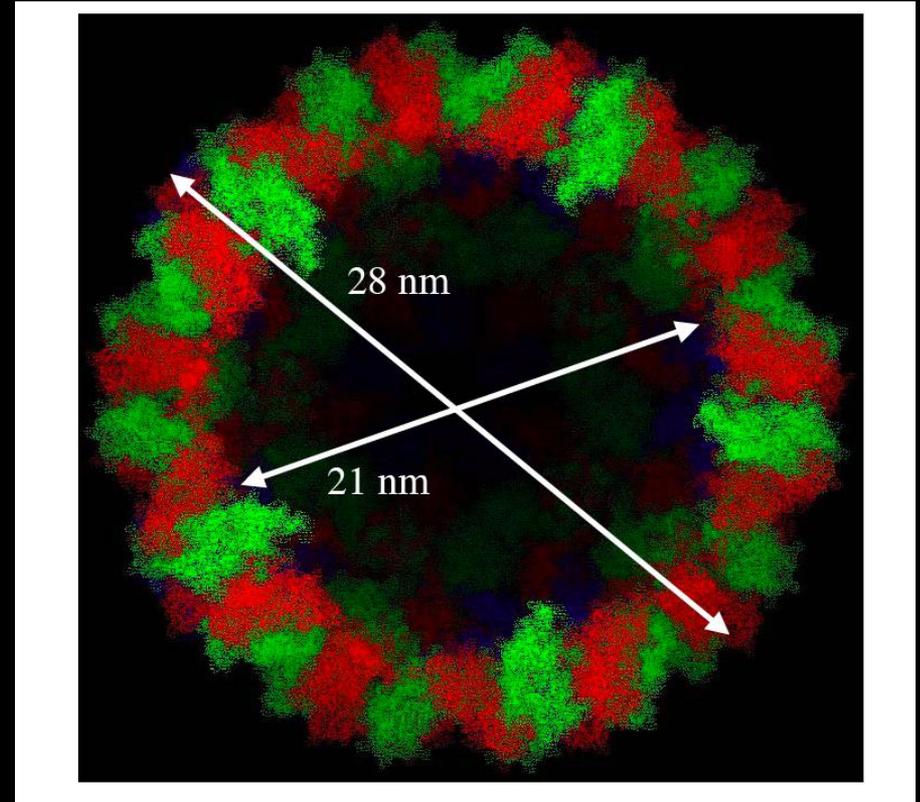
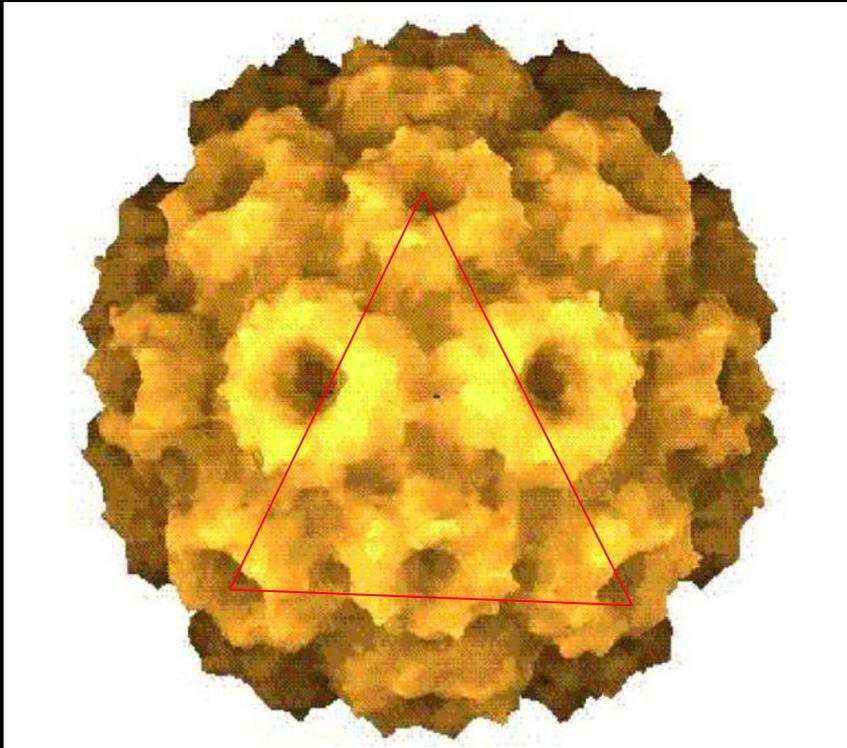


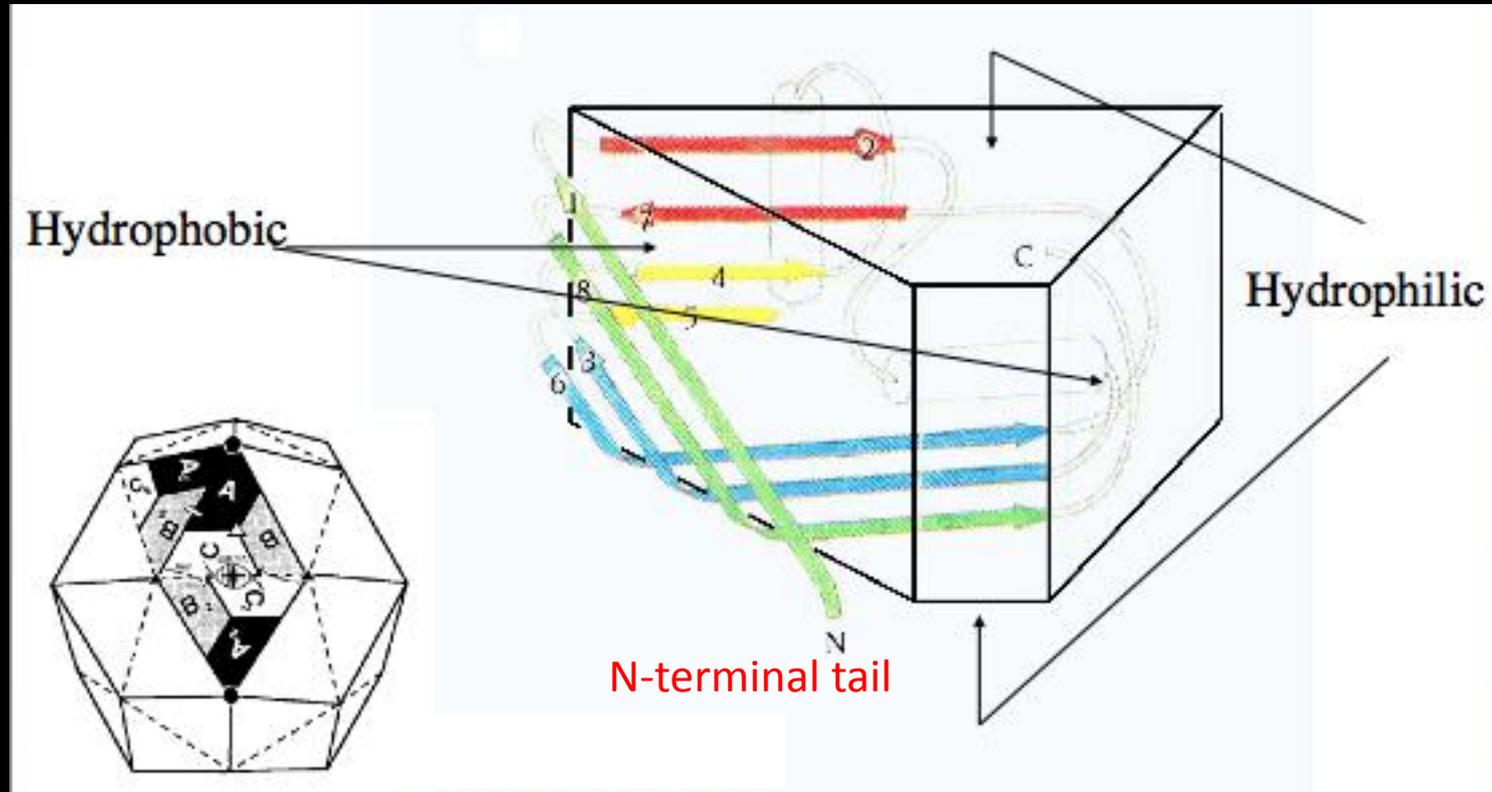
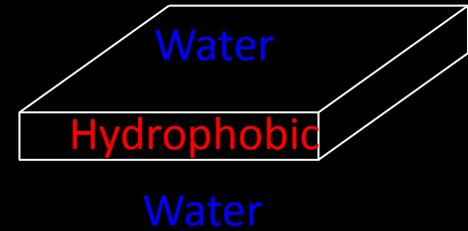
## II ) Fundamental Interactions.

Cowpea Chlorotic Mottle Virus: CCMV



- Attractive Interactions

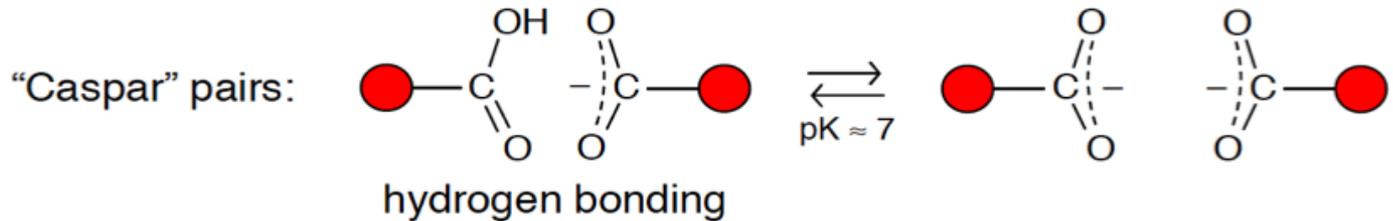
# A) Capsid Proteins: Amphiphilic



- Layer: *Spontaneous Curvature*
- Expect reversible, thermodynamic assembly

B)

- pH sensitive “Caspar-pairs”.

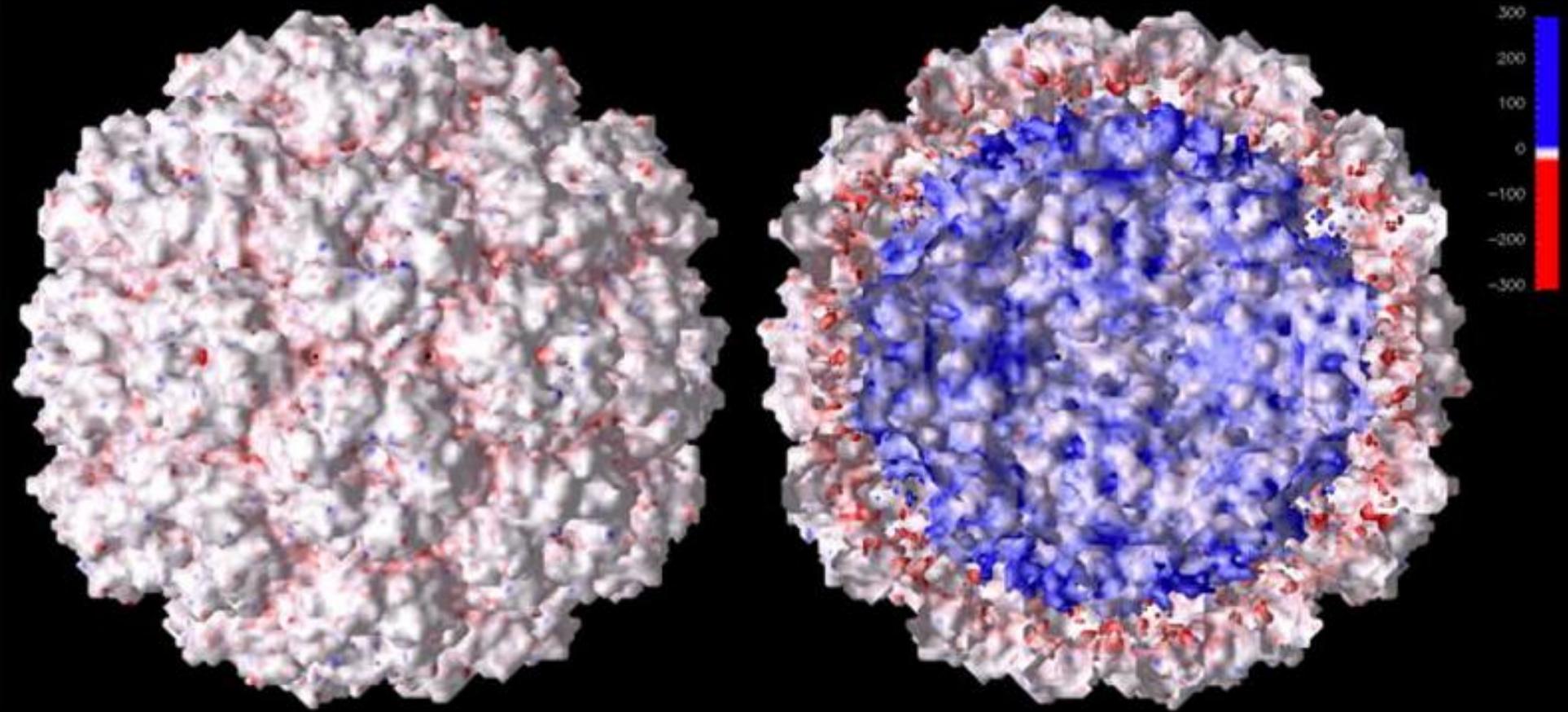


Kegel & van der Schoot,  
Biophys J., 2006

- Strength attractive interactions increases with acidity.

