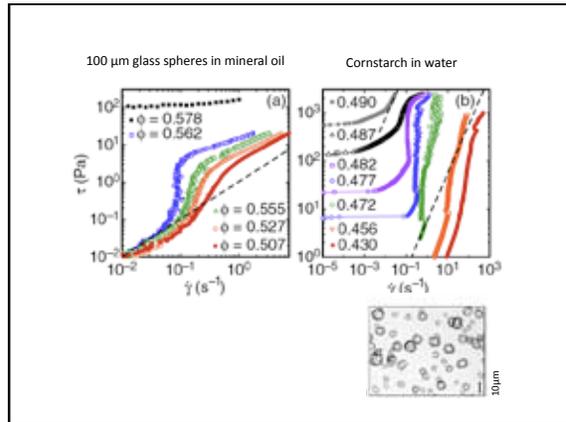
Rheometry

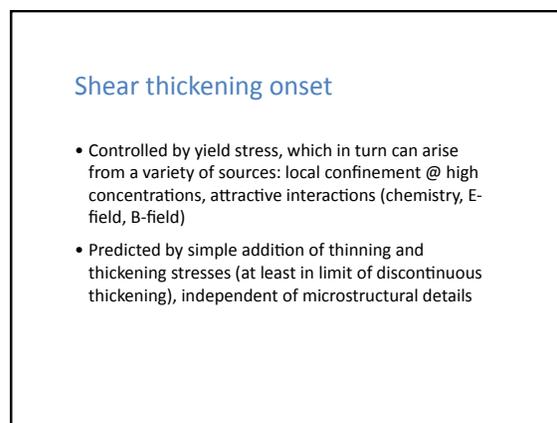
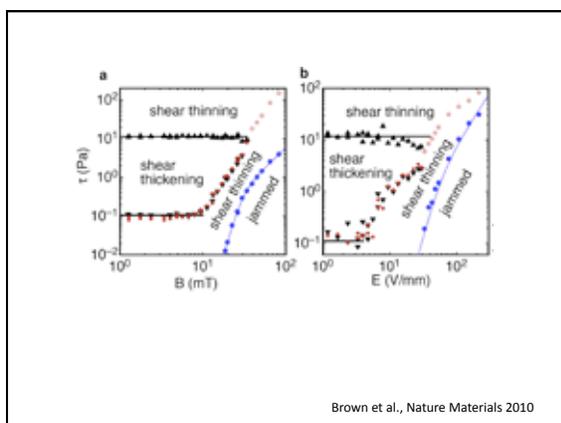
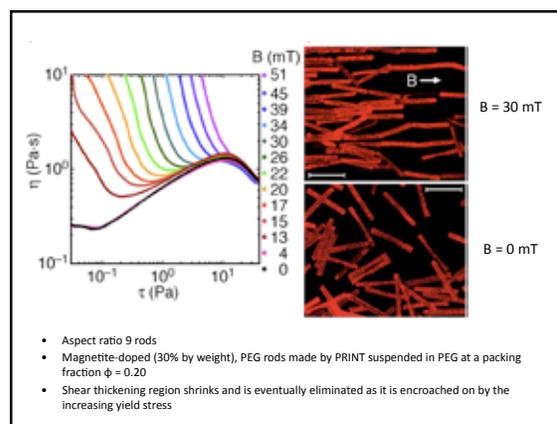
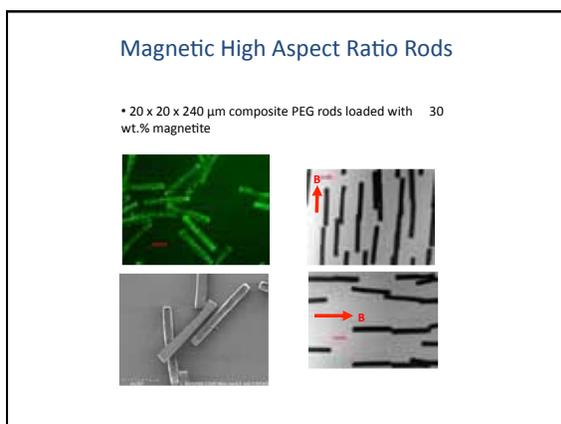
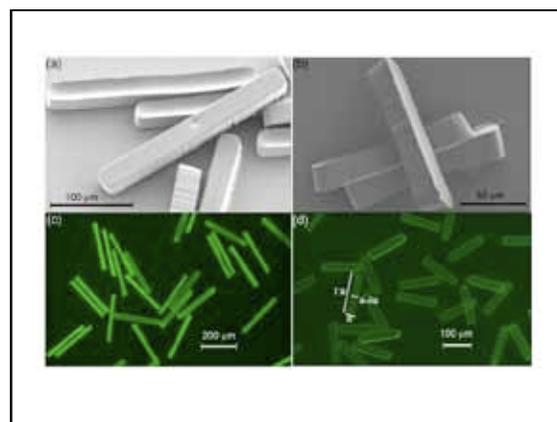
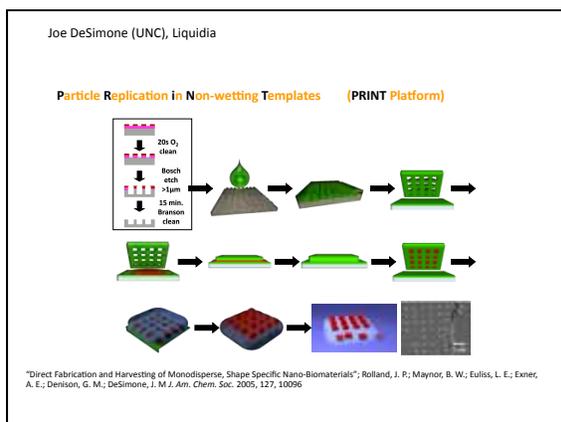