Repulsive

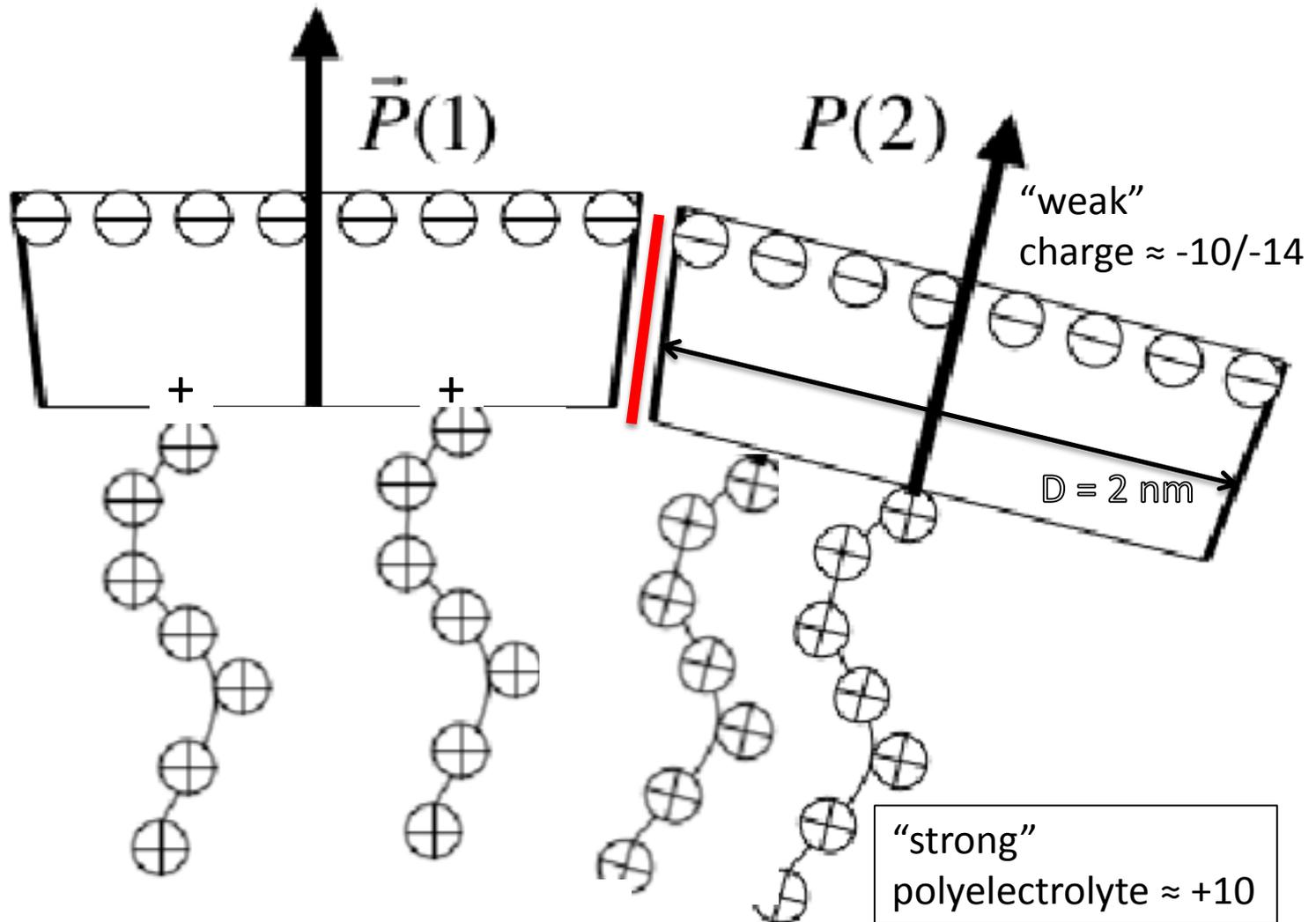
## C) Electrostatic Interactions



- Water-accessible equipotential surfaces. Blue positive; Red negative.
- *Inside-Outside Voltage Difference*

(McCammon et al.)

# Electrical Charges CCMV Dimers



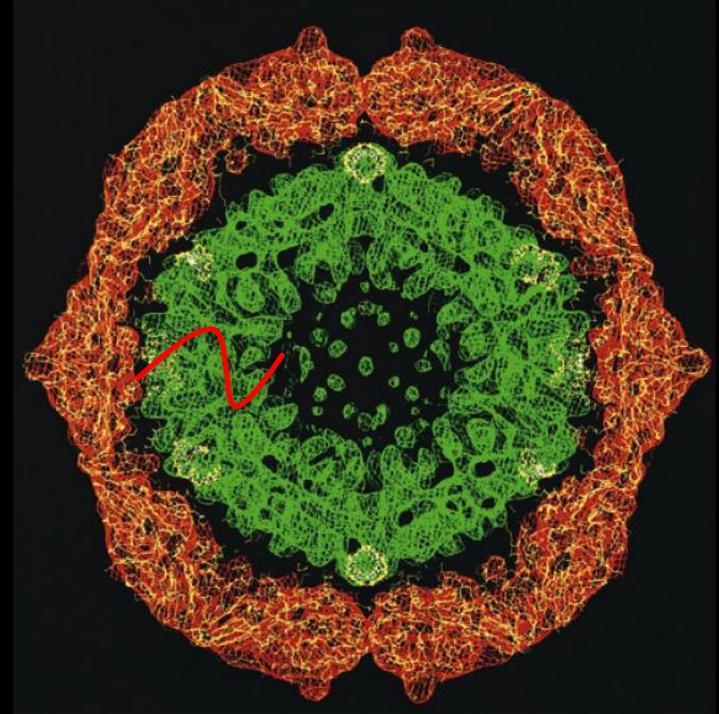
# Some “just so” questions about CCMV electrostatics

RNA has a total negative charge  $\approx -3,000$

Positive tail charge  $\approx 90 \times 20 = +1,800$

- If tail/RNA neutralization promotes viral assembly, *why neutralize only a fraction of RNA charge?*

Outer layer charge  $\approx -28 \times 90 = -2,520$



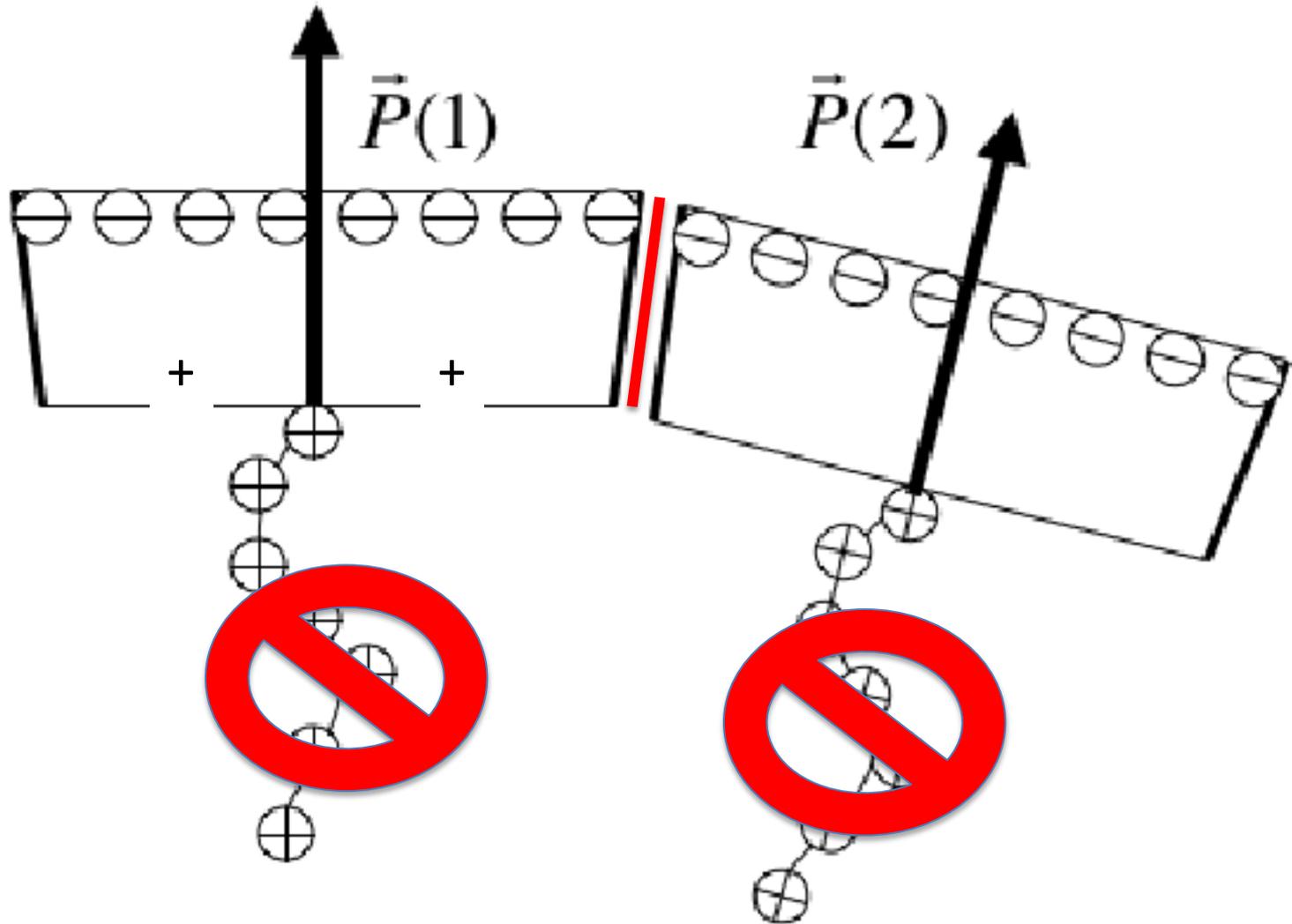
2) *What's the role of the large negative protein charge?*

- Prevents aggregation of viruses.
- Prevents RNA from sticking to capsids.

III)

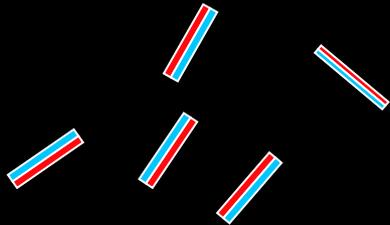
# Self-Assembly Empty Capsids

Electrostatic Repulsion vs. Hydrophobic Attraction

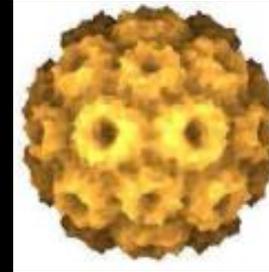


- Treat viral assembly as a *chemical reaction*:

90 Free CP Dimers (“subunits”)



Assembled T=3 capsid



Thermal Equilibrium

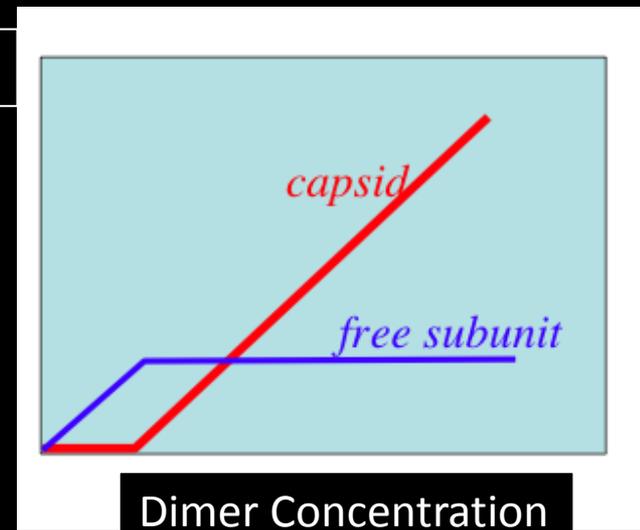
$$dG = \sum_{\text{components } i} m_i dN_i = 0$$

“Law of Mass Action”

$$K_{\text{capsid}} = \frac{[\text{T=3 capsid}]}{[\text{CP dimers}]^{90}}$$

$$K_{\text{capsid}} \propto \exp(90\Delta G / k_B T)$$

Concentrations



Dimer Concentration

- $\Delta G$  = assembly energy/dimer

“Signature” of Thermodynamic Self-Assembly

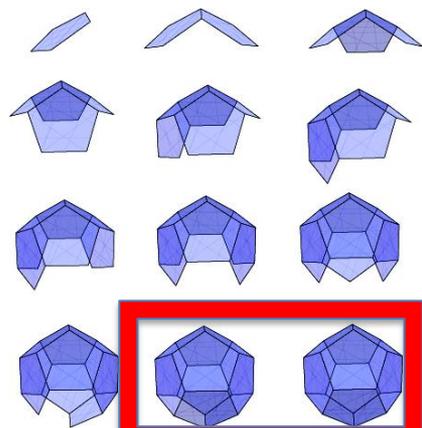
# Empty capsid assembly experiments

\* Acidic environment (low pH)

$$dG = 0$$

$$K_{\text{capsid}} = \frac{[\text{T=3 capsid}]}{[\text{CP dimers}]^{90}}$$

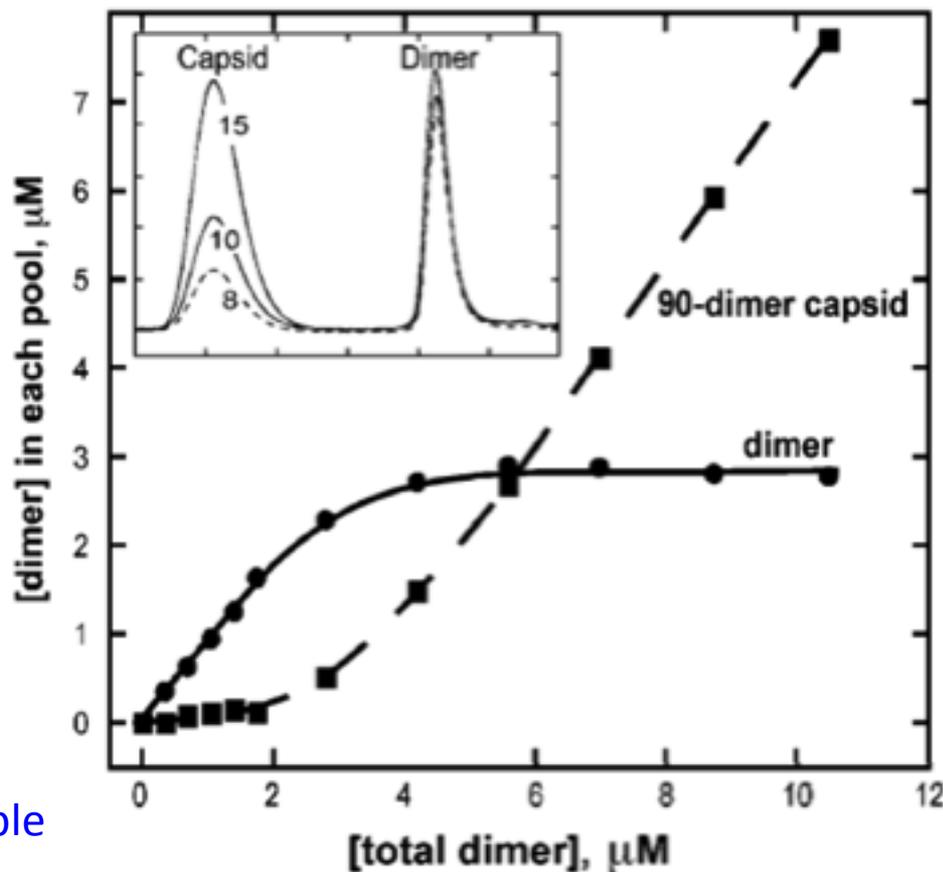
- $\Delta G \approx 30 k_B T/\text{dimer}$ .
- Capsid assembly is *irreversible!*?



reversible

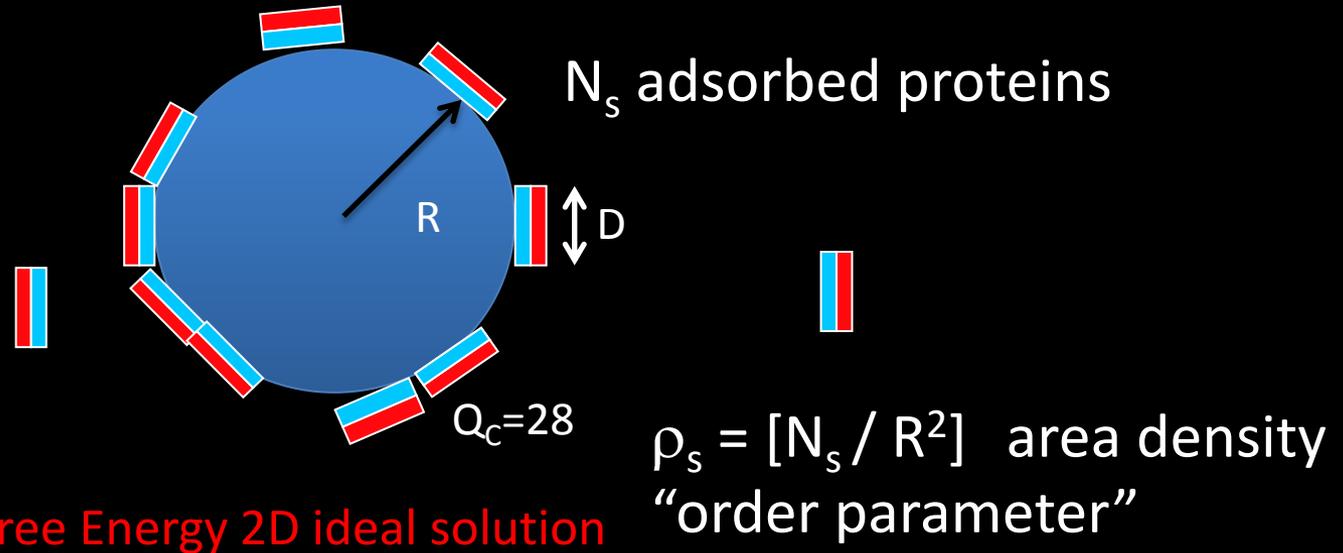
irreversible

## Chromatography



Zlotnick (2005) Nano Letters **5**, 765-70

# Capsid Van der Waals/Landau Free Energy



$$F_S(N_s, R) / N_s \gg k_B T \ln r_s D^2 - e + v_s r_s + w_s r_s^2 + \dots$$

$\epsilon$ : Adsorption energy proteins on sphere.

$V_s$ : Second "virial coefficient"

$W_s$ : Third virial coefficient  $\approx k_B T D^4$

Thermal equilibrium:

$$\frac{\mu F}{\mu N_s} = m \quad \text{Chemical potential proteins}$$

# Second Virial Coefficient

- Electrostatics vs Hydrophobicity

$$v_S = v_{DH} - J$$

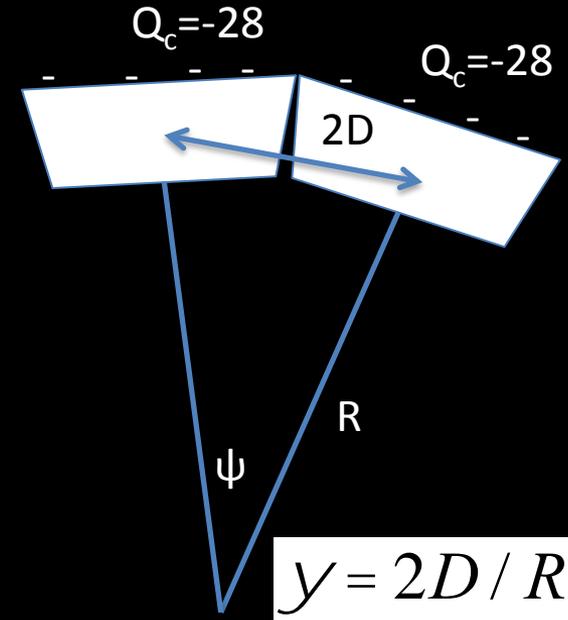
“Bjerrum Length”  $\approx$  nm

$$v_S(y) / k_B T = \frac{l_B Q_C^2}{k} - J_S(y)$$

“Debye Parameter”  
 $\approx$  1/nm

Angle-dependent  
hydrophobic attraction

Capsid Proteins



Optimal Angle/ Radius

$$y_c = 2D / R_c$$

CCMV (pH=5):  $Q_C = 20$

$$v_{DH} / k_B T = 400 \text{ nm}^2$$

$$v_S(y = y_c) / k_B T \gg -70 \text{ nm}^2$$

$$J_S(y = y_c) / k_B T \gg -470 \text{ nm}^2$$

Measured for empty shells

CP-CP Hydrophobic Attraction

Capsid Radius

# Debye-Hückel Theory of Aqueous Electrostatics

Macro ion Charge Density (CPs/RNA)

$$\Delta\phi - \kappa^2\phi = -\frac{4\pi e\rho_M(\vec{r})}{\epsilon}$$

Dielectric Constant Water

Electrical Potential

Debye parameter

$$\kappa^2 = 4\rho e^2 [\text{Salt}] / \epsilon k_B T$$

Electrostatic Free Energy

Bjerrum length

$$\frac{e^2}{\epsilon l_B} = k_B T$$

Sheet of charges

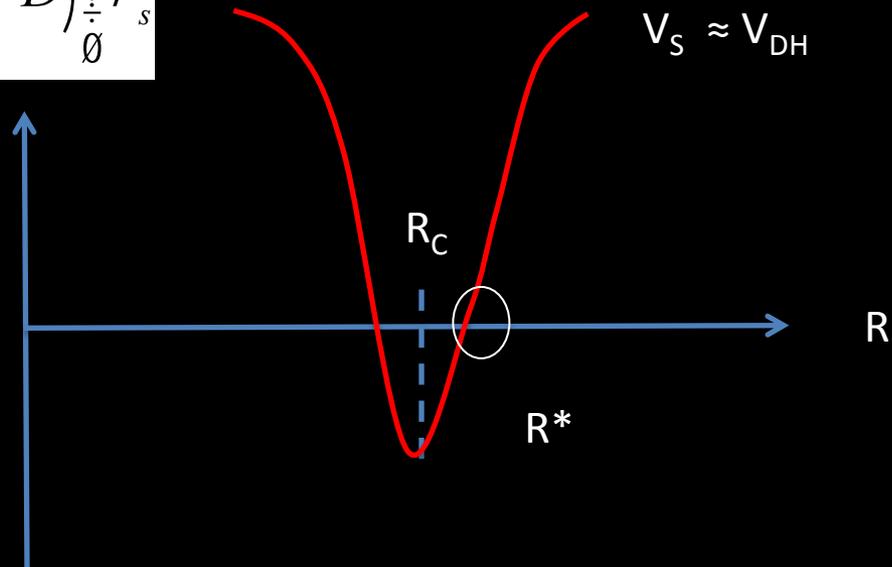
$$F_{DH} = \int d^3r \left( \frac{1}{2} e\rho_M(\vec{r})\phi + \frac{\epsilon}{8\pi} \kappa^2 \phi^2 \right)$$

$$F_{DH} / k_B T = \frac{\epsilon l_B Q_C^2 \ddot{O} N_S^2}{\epsilon k \ddot{\emptyset} R^2}$$