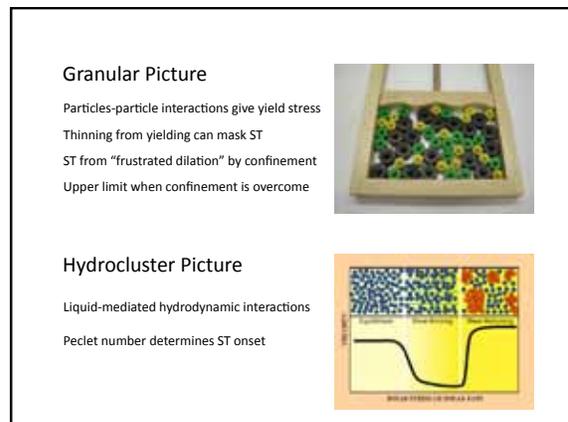
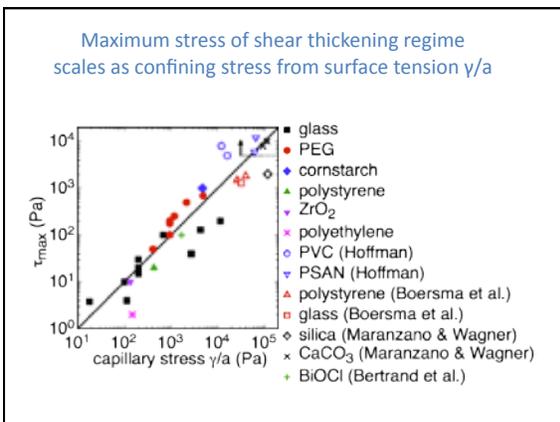
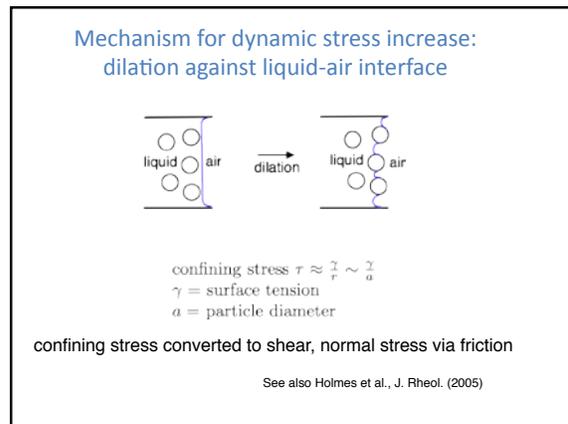
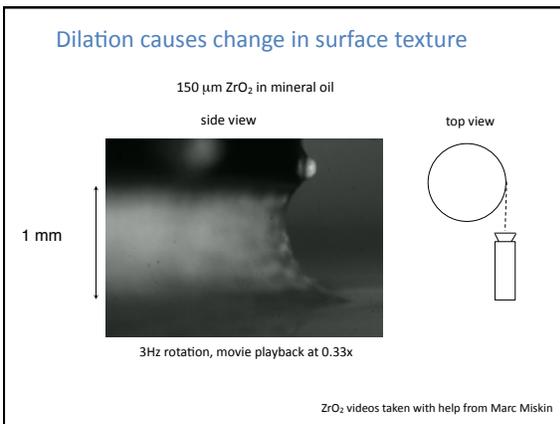
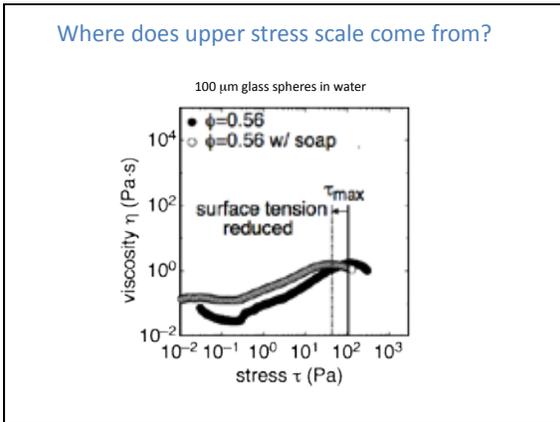
constant applied torque T
measured rotation rate ω