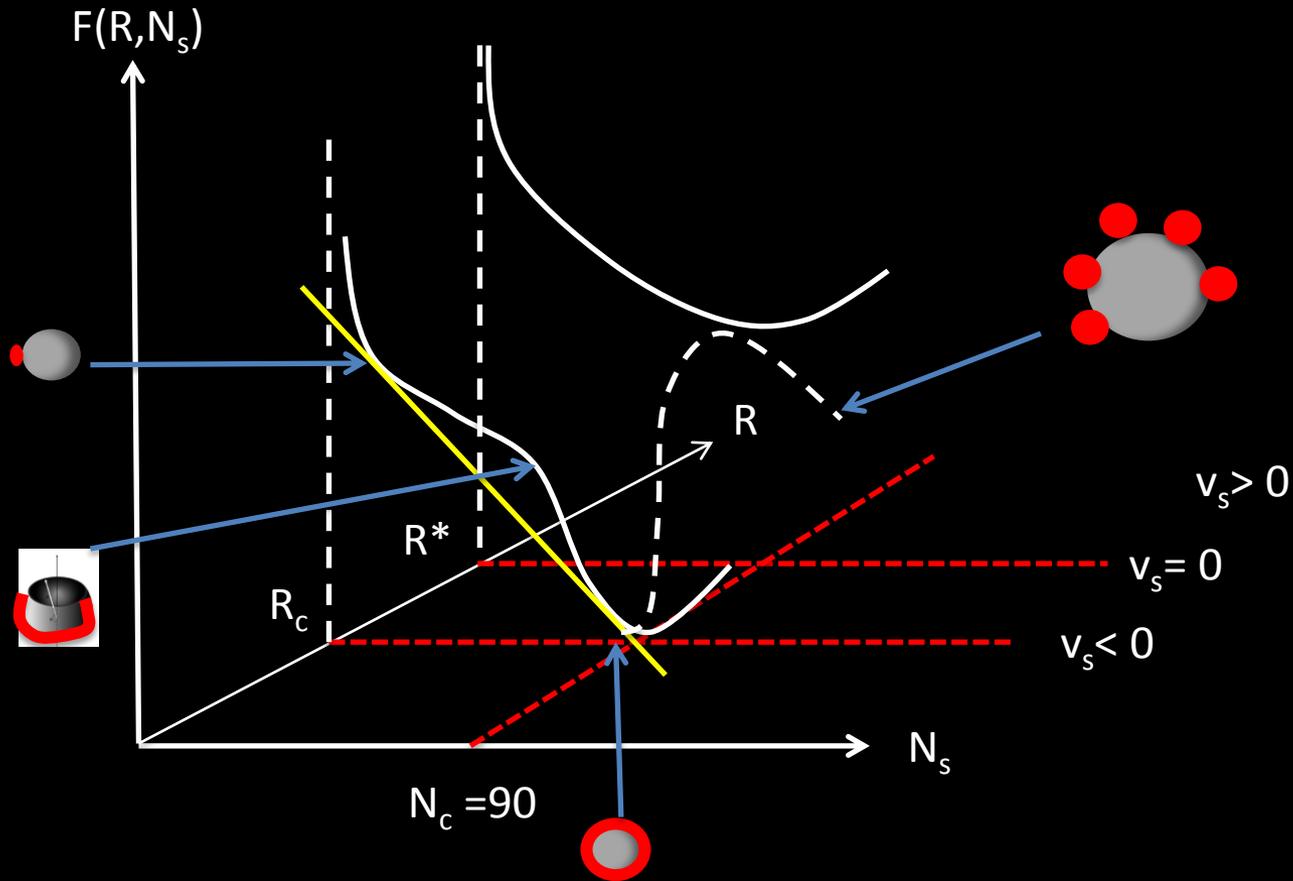
# Summary

- *Delicate balance between large repulsive interactions and large attractive interactions*
- *Second virial coefficient depends on the sphere radius  $R$ .*

$$F_2 / N_s = k_B T \left[ \frac{2}{3} \frac{l_B Q_C^2}{k} - J_S \left( R_c / D \right) \right] r_s^3$$



# Free energy "landscape" $F(R, N_s)$



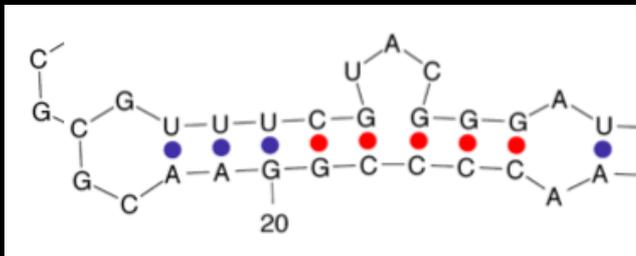
$$\frac{\nabla F}{\nabla N_s} = m$$

"Common-tangent construction"

Phase-coexistence: nearly closed shells and nearly bare spheres

# IV) Condensation RNA genome molecules

- Highly branched, highly charged “polyelectrolyte”

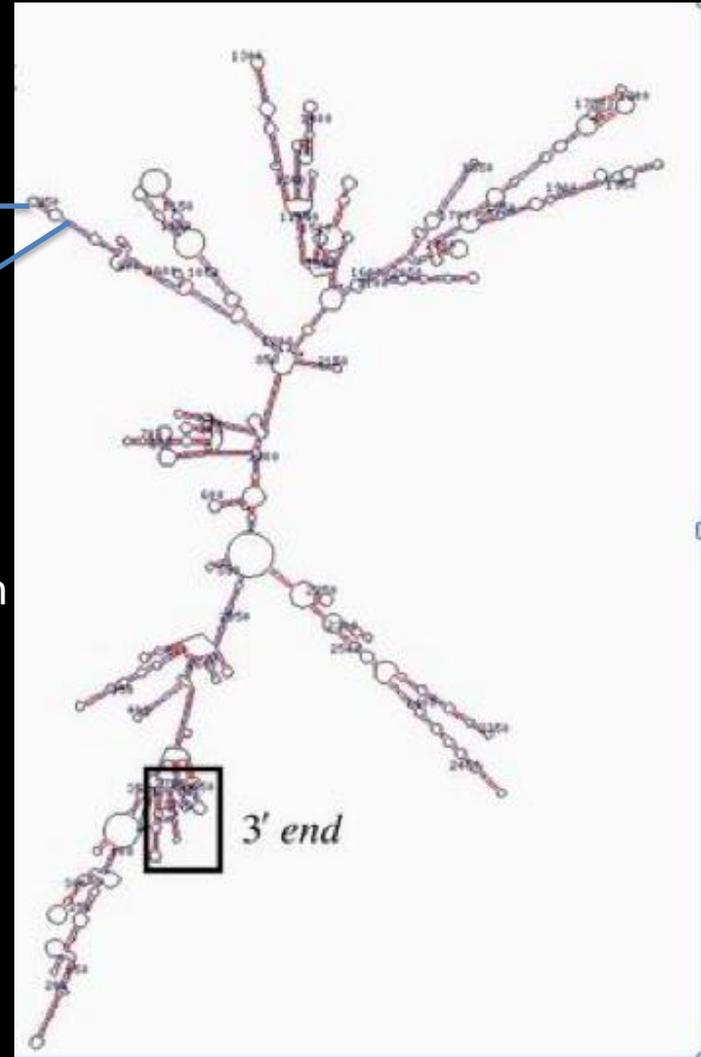


Paired stretches  $l \approx 5$  bp

Neutron scattering  $R \approx 11$  nm  
(no “condensing agents”)

- Highly compactified

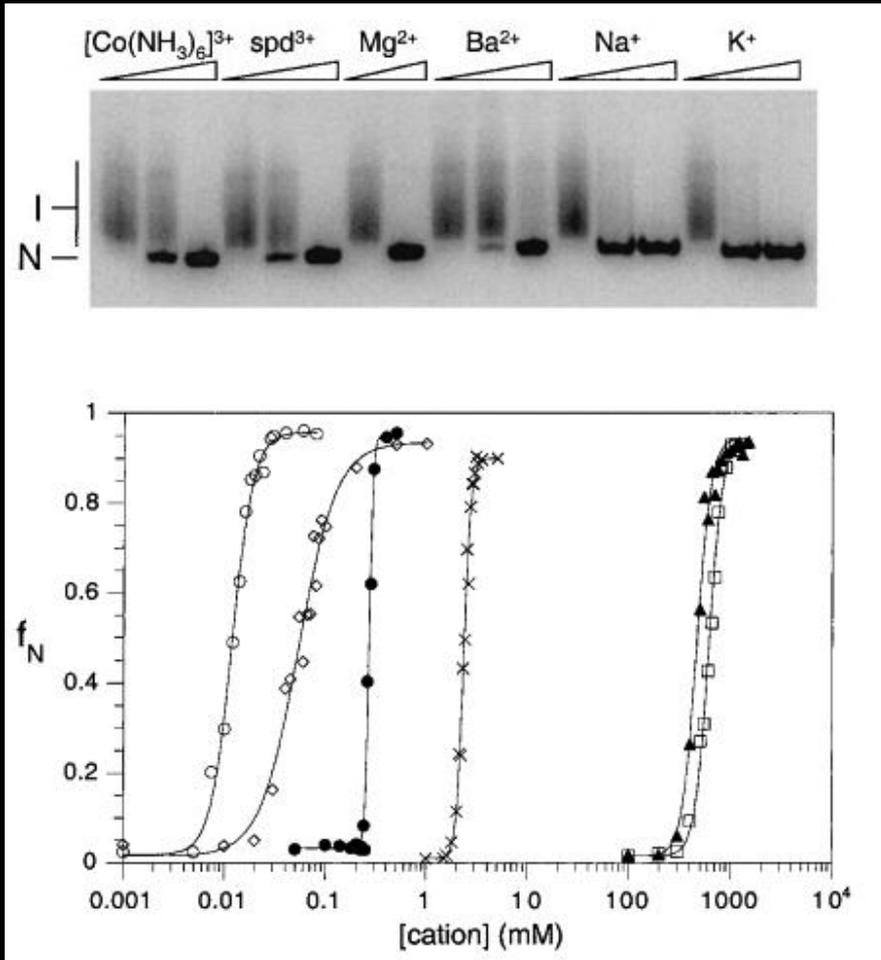
$\approx 100$  nm



CCMV RNA 1

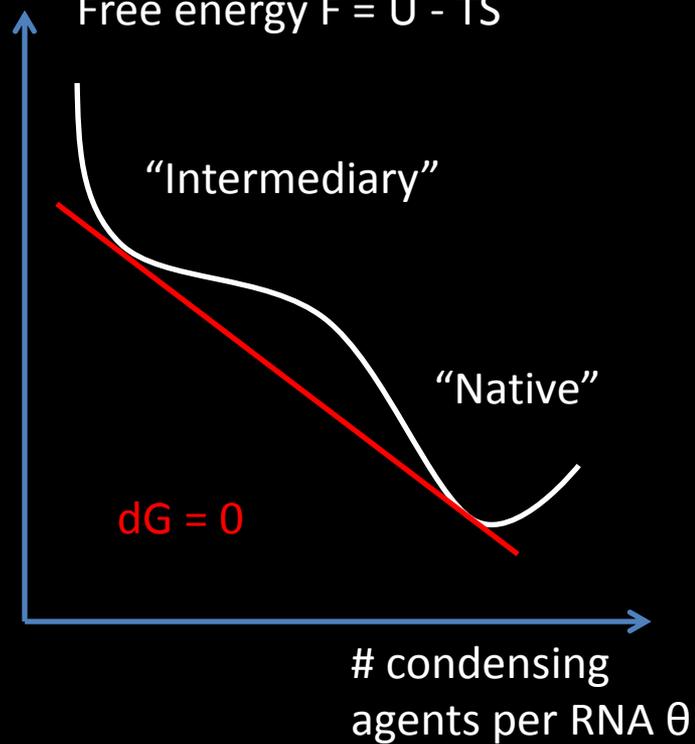
# RNA Condensation

Condensing Agent



Condensing agent concentration  
(polyvalent counterions)

Free energy  $F = U - TS$



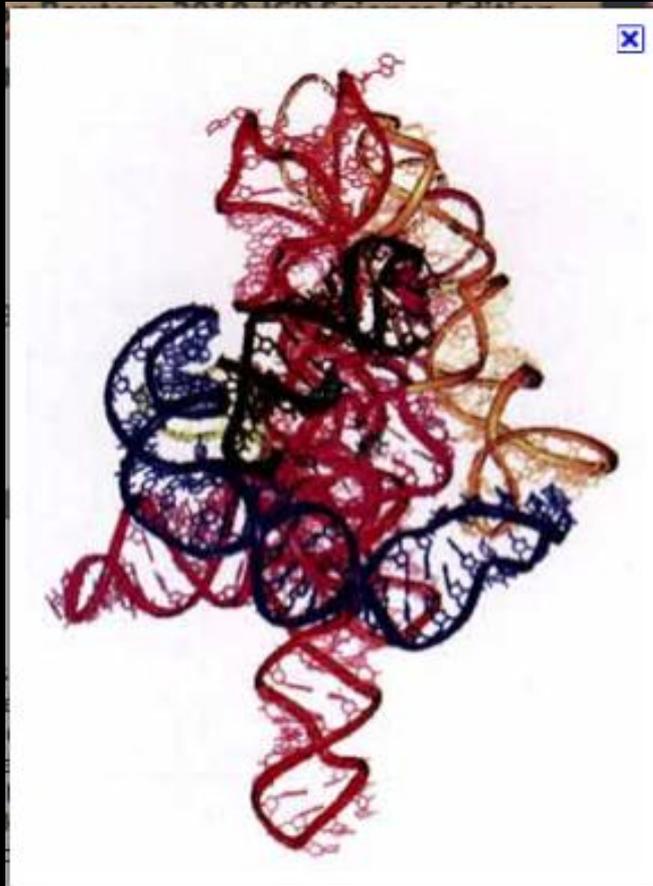
Gibbs Free Energy

$$G = F - \mu([\text{agent}]) \theta$$

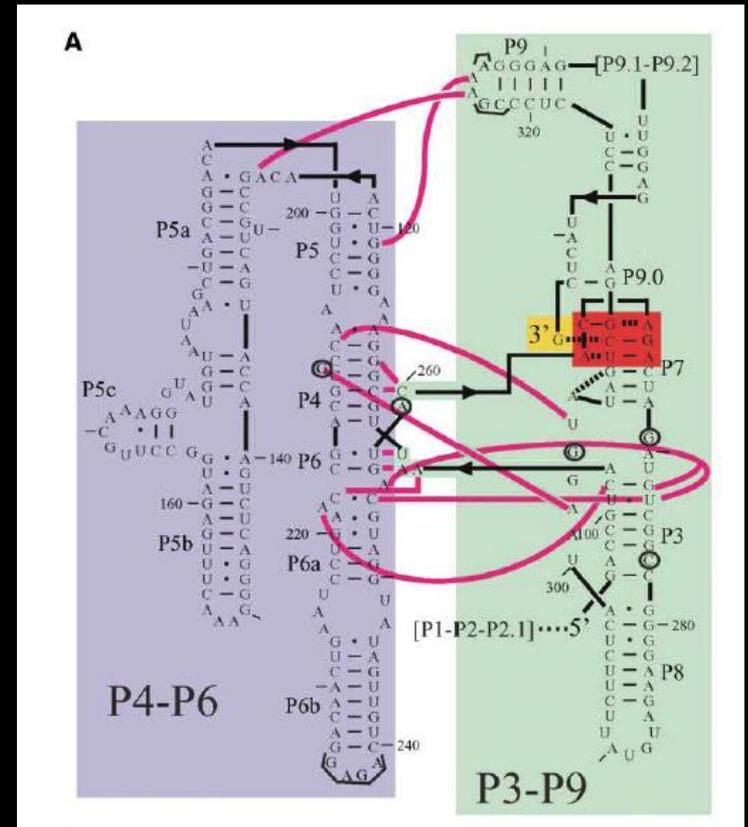
Chemical potential  
condensing agent

- Highly cooperative, first-order phase transition.

N state: folded

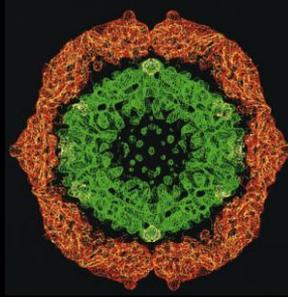


Tertiary contacts



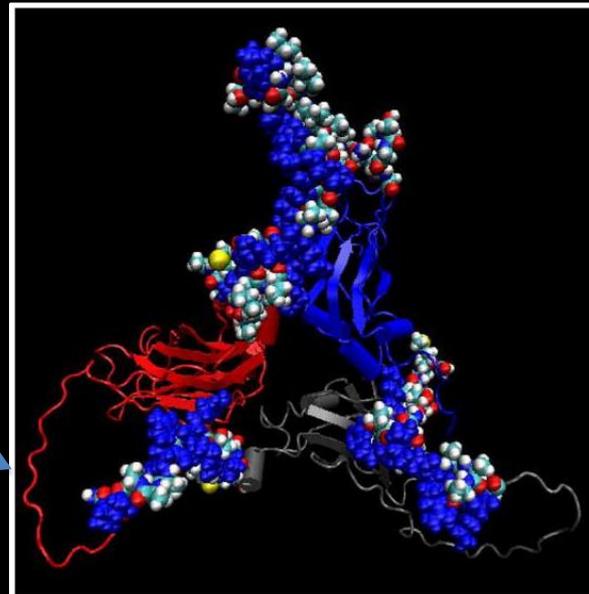
- Ribozyme (Tetrahymena) RNase (Cech)

- RNA inside T=3 virus:  
*Highly condensed*



What are the condensing agents ?

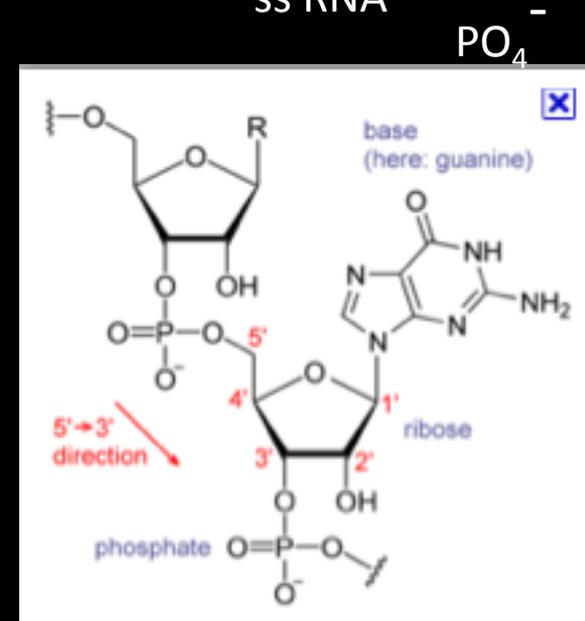
CCMV Capsid protein



Disordered N-Terminal Tail:  
+ 10 charges

*RNA Condensing Agent*

ss RNA

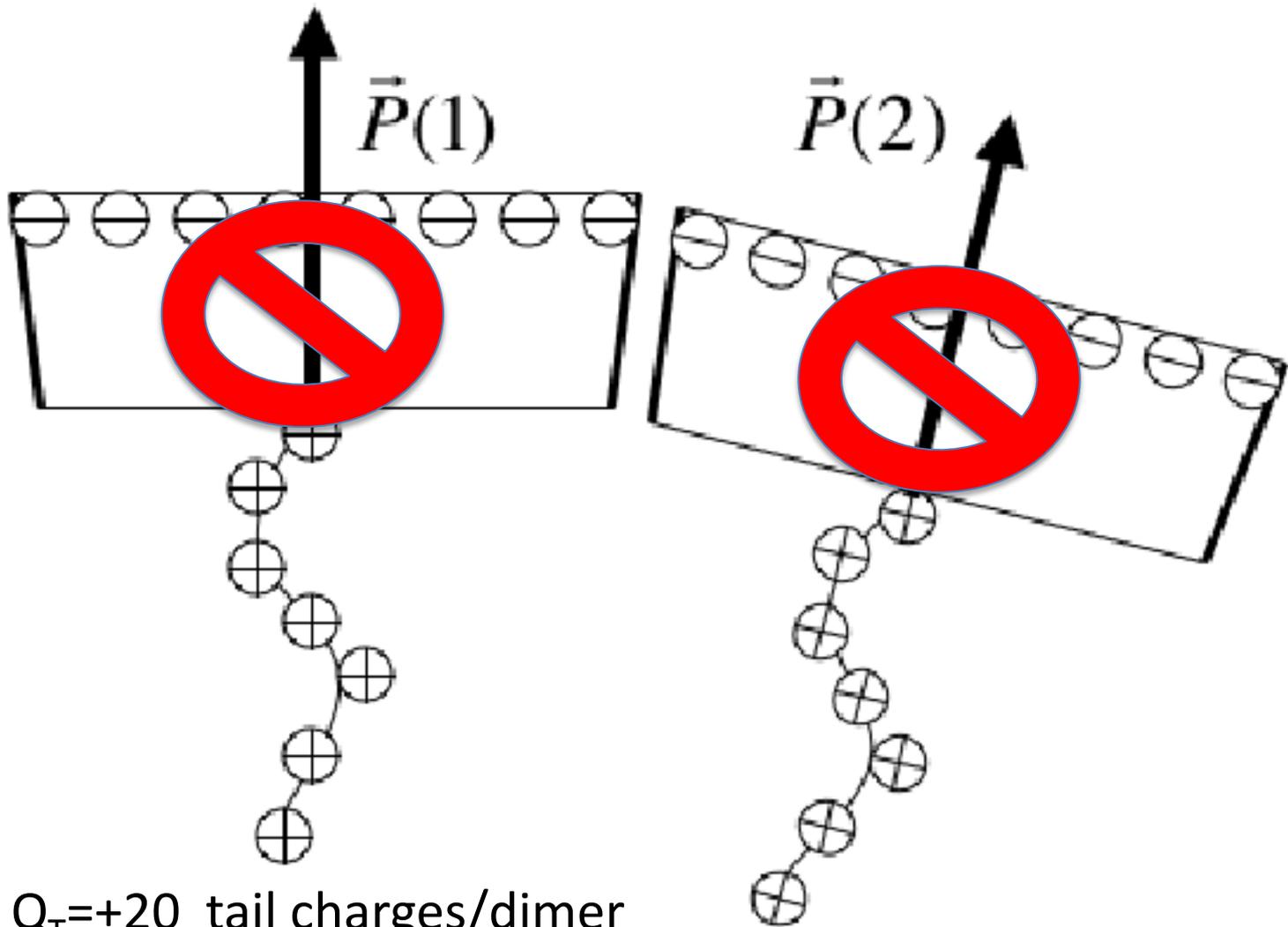


Numerical Simulation:  $\epsilon$  (tail/RNA)  $\approx$  10-15  $k_B T$

Zhang et al.