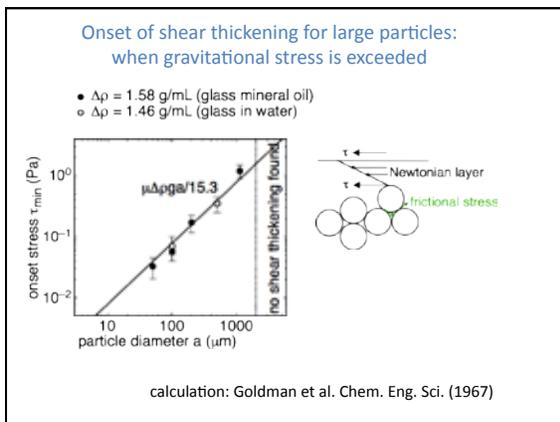
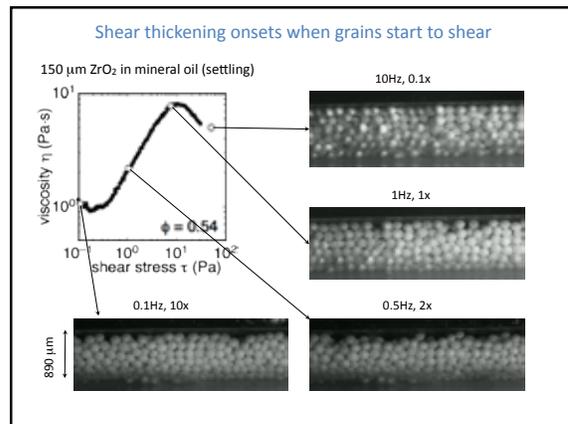
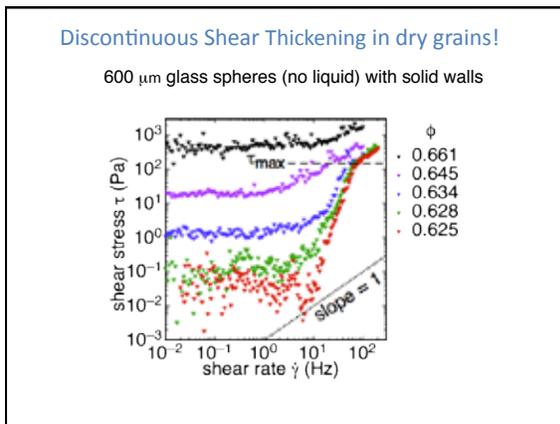
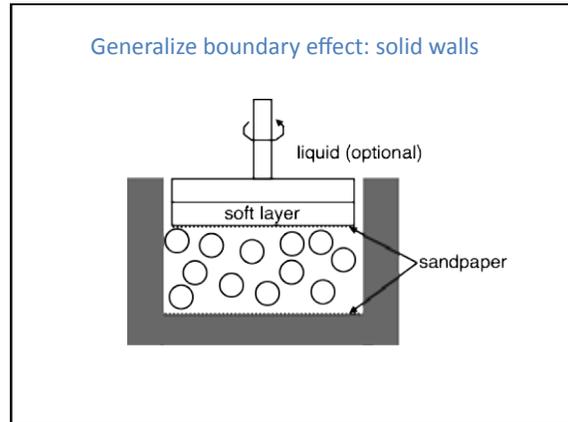
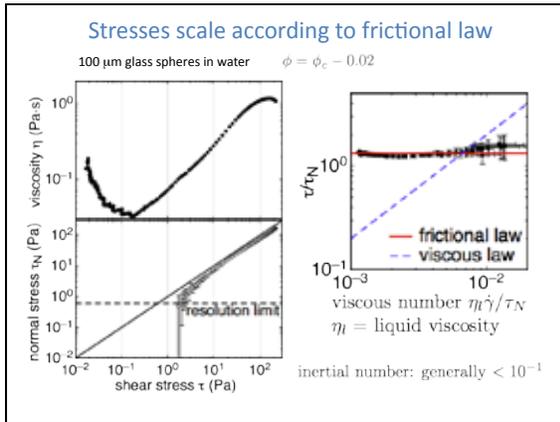
shear stress $\tau = T/\pi r^3$
shear rate $\dot{\gamma} = \omega r/d$
viscosity $\eta = \tau/\dot{\gamma}$ in steady state