Remove Protein Cores

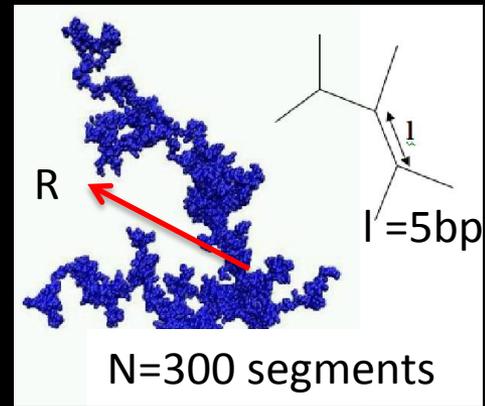
CCMV Dimers



$Q_T = +20$  tail charges/dimer

# Condensation of CCMV RNA

Good Solvent: “fractal”  
 $R(N) \approx N^{1/2}$



- Flory-Landau mean-field theory

## Entropic Elasticity U

$$F_F(R, N) = k_B T \frac{R^2}{R_0^2} + V(q) \frac{N^2}{R^3} + W \frac{N^3}{R^6} + \dots$$

$$R_0(N) = l N^{1/4}$$

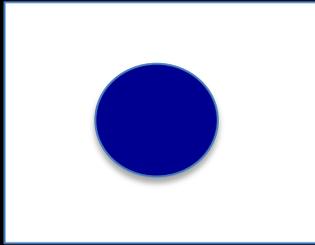
Radius gyration of an “ideal” Flory-Stockmayer branched polymer

Linear polymers: much larger

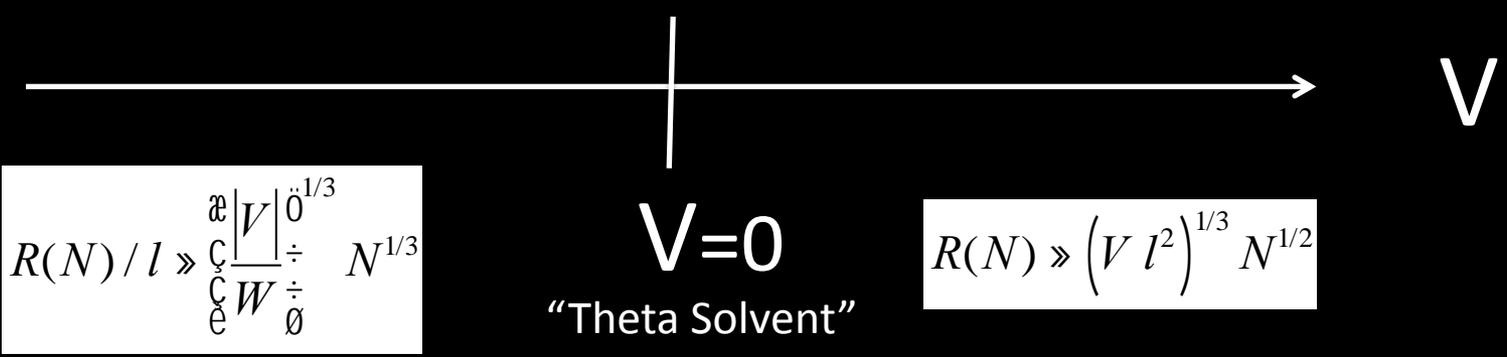
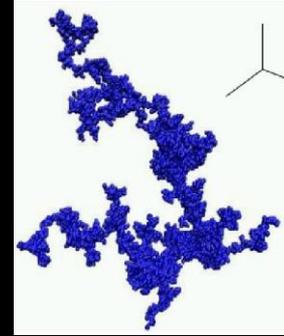
$$R_0(N) = l N^{1/2}$$

- $V(\theta)$ : Second Virial Coefficient.  $\theta$ : # tails / segment
- $W$ : Third Virial Coefficient  $\approx k_B T l^6$

Condensed Globule



Swollen Fractal



$$R(N)/l \gg \frac{\alpha |V|^{1/3}}{\zeta} \frac{1}{\zeta} N^{1/3}$$

$V=0$   
"Theta Solvent"

$$R(N) \gg (V l^2)^{1/3} N^{1/2}$$

CCMV RNA genome free in solution  
 $R \approx 11 \text{ nm}$

- $l = 0.5 \text{ nm}$
- $V(\theta=0)/k_B T = 1-10 \text{ nm}^3$

\* No phase transition

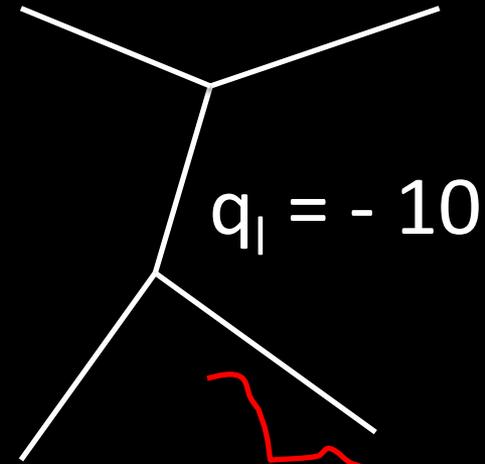
# Second Virial Coefficient

Segment charge

Tail fraction

Bjerrum Length

$$V(q) / k_B T = \frac{3l_B}{k^2} \left( -q_l + Q_T q \right)^2 - \bar{V}_{RNA}$$



Non-electrostatic

$Q_T = +10$   
(CCMV)

Polyvalent Counterion Charge

Debye parameter

Maximum concentration

$$q_{\max} = q_l / Q_T \gg 0.5$$

Neutralization

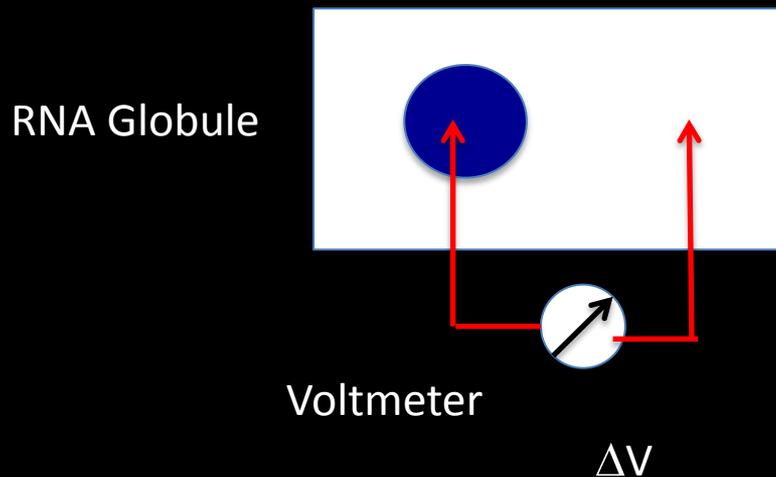
$$V(q=0) / k_B T \gg 100 \text{ nm}^3 - \bar{V}_{RNA} \gg 1 - 10 \text{ nm}^3$$

(free RNA in solution)

$$V(q = q_l / Q_T) / k_B T = -\bar{V}_{RNA} \gg -100 \text{ nm}^3$$

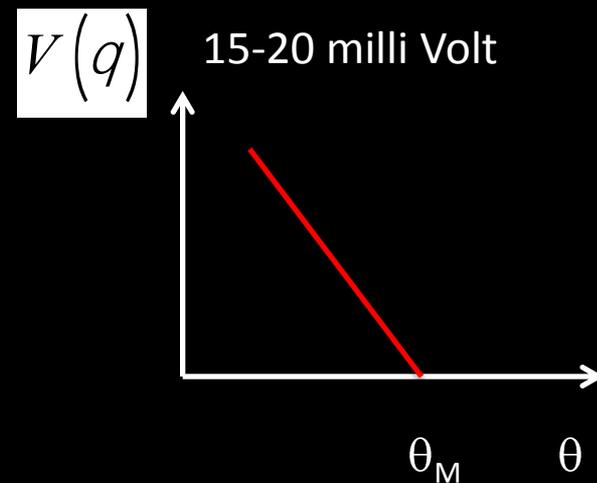
RNA/tail association:  
"unveils" strong RNA self-attraction





$$eDV_D(q) / k_B T = \frac{l_B (-q_l + Q_T q)}{k^2 R^3}$$

“Donnan Potential”



Charge  
neutral

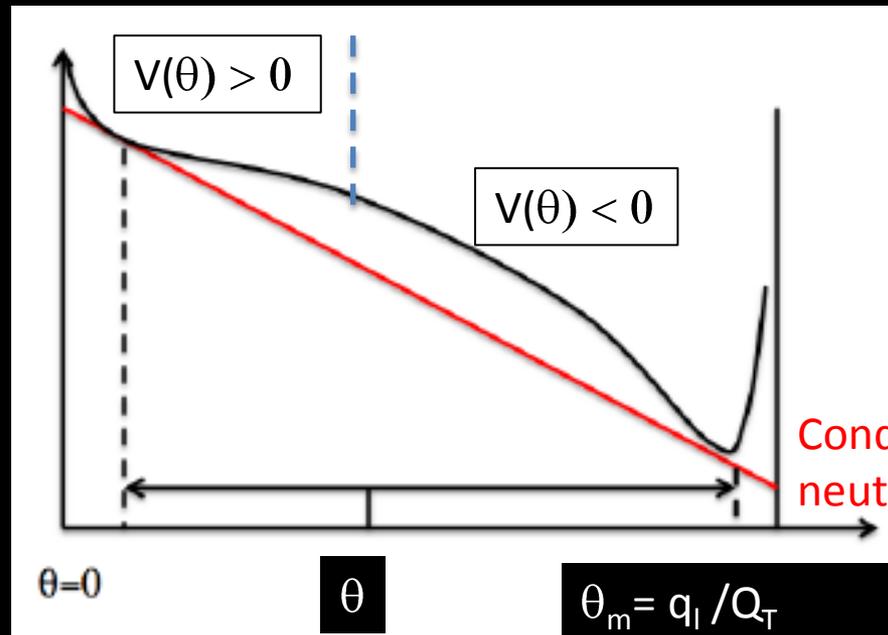
RNA/tail affinity

$$F_{RNA}(R, q) = F_F(R, q) + k_B T N q \ln q - (m + e) N q$$

Chemical potential tails

\* Minimize with respect to R

Swollen,  
charged



Condensed,  
neutralized

- Common-tangent Construction: Phase Coexistence

Gel swelling/shrinking

Large, reversible first-order phase transition



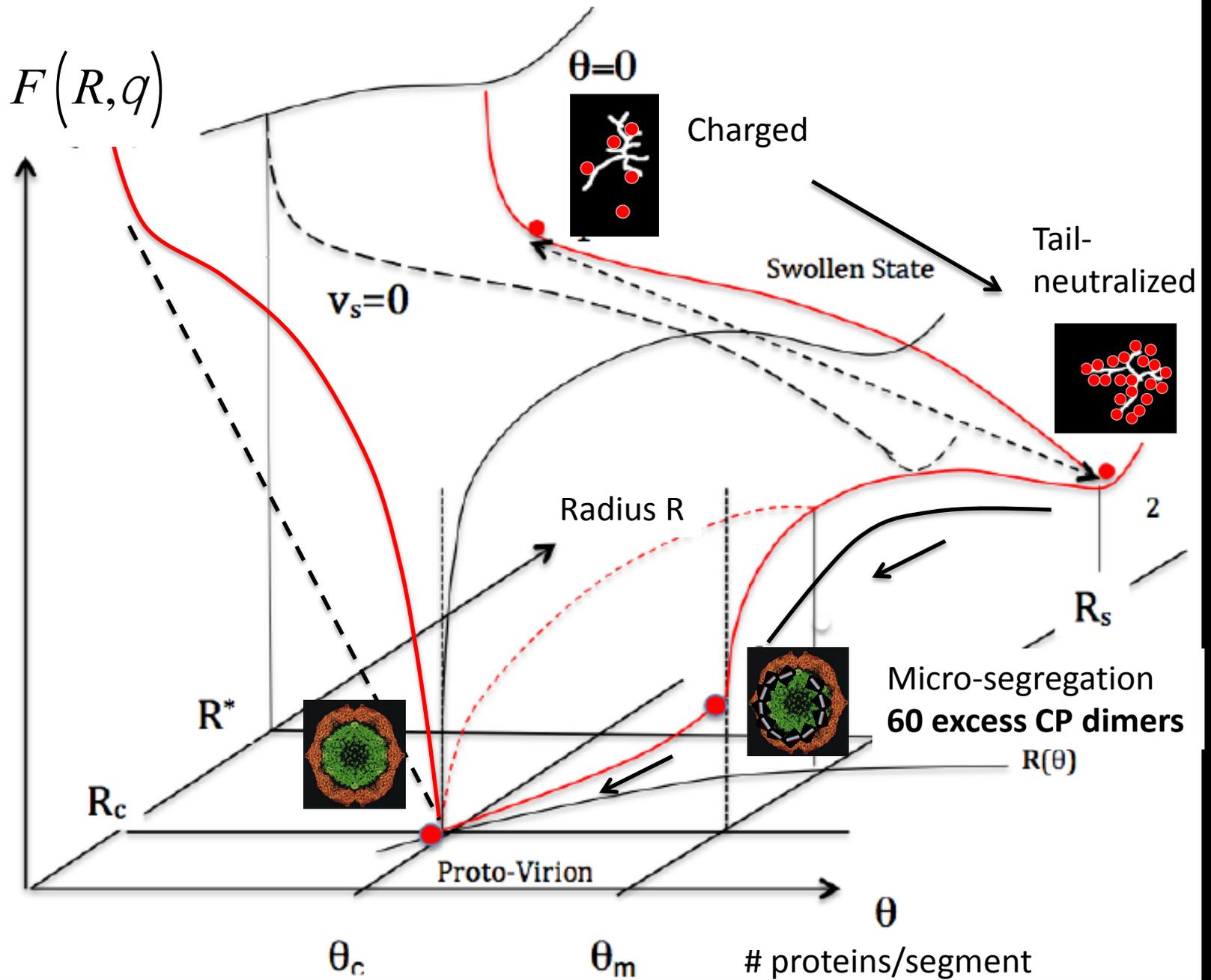
## V) Free Energy Landscape

Combine:

$$F(R, q) = F_{RNA}(R, q) + F_S(R, Nq)$$

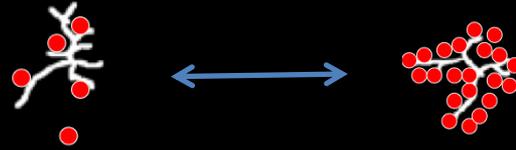


# Surface-Adsorbed CPs = # Tails



# Is this processes thermodynamic reversible self-assembly?

Step 1



Reversible

Protein-RNA assembly

Same CP chemical potentials

Step 2



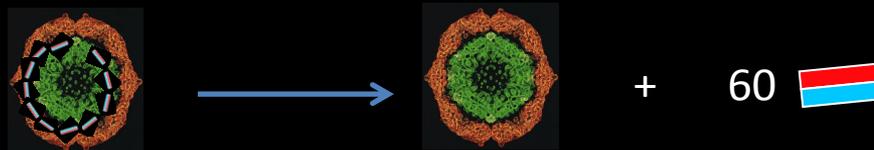
Irreversible

Micro-segregation

Lowered CP chemical potential  
Enhanced RNA self-attraction

$$k_B T \ln \left( \frac{\rho_B \rho_C^2}{\rho} - J_s \left( \frac{R_c}{D} \right) \frac{\partial}{\partial r_s} \right)$$

Step 3



Irreversible

Protein expulsion

Lowered CP chemical potential

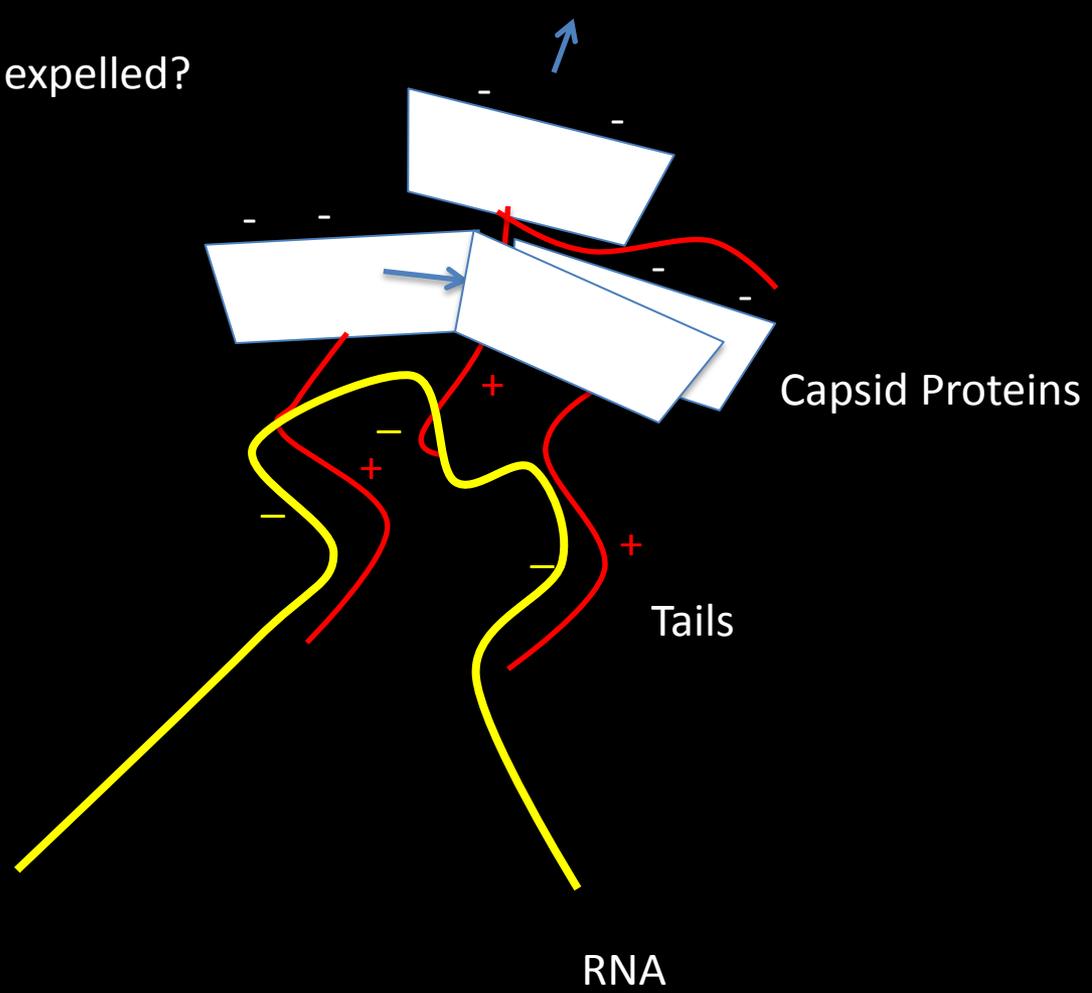
$$e^{-DV_D Q_C}$$

“Michaelis-Menten like”

Donnan Potential + Protein Self-repulsion

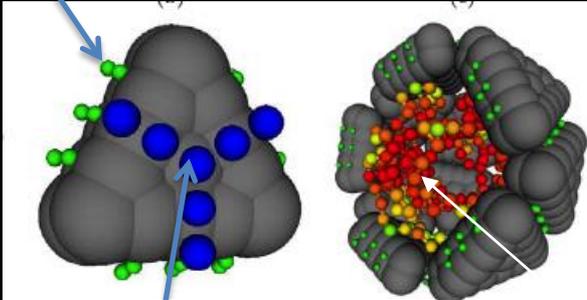
How are excess proteins expelled?

Brownian Ratchet:



# How good is mean-field theory?

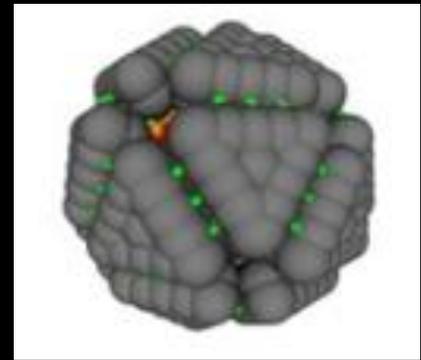
Protein-Protein binding sites



Flexible *linear* polymer genome

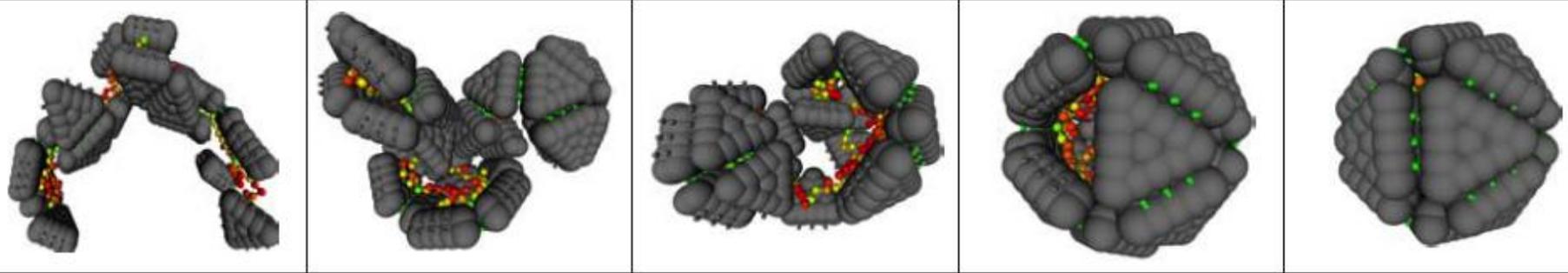
genome binding sites”

- \* Genome molecule: no branching.
- \* Assembled state: # binding sites = chain length