suspensions:
 6 μm - 1000 μm glass spheres in various liquids
 14 μm cornstarch in water w/ CsCl (density matched)
 125 μm polyethylene spheres in silicone oil (density matched)
 150 μm ZrO₂ spheres in mineral oil
 PEG (various shapes) in liquid PEG









When to walk on water...

Use suspension with particle size that gives largest shear thickening range

Smaller size gives better confinement γ/a as well as lower onset for particle diam. $> 10\mu\text{m}$

Use colloids?

Shear thickening starts when shear stress exceeds stress scale of particle interactions

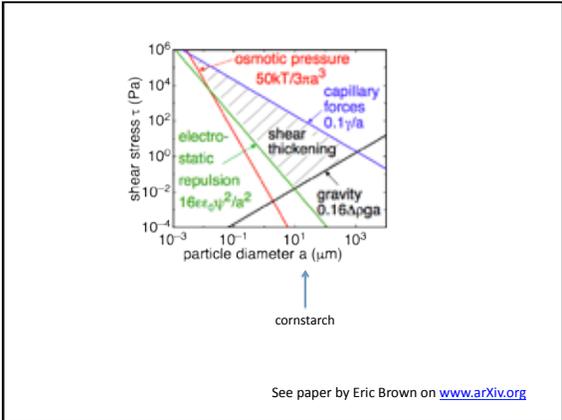
- Brownian colloids (osmotic pressure):
(Farr et al. 1997, Bergenholtz et al. 2002, Maranzano & Wagner 2002)

$$\tau_{min} = \frac{50kT}{3\pi a^3}$$

- Charge-stabilized colloids (electrostatic potential ψ):
(Hoffman 1982, Maranzano & Wagner 2001)

$$\tau_{min} = \frac{16\epsilon\epsilon_0\psi^2}{a^2}$$

- Induced dipole attractions from applied electric field E
(Brown et al. 2010)

$$\tau_{min} = 12\pi\epsilon_0\epsilon\beta^2 E^2$$


How and when to walk on water...




How: fast ... to generate large stresses

When: after adding 40+ percent of ~10μm particles

To sum up shear thickening:

- Conditions for Discontinuous Shear Thickening:
 - dilation under shear
 - confining stress in response to dilation
 - dominance of confining stress
 - particle interactions small so that yield stress well below upper limit due to confinement
- Granular picture can explain phase diagram & many of the details (frictional law for stresses, dilation, shear banding,...)
- Outstanding issues: Transient response? Compressive vs. shear vs. tensile stresses?