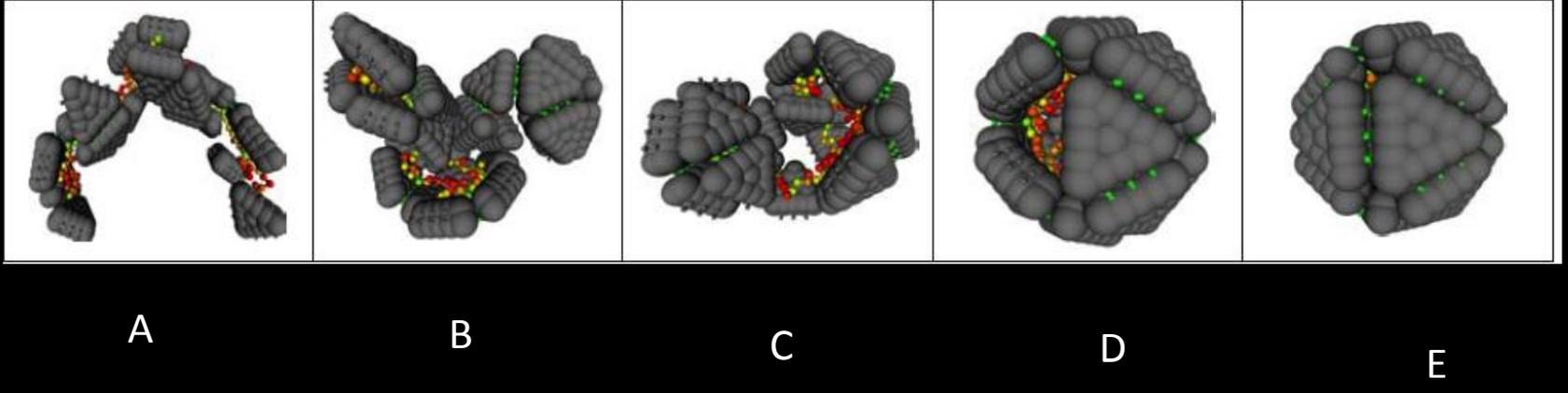
Toy T=1 Virus

Protein-genome affinity  $\varepsilon >$  Protein-protein affinity  $J$



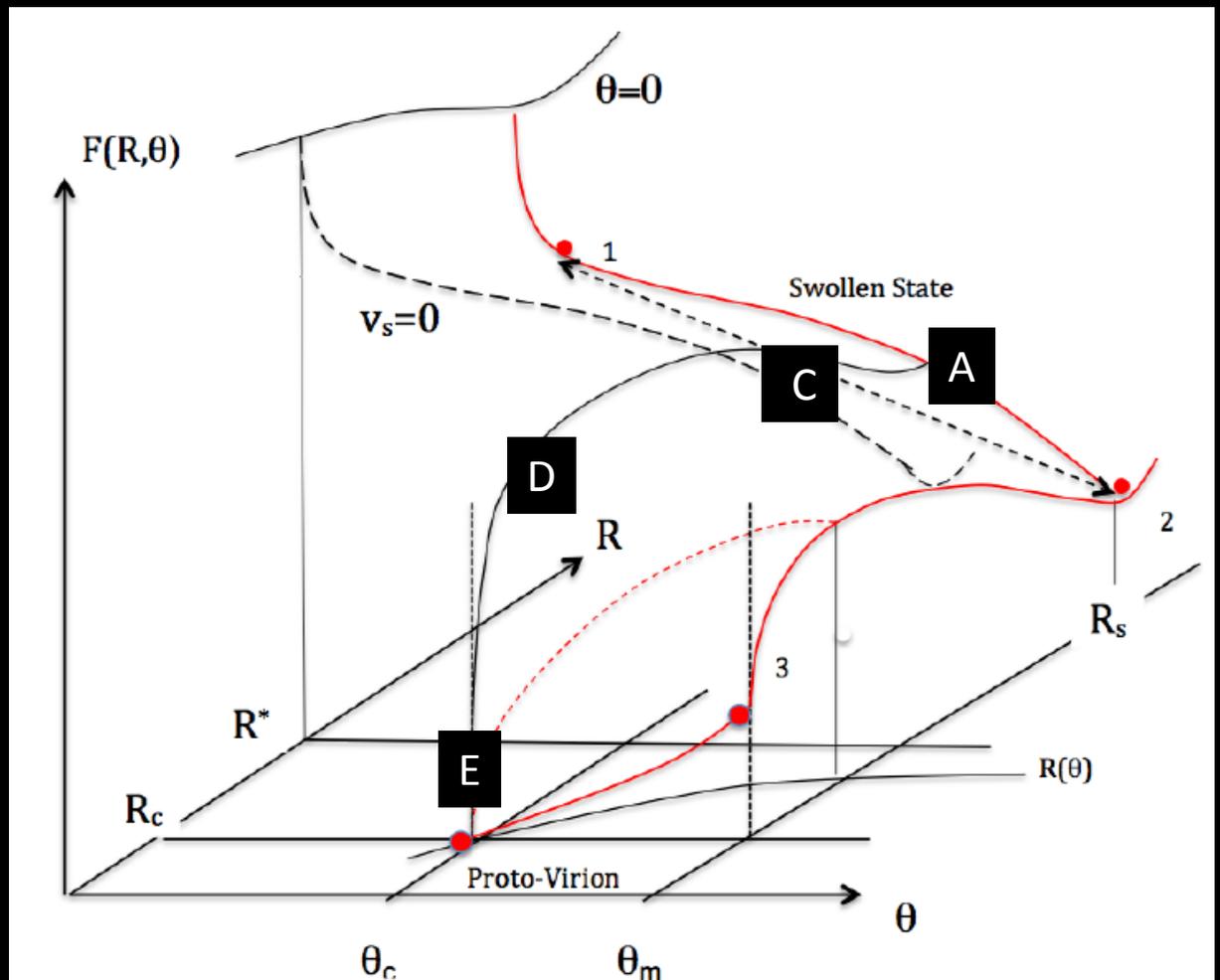
time

\* RNA/Protein pre-assembly condensate

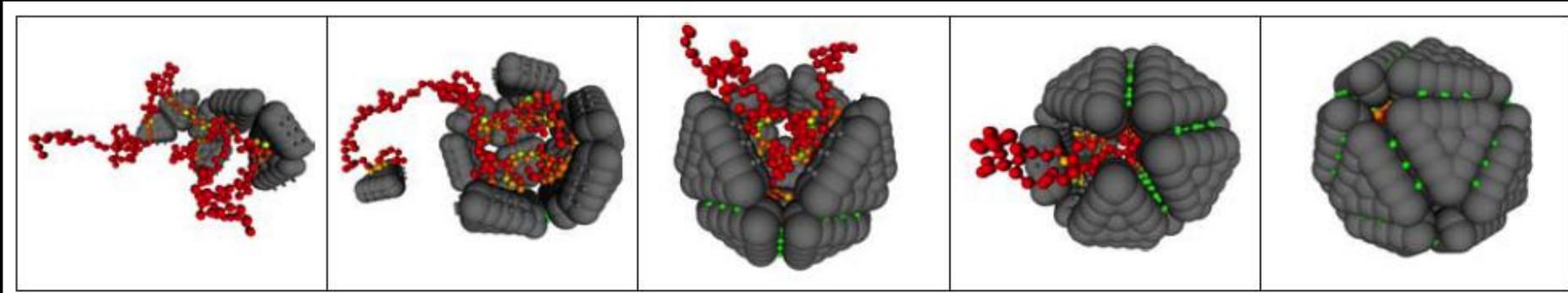


Problem: Optimal angles visible in A-C

- Local correlations.

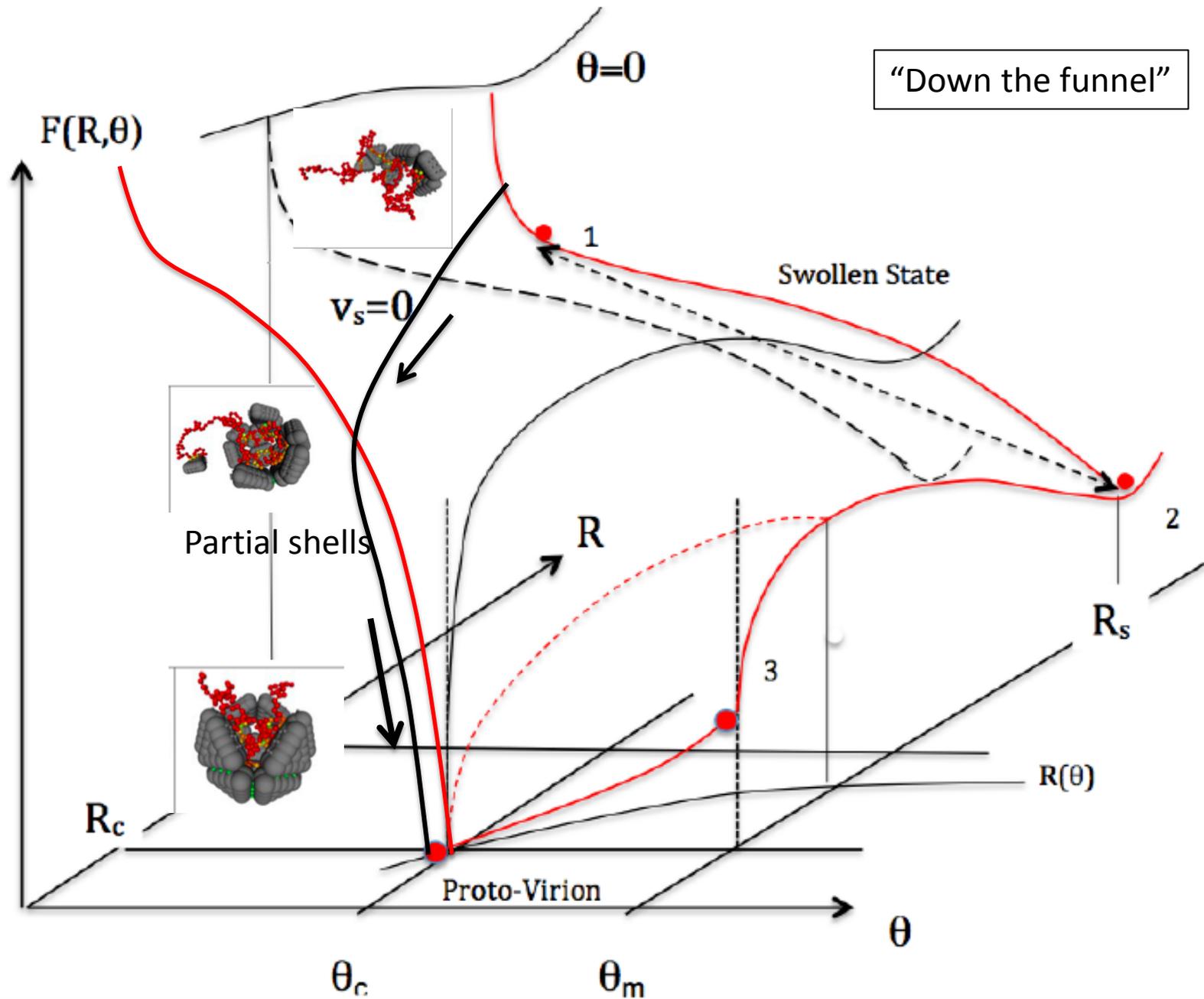


Genome-protein affinity  $\varepsilon$  weaker than protein-protein affinity  $J$



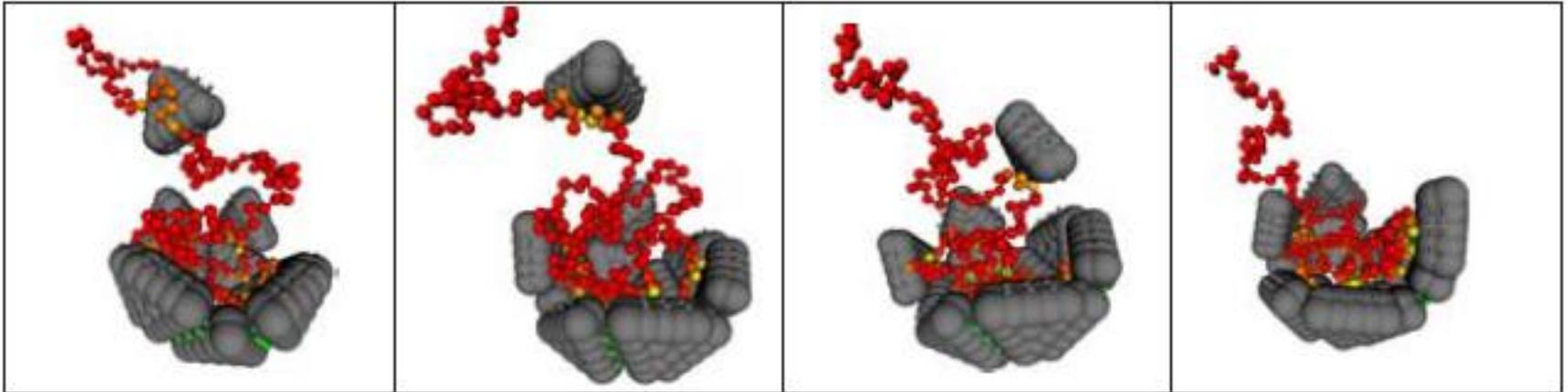
RNA “glues” capsomers together one-by-one

\* *Heterogeneous nucleation* of a shell on a flexible RNA scaffold



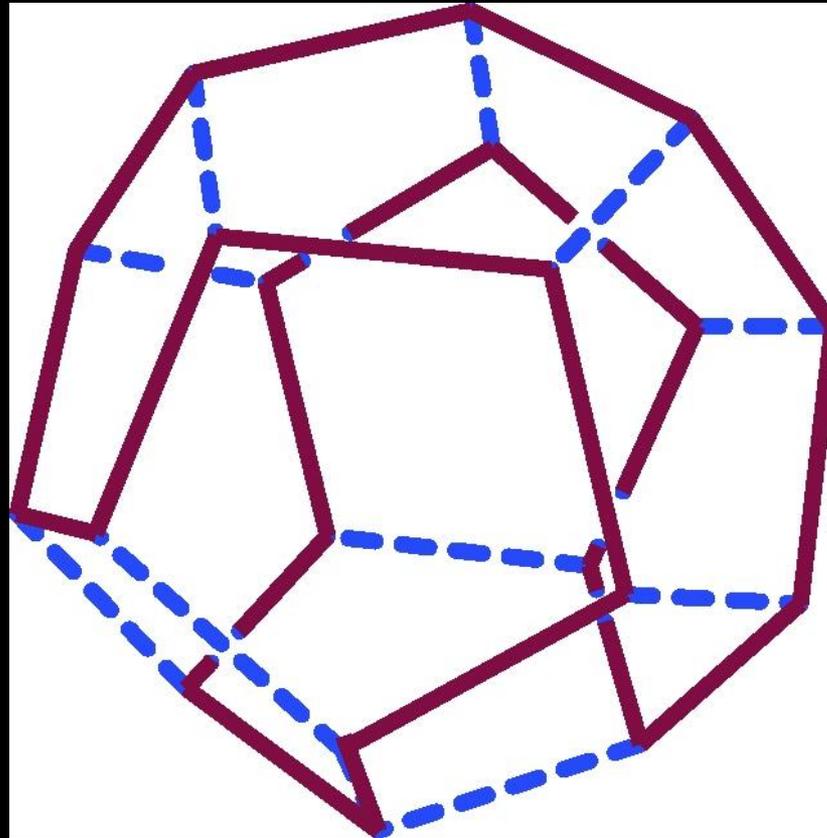
Many possible assembly pathways

## “Antenna-Assembly”



(Hu and Shklovskii)

## “Hamiltonian Cycle”



- Graph-theoretic problem  
(R.Twarock)

## Conclusions

- 1) Assembly of small ss RNA viruses can be viewed as the combination of reversible RNA condensation + quasi-reversible shell formation.
- 2) Combination of two simple thermodynamic assembly processes produces a more complex free energy landscape with different possible multi-step irreversible pathways.
- 3) Viral assembly appears intermediate between Mark I and Mark II assembly.